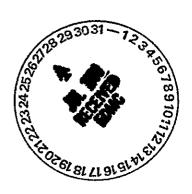
START

NATIONAL PRIORITY LIST U.S. DOE HANFORD 100 AREA

AUGUST 17, 1987



Facility name: U.S. DOE Hanford 100 Area Rece

Received SEP 1 4 1917

Location: Hanford Site

EPA Region: X

Person(s) in charge of the facility: J. J. Keating, Asst. Mgr.

Safety, Environment and Security

<u>509 - 376 - 7334</u>

Name of Reviewer: D. M. Bennett, EPA Region X

Date: 8-17-87

General description of the facility:
(For example: landfill, surface impoundment, pile, container; types of hazardous substances; location of the facility; contamination route of major concern; types of information needed for rating; agency action,

etc.)

The 100 Area Site contains the 116-B-4, 116-B-6, 116-C-2, 116-D-1A,

116-D-18, 116-F-5, 116-F-10, 116-H-3, 116-K-2, 116-B-1, 116-B-3, 116-C-1,

116-D-2,116-DR-1,116-DR-2,116-DR-4,116-F-1,116-F-2,116-F-3,116-F-4,

116-H-1,116-H-4,116-K-1,116-K-2,116-B-5,116-B-10,116-C-2,116-D-3,

116-D-4, 116-F-9, 116-F-13, 116-B-2, 116-D-1A, 116-D-1B, 116-DR-3, 116-F-3,

116-KE-3,116-KW-2,116-DR-6,116-F-1,116-F-6,116-F-12,116-H-2,116-K-2,

116-DR-8,116-F-7,117-B,117-C,117-D,117-H,100KE*1,100KE*2,100KE*3,100KW*1

100KW*2,116-B-9,116-D-6,116-F-11,116-DR-7,116-KE-1,116-KW-1,116-KE-2,

107-B, 107-C, 107-D, 107-DR, 107-F, 107-H, 107-KE, 107-KW, White Bluffs

Pickling Acid Crib, 118-B-1, 118-B-4, 118-B-5, 118-C-2, 118-D-1, 118-D-3,

<u>118-D-5,118-DR-1,118-F-1,118-F-3,118-F-7,118-H-1,118-H-2,118-H-3,</u>

Scores: $S_M = 46.38$ ($S_{GW} = 6.12$ $S_{SW} = 80.00$ $S_a = 0.00$)

S_{FE} = 0.00

 $S_{DC} = 0.00$

FIGURE 1 HRS COVER SHEET

> PLEASE RETURN TO: ENVIRONMENTAL DIVISION RESOURCE CENTER

118-H-4, 118-H-5,188-K,118-B-2,118-B-3,118-D-4,118-F-1,118-H-3,

118-B-6,118-F-5,118-F-6,118-B-1,118-B-7,118-C-1,118-D-1,118-D-2,

118-F-2,118-F-4,118-H-1,118-K,1008/C Burn Pit,100DR Burn Pit,100F

Burn Pit,100N Burn Pit and 100K Burn Pit sites.

FIGURE 1 (Continued)
HRS COVER SHEET

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HRS Ground Water Route Work Sheet

Rating Factor	Assigned Value	plier	Score	Score	(Section)
Observed Release	45	1	45	45	3.1
If observed release is given If observed release is given	a score o	f 45, pro f 0, proc	ceed to	line 4. line 2.	
Route Characteristics				_	. 3.2-
Depth to Aquifer of Concern Net Precipitation	0	2 1	0	6 3	
Permeability of the Unsaturated Zone	Ö	1	0	3	
Physical State	0	1	0	3	
Total Route Characteristics	Score		0	15	
Containment	0	1	0	3	3.3
Waste Characteristics Chemical					3.4
a. Toxicity/Persistence Hazardous Waste Quantity	18 8	1	18 8	18 8	
Total Waste Characteristics	Score		26	26	
Targets					3.5
Ground Water Use Distance to Nearest Well/ Population Served		3 1	3 0	9 40	
Total Targets Score			3	59	
If line 1=45 (1x4x5) If line 1=0 (2x3x4x5)			3510	5733	o feviser
. Line 6/57330 * 100 Sc	(gw) =	6.12			k

HRS Surface Water Route Work Sheet

Site: U.S. DOE Hanford 100 Area				8	3/17/87
Rating Factor	Assigned Value	plier		Max. Score	Ref. (Section)
1. Observed Release	45	1	45	45	4.1
If observed release is given If observed release is given	a score of a score of	45, proc 0, proce	eed to	line 4.	
2. Route Characteristics Facility Slope & Intervening Terrain	0	1	0	3	4.2
1-yr. 24-hr. Rainfall Distance to Nearest Surface	0 0	1 2	0	3 6	
Water Physical State	0	1	0	3	
Total Route Characteristics	Score		0	15	
3. Containment	0	1	0	3	4.3
4. Waste Characteristics a. Chemical Toxicity/Persistence Hazardous Waste Quantity	18 8	1	18 8	18 8	4.4
Total Waste Characteristics	Score		25	26	
5. Targets Surface Water Use Distance to a Sensitive	3	3 2	9	9 6	4.5
Environment Population Served/Distance to Water Intake Downstream	35	1	35	40	٠.
Total Targets Score	********		44	55	
6. If line 1=45 (1x4x5) If line 1=0 (2x3x4x5)		5148	0 64:	350	
7. Line 6/64350 * 100 Sc(sw)= 80	00.00		****	1/11/8

HRS Air Route Work Sheet

ROUTE NOT SCORED

: 		SCORED	<u> </u>		
	Assign Value	ed Multi plier	- Score	Score	(Section
. Observed Release	0	1			
Date and Location:					
Sampling Protocol:					
If observed release is given If observed release is given	a score	of 45, pr of 0, the	ocee& t	o line 2	. •
. Waste Characteristics					5.2
a. Chemical Reactivity and Incompatibility	0	1	0	3	
Toxicity Hazardous Waste Quantity	0	3 1	0	9 8	
Total Waste Characteristics			_	20	
Targets					5.3
Population Within 4-Mile Radius	0	1	0	30	
Distance to Sensitive Environment	0	2	0	6	
Land Use	0	1	0	3	
Total Targets Score			0	39	
Multiply 1 x 2 x 3			0	35100	,

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Groundwater Route Score (Sgw)	6.12	37.45
Surface Water Route Score (Sgw)	80.00	6400.00
Air Route Score (S _a)	0.00	0.00
$s_{gw}^2 + s_{sw}^2 + s_a^2$	-	6437.45
$\sqrt{\frac{s_{gw}^2 + s_{sw}^2 + s_a^2}{s_{gw}^2 + s_{sw}^2 + s_a^2}}$	-	80.23
$\sqrt{S_{gw}^2 + S_{sw}^2 + S_a^2} / 1.73 = S_M =$	-	46.38_

FIGURE 10 WORKSHEET FOR COMPUTING S_M

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DOCUMENTATION RECORDS FOR HAZARD RANKING SYSTEM

INSTRUMENTATIONS: As briefly as possible summarize the information you used to assign the score for each factor (e.g., "Waste quantity = 4,230 drums plus 800 cubic yards of sludge"). The source of information should be provided for each entry and should be a bibliographic-type reference, include the location of the document.

FACILITY NAME: U.S. DOE Hanford 100 Area
· •
LOCATION: Hanford Site, Benton County, Washington
DATE SCORED: 8-17-87
PERSON SCORING: R. D. Stenner, Pacific Northwest Laboratory for DOE

PRIMARY SOURCE(S) OF INFORMATION (e.g., EPA region, state, FIT, etc.):

The information was taken from Department of Energy documents and databases associated with the Hanford Site, as well as from other publicly available documents addressing conditions at or in the vicinity of the Hanford Site. Information was also gathered through telephone and personal communications with responsible individuals (such information is referenced accordingly in the package).

FACTORS NOT SCORED DUE TO INSUFFICIENT INFORMATION:

Even though air concentrations of some of the constituents of interest can be detected above background offsite, no air monitoring data were found sufficient for HRS scoring of the Hanford CERCLA sites. These constituents of interest detected above background offsite are present in the routine gaseous effluents from operating facilities at Hanford. Therefore, the air route rating factors were not scored.

COMMENTS OR QUALIFICATIONS:

The Department of Energy has completed a preliminary assessment of the hazardous waste sites located on the Hanford Site. This work served as the primary basis for developing the scores for the aggregate 100 Area Site. These preliminary assessment efforts are documented in the "Draft Phase I Installation Assessment of Inactive Waste-Disposal Sites at Hanford" volumes and respective addenda which are referenced throughout the package.

GROUND WATER ROUTE

1 OBSERVED RELEASE

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Contaminants detected (5 maximum):

An observed release to the ground water beneath the 100 Areas can be verified using several different substances. For the purpose of documenting observed release to the ground water it was determined that the strontium-90 (Sr-90) and chromium (Cr) would provide sufficient evidence. Evidence of observed release can be given in terms of a comparison of background concentrations and down gradient concentrations for both Sr-90 and Cr. Wells used to verify observed release of Sr-90 are listed in the following table.

100 Area Wells for Sr-90

BACKGROUND			DOWN GRADIENT				
Date	Well	Concentration (pCi/L)	Date	Well	Concentration (pCi/L)		
4/10/87	1-H4-5	<0.753 pCi/L	4/10/87 5/29/87	1-H4-4 1-N-4	8.05 pCi/L 12.5 pCi/L		

The wells used to verify observed release of Cr to the ground water beneath the 100 Area are listed in the following table.

100 Area Wells for Cr

BACKGROUND			DOWN GRADIENT				
Date	Well_	Concentration (µg/L)	Date	Well	Concentration (µg/L)		
6/12/87	6-77-54	<10 µg/L	5/21/87 3/25/87 3/18/87	1-83-1 1-D5-12 1-K-20	62 μg/L 1,560 μg/L 137 μg/L		

Reference 5, page 3.14; Reference 2; Reference 3, pages 19, 23, 274, 13,16, and 21; Reference 1; Reference 10; Reference 18

Rationale for attributing the contaminants to the facility:

The nine 100 Areas (B, C, D, DR, KE, KW, F, H, and N) border the Columbia River in the northernmost part of the Hanford Site. Each of the nine areas has one production reactor. Eight of these reactors have been shut down; only the N Reactor, used for both plutonium and electricity production is still in operation. When the reactors were operational, cooling water was drawn from the river and treated with alum, sulfuric acid and chlorine. Excess sulfuric acid was used to maintain the pH of the water within a desired range. To control oxidation of aluminum parts in the reactor, sodium dichromate was used to maintain an oxidation coating on aluminum parts. The chlorine was added for algae control in the settling basins; at times copper

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sulfate was added for additional algae control. As the cooling water passed through the reactor various activation products (e.g. Co-60, Ni-63, etc.) were dissolved in the cooling water. When a fuel element ruptured or developed a pin hole in its' cladding some fission products such as Sr-90 would dissolve in the cooling water. Chromic acid, oxalic acid and nitric acid were used for dummy fuel-element decontamination. In addition to vertical safety rods for emergency reactor shutdown, the reactors were equipped with hoppers of nickel-plated boron steel balls, nickel-plated carbon steel balls, and stainless steel balls that would drop into the vertical safety rod channels for emergency shutdown. This system required no supplementary power source. Although it was never used, a third safety system, one involving the use of potassium borate solution, was in place at the reactors. A supplementary control system, in addition to the normal horizontal control rods, was incorporated into the reactors. This supplementary control system consisted of a Poison(a) Column Control Facility that could charge selected process tubes with a lead-cadmium poison to absorb neutrons. Boron-carbide aluminum poison splines were also used for supplementary control. These activities resulted in waste containing the contaminants discussed above being disposed of in the 100 Area. Down gradient wells show contaminant concentrations significantly above those found in the background wells.

Reference 4, pages 2.11-2.13

2 ROUTE CHARACTERISTICS

Depth to Aquifer of Concern

Name/description of aquifer(s) of concern:

The aquifer of concern is the unconfined aquifer which is comprised of the glacio fluvial sediments of the Hanford formation and the late deposits of the Ringold formation. It generally slopes downward from west to east; depth to ground water is from 10 to 15 meters (34 to 48 feet). It is bounded below by either the basalt surface or, in places, the relatively impervious clays and silts of the lower unit of the Ringold formation. Laterally, the unconfined aquifer is bounded by the anticlinal basalt ridges that ring the basin. The Yakima and Columbia Rivers, however, do not entirely transect these sediments and therefore do not constitute a discontinuity for HRS Scoring purposes. The basalt ridges above the water table have a low permeability and act as a barrier to lateral flow of the ground water. The saturated thickness of the unconfined aquifer is greater than 61 m in some areas of the Hanford Site and pinches out along the flanks of the basalt anticlines.

Recharge to the unconfined aquifer originates from several sources. Natural recharge occurs from precipitation at higher elevations and runoff from

⁽a) The term poison refers to a material's ability to absorb neutrons and thus control the rate of fission.

ephemeral streams to the west, such as Cold Creek and Dry Creek. The Yakima River recharges the unconfined aquifer as it flows along the southwest boundary of the Hanford Site. The Columbia River recharges the unconfined aquifer during high stages when river water is transferred to the aquifer along the river bank. The unconfined aquifer receives little recharge from precipitation directly on the Hanford Site because of a high rate of evapotranspiration under native soil and vegetation conditions. Large scale artificial recharge occurs from offsite agricultural irrigation and liquid-waste disposal in the operating areas at Hanford.

Underlying the surface sands is a mixture of sand and gravel extending to a depth of about 200 feet (60 meters). Basaltic rock starts at that depth and extends downward over 1.9 miles (3000 meters).

Reference 5, page 2.5; Reference 4, pages 2.11-13; Reference 9

Depth(s) from the ground surface to the highest seasonal level of the saturated zone [water table(s)] of the aquifer of concern:

Not applicable.

Depth from the ground surface to the lowest point of waste disposal/storage: Not applicable.

Net Precipitation

Mean annual or seasonal precipitation (list months for seasonal): Not applicable.

Mean annual lake or seasonal evaporation (list months for seasonal): Not applicable.

Net precipitation (subtract the above figures):
Not applicable.

Permeability of Unsaturated Zone

Soil type in unsaturated zone:

Not applicable.

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Permeability associated with soil type:

Not applicable

Physical State

Physical state of substances at time of disposal (or at present time for generated gases):

Not applicable.

3 CONTAINMENT

Containment

Method(s) of waste or leachate containment evaluated:

Not applicable.

Method with highest score:

Not applicable.

4 WASTE CHARACTERISTICS

Toxicity and Persistence

Compound(s) evaluated:

These substances are associated with production reactor operation (See Tables 1 and 2):

Substances in the 100 Area

uranium lead cadmium sodium dichromate sodium oxalate sulfuric acid sulfamic acid	mercury plutonium-238 plutonium-239 plutonium-240 tritium cobalt-60 stronger sulfato	cesium-137 cesium-134 europium-152 europium-154 europium-155 sodium sulfamate nickel-63
potassium borate	copper sulfate	sodium hydroxide

Reference 4, pages 2.11-2.13; Reference 6 (The numerous individual pages are not included in the package, please consult the reference 6 document using the list of individual sites listed on the cover sheet of this package for verification of which individual sites contain which substances); Reference 19

Compound with highest score:

Several of these substances results in a score of 18. Uranium, lead, mercury sodium dichromate and plutonium are among those having scores of 18.

Substance	Toxicity	Persistence	TOTAL
	Score	Score	Score
uranium lead mercury dichromate plutonium	3 3 3 3 3	3 3 3 3	18 18 18 18

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Reference 7, pages 794-797; REference 20.

Hazardous Waste Quantity

0

Total quantity of hazardous substances at the facility, excluding those with a containment score of 0 (Give a reasonable estimate even if quantity is above maximum):

The total quantity of hazardous substances associated with the aggregate 100 Area Site is estimated to be 4.3 billion cubic yards. Table 1 presents the individual sites and associated data for the 100 Area Liquid Waste Sites that were used in creating the aggregate 100 Area Site. Table 2 presents the individual sites and associated data for the 100 Area Solid Waste Sites that were used in creating the aggregate 100 Area Site.

References are listed by waste site in Table 1.

Basis of estimating and/or computing waste quantity:

The volume of wastes disposed of to waste sites at the 100 Area was calculated as the sum of the volumes received by individual waste sites. The basis for this estimate of liquid waste volume is given in Table 1. For purposes of developing the inventories, individual waste sites were aggregated into groups that received similar wastes. These groups, and the individual sites comprising each group, are identified in Table 1. Also included in this table are a description of the general types of waste received by the group of sites, the period of operation covered by the group, the estimated waste volume received by the group, and the references for the waste volumes received by the individual sites. In some cases, an individual site received more than one type of waste. If so, the site was assigned to more than one group and its waste volume was divided into the volumes received by each particular site group. Similar information for 100 Area sites receiving solid wastes is presented in Table 2.



TABLE 1. 100 AREA LIQUID WASTE SITES

Area No. and Name	Waste Types	_Years_	Estimated Volume	No. of Cubic Yds	References
100 Area Decontamination Waste Cribs (116-B-4, 116-B-6, 116-C-2, 116-D-1A, 116-D-1B, 116-F-5, 116-F-10, 116-H-3, 116-K-2)	Decontamination solutions consisting of chromic, oxalic, and/or sulfuric acid solutions neutralized with soda ash. Solutions were used to decontaminate dummy fuel elements and reactor hardware.	1947-1970	2.3 x 10 ⁶ L	3,000	Reference 6, pages 13-14, 17-20, 45-48, 59-60, 61-62, 113-114, 121- 122, 151-152, 173-174
100 Area Contaminated Reactor Effluent Cribs and Trenches (116-B-1, 116-B-3, 116-C-1, 116-D-2, 116-DR-1, 116-F-1, 116-F-2, 116-F-3, 116-F-4, 116-II-1, 116-II-4, 116-K-1, 116-K-2)	Reactor effluent contaminated when fuel element cladding failed. Radiological contamination consisted chiefly of mixed fission products. Chemical contamination consisted of sodium dichromate added to reactor coolant as oxidant.	1946-1971	3.0 x 10 ¹¹ L	3.9 x 108	Reference 6, pages 7-8, 11-12, 43-44, 63-64, 87-88, 89-90, 93-94, 105-106, 107- 108, 109-110, 111-112, 147- 148, 153-154, 171-172, 173-174

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Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Laboratory Waste Cribs and Trenches (116-B-5, 116-B-10, 116-C-2, 116-D-3, 116-D-4, 116-F-9, 116-F-13)	Wastes associated with labs and shops in the 108 Buildings, including 108-B Tritium Lab and Tube Examination Facility, 108-C Metal Examination Facility and 100 F Biology Labs. Contaminants consisted mainly of tritium and low levels of mixed fission and activation products. Negligible chemical contamination except Biology Lab facilities, which received nitrate from animal wastes.	1950-1976	3.2 x 10 ⁸ L	4.2 x 10 ⁵	Reference 6, pages 15-16, 23-24, 45-48, 65-66, 67-68, 119-120, 127-128
100 Area Fuel Storage Basin Cribs and Trenches (116-B-2, 116-D-1A, 116-D-1B, 116-DR-3, 116-F-3, 116-KE-3, 116-KH-2)	Storage basin cooling water contaminated with low levels of mixed fission products due to failure of fuel element cladding.	1946-1971	2.0 x 10 ⁷ L	2.6 x 10 ⁴	Reference 6, pages 9-10, 59-60, 61-62, 91-92, 109-110, 179-180, 195-196

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Area No. and Name	Waste Types	<u>Years</u>	Estimated Volume	No. of Cubic Yds	References
100 Area Reactor Decontamination and Maintenance Effluents Cribs and Trenches (116-DR-6, 116-F-1, 116-F-6, 116-F-12, 116-H-2, 116-K-2)	Reactor effluents during reactor maintenance activities. Much of the waste volume was uncontaminated. Radionuclides present were mainly low levels of mixed fission products and activation products. The main chemical contaminant was sulfuric acid, which was used for reactor decontamination. Sodium dichromate was also present in some of the effluents.	1945-1971	6.1 x 10 ⁸ L	8.0 x 10 ⁵	Reference 6, pages 95-96, 105-106, 115- 116, 125-126, 149-150, 173-174
100 Area Reactor Confinement Seal Pit Drainage Cribs (116-DR-8, 116-F-7, 117-B, 117-C, 117-D, 117-H)	Drainage from 117 Building confinement system seal pits. Very low levels of contamination.	1960-1969	2.2 x 10 ⁶ L	2,900	Reference 6, pages 42-44, 49-50, 85-86, 99-100, 117-118, 167-168
100 Area Sulfuric Acid Sludge Disposal Sites (100KE*1, 100KE*2, 100KE*3, 100KH*1, 100KH*2)	Mercury-contaminated sludge from sulfuric acid storage tanks.	1955-1971	3.7 x 10 ⁴ L	48	Reference 6, pages 181-183, 185-187, 189- 191, 197-199, 201-203



Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Miscellaneous Floor and Sink Drain Cribs (116-B-9, 116-D-6, 116-F-11)	Drainage from floor and sink drains. Very low levels of radioactive contamination. No chemical contamination.	1952-1967	3.4 x 10 ⁵ L	440	Reference 6, pages 21-22, 69-70, 123-124
100 Area Potassium Borate Crib (116-DR-7)	Liquid potassium borate solution. No radioactive contamination.	1953	4 x 10 ³ L	5.2	Reference 6, pages 97-98
100 Area Gas Purification Condensate Cribs (116-KE-1, 116-KW-1)	Condensate from reactor gas purification system. Contaminated with tritium and carbon-14 and small amounts of fission products.	1955-1971	1.6 x 10 ⁶ L	2.1 x 10 ³	Reference 6, pages 175-176, 193-194
· 100 Area Ion Exchange Regenerant Crib (116-KE-2)	Wastes from regeneration of ion exchange columns. Contained sulfuric acid, caustic, and mixed fission products and activation products.	1955-1971	3.0 x 10 ⁶ L	3.9 x 10 ³	Reference 6, pages 177-178

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Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Retention Basin Leakage (107-B, 107-C, 107-D, 107-DR, 107-F, 107-H, 107-KE, 107-KW)	Reactor effluent containing very small amounts of fission products and activation products. Chemical contamination limited to small amounts of sodium dichromate added to reactor coolant as oxidant.	1944-1971	3.0 x 1012 L	3.9 x 10 ⁹	Reference 16
100 Area White Bluffs Pickling Acid Crib	Waste nitric and hydrofluoric acid used to pickle galvanized piping. Acid may or may not have been neutralized.	1943-1945	7 х 10 ⁵ L	900	Reference 16
		TOTAL	3.3 x 1012 L	4.3 x 10 ⁹	

TABLE 2. 100 AREA SOLID WASTE SITES

Area No. and Name	Waste Types	<u>Years</u>	Estimated Volume	No. of Cubic Yds	References
100 Area Metallic Waste Burial Grounds (118-B-1, 118-B-4, 118-B-5, 118-C-2, 118-D-1, 118-D-3, 118-D-5, 118-DR-1, 118-F-1, 118-F-3, 118-F-7, 118-H-1, 118-H-2, 118-H-3, 118-H-4, 118-H-5, 118-K)	Metallic waste consisting mainly of discarded reactor hardware, including such items as aluminum tubes and spacers; lead and cadmium containing slugs, graphite, desiccant, aluminum and boron splines; lead; and mercury. Radioactive contaminants, primarily activation products.	1944-1975	1,000 m ³	1,300	Reference 6, pages 25-27, 33-34, 35-36, 55-56, 71-72, 77-79, 83-84, 101-102, 129-131, 135-136, 143-144, 155-157, 159-160, 161-162, 163-164, 165-166, 205-207
100 Area Construction Waste Burial Grounds (118-B-2, 118-B-3, 118-D-4, 118-F-1, 118-H-3)	Construction waste, demolition debris. Small amounts of radioactive contamination.	1953-1967	48,000 m ³	63,000	Reference 6, pages 29-30, 31- 32, 81-82, 129- 131, 161-162
100 Area Lab Waste Burial Grounds (118-8-6, 118-f-5, 118-f-6)	Low-level radioactive wastes from 100 Area laboratories, including tritium wastes and animal wastes.	1950-1975	20,000 m ³	26,000	Reference 6, pages 37-38, 139-140, 141-142

100 m

Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Miscellaneous Solid Waste Burial Grounds (118-B-1, 118-B-7, 118-C-1, 118-D-1, 118-D-2, 118-F-2, 118-F-4, 118-H-1, 118-K)	Radioactivity contaminated trash and miscellaneous solid wastes from 100 Area facilities.	1944-1975	61,000 m ³	80,000	Reference 6, pages 25-27, 39- 40, 51-53, 71- 72, 73-75, 133- 134, 137-138, 155-157, 205-207
100 Area Burn Pits (100 B/C Burn Pit, 100/DR Burn Pit, 100F Burn Pit, 100N Burn Pit, 100K Burn Pit)	Combustible trash from 100 Area buildings and offices.	1943-1971	130,000 m ³	170,000	Reference 6, pages 57-58, 103-104, 145- 146, 169-170, 209-210
		TOTAL.	260,000 m ³	340,000 yd ³	

5 TARGETS

Ground Water Use

Use(s) of aquifer(s) of concern within a 3-mile radius of the facility:

There are no wells which draw water for drinking or irrigation purposes within 3 miles of the 100 Area. Therefore, there is no use of the aquifer of concern within 3 miles of the 100 Area.

Reference 8

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Distance to Nearest Well

Location of nearest well drawing from <u>aquifer of concern</u> or occupied building not served by a public water supply:

The nearest wells are 699-72-101A, B, and C that are located at the Vernita Bridge Rest Stop.

Reference 8; Reference 9, page II.1-65

Distance to above well or building:

The wells located at the Vernita Bridge Rest Stop are just over 3 miles from the boundary of the 300 Area Site.

Reference 8

Population Served by Ground Water Wells Within a 3-Mile Radius

Identified water-supply well(s) drawing from aquifer(s) of concern within a 3-mile radius and populations served by each:

There are no wells which draw water for drinking or irrigation purposes within 3 miles of the 100 Area. Therefore, there is no use of the aquifer of concern within 3 miles of the 100 Area.

Reference 8

Computation of land areas irrigated by supply well(s) drawing from aquifer(s)
of concern within a 3-mile radius, and conversion to population (1.5 people per acre):

There are no documented irrigation wells within 3 miles of the site.

Total population served by ground water within a 3-mile radius:

The total population served by ground water within 3 miles is zero.

SURFACE WATER ROUTE

1 OBSERVED RELEASE

Contaminants detected in surface water at the facility or downhill from it (5 maximum):

The observed release of contaminants to the Columbia River adjacent to the 100 Area can be verified by comparison of upstream and downstream concentrations of selected constituents. Verification of observed release does not require evidence of release for each contaminant disposed at the facility, but relies on identification of at least one constituent in the surface water that can be attributed to the facility. Strontium-90 (Sr-90) was chosen to illustrate observed release from the 100 Area. The following table compares upstream concentrations of Sr-90 with concentrations of Sr-90 at designated surface water use points. The water intakes designated as use points are in the 100N and 100D areas.

BACKGROUND				DOWN GRADIENT(a)			
Date	Sample Point	Concentration (pCi/L)	••	Date	Sample Point	Concentration (pCi/L)	
11/23/82- 12/21/82	Prst Rap	0.34 * 0.18		12/18/82 12/18/82	100N 100D	28 ± 0.47 1.1 ± 0.05	

(a) This data represents an analysis of the composite of river water samples taken every 0.5 mile for the river interval indicated on the date shown.

Reference 11; Reference 5, page 3.14; Reference 12, pages 15 and B.3; Reference 10

Rationale for attributing the contaminants to the facility:

The nine 100 Areas (B, C, D, DR, KE, KW, F, H, and N) border the Columbia River in the northernmost part of the Hanford Site. Each of the nine areas has one production reactor. Eight of these reactors have been shut down; only the N Reactor, used for both plutonium and electricity production is still in operation. When the reactors were operational, cooling water was drawn from the river and treated with alum, sulfuric acid and chlorine. Excess sulfuric acid was used to maintain the pH of the water within a desired range. To control oxidation of aluminum parts in the reactor, sodium dichromate was used to maintain an oxidation coating on aluminum parts. The chlorine was added for algae control in the settling basins; at times copper sulfate was added for additional algae control. As the cooling water passed through the reactor various activation products (e.g, Co-60, Ni-63, etc.) were dissolved in the cooling water. When a fuel element ruptured or developed a pin hole in its' cladding some fission products such as Sr-90 would dissolve in the cooling water. Chromic acid, oxalic acid and nitric

acid were used for dummy fuel-element decontamination. In addition to vertical safety rods for emergency reactor shutdown, the reactors were equipped with hoppers of nickel-plated boron steel balls, nickel-plated carbon steel balls, and stainless steel balls that would drop into the vertical safety rod channels for emergency shutdown. This system required no supplementary power source. Although it was never used, a third safety system, one involving the use of potassium borate solution, was in place at the reactors. A supplementary control system, in addition to the normal horizontal control rods, was incorporated into the reactors. This supplementary control system consisted of a Poison(a) Column Control Facility that could charge selected process tubes with a lead-cadmium poison to absorb neutrons. Boron-carbide aluminum poison splines were also used for supplementary control.

These activities resulted in strontium-90 contaminated waste being disposed of in the 100 Area, and down gradient samples show strontium-90 concentration levels significantly above the strontium-90 concentrations found in background samples.

Reference 4, pages 2.11-2.13

2 ROUTE CHARACTERISTICS

Facility Slope and Intervening Terrain

Average slope of facility in percent:

Not applicable.

Name/description of nearest down slope surface water:

The Columbia River, which originates in the mountains of eastern British Columbia, Canada, flows through the northern edge of the Hanford Site and forms part of the Hanford Site's eastern boundary. The river drains a total area of approximately 70.800 km² enroute to the Pacific Ocean. The flow of the Columbia River is regulated by 11 dams within the United States, 7 upstream and 4 downstream of the Site. Priest Rapids Dam is the nearest impoundment upstream of the Site, and McNary Dam is the nearest dam downstream. (The Hanford reach of the Columbia River extends from Priest Rapids Dam to the head of Lake Wallula, which is created by McNary Dam.) This is the only stretch of the Columbia River within the U.S. that is not impounded by a dam. The width of the river wries from approximately 300 m to about 1000 m. The flow through this stretch of the river is relatively swift, with numerous bends and several islands present throughout the reach. The ground water beneath the site discharges directly into the Columbia River as evidenced by seeps and springs along the river shore.

⁽a) The term poison refers to a material's ability to absorb neutrons and thus control the rate of fission.

The flow rate of the Columbia River in this region is regulated primarily by Priest Rapids Dam. Hanford reach flows fluctuate significantly because of the relatively small storage capacity and operational practices of the nearby upstream dams. A minimum flow rate of 1,000 cubic meters per second (36,000 cubic feet per second) has been established at Priest Rapids. Typical daily flows range from 1,000 cubic meters per second (36,000 cubic feet per second) to 7,000 cubic meters per second (250,000 cubic feet per second) with peak spring runoff flows of up to 12,600 cubic meters per second (450,000 cubic feet per second) being recorded. Typical annual average flows at Priest Rapids Dam are 3,100 cubic meters per second (110,000 cubic feet per second) to 3,400 cubic meters per second (120,000 cubic feet per second). Monthly mean flows typically peak from April through June and are at the lowest levels from September through October.

Reference 5, page 2.1-2.3

Average slope of terrain between facility and above-cited surface water body in percent:

Not applicable.

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Is the facility located either totally or partially in surface water?

. Not applicable.

Is the facility completely surrounded by areas of higher elevation?
Not applicable.

1-Year 24-Hour Rainfall in Inches

Not applicable.

Distance to Nearest Down slope Surface Water

Not applicable.

Physical State of Waste

Not applicable.

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3 CONTAINMENT

Containment

Method(s) of waste or leachate containment evaluated: Not applicable.

Method with highest score:

Not applicable.

4 WASTE CHARACTERISTICS

Toxicity and Persistence

Compound(s) evaluated

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These substances are associated with production reactor operation (See Tables 1 and 2):

Substances in the 100 Area

uranium lead cadmium sodium dichromate sodium oxalate sulfuric acid sulfamic acid	mercury plutonium-238 plutonium-239 plutonium-240 tritium cobalt-60 strontium-90	cesium-137 cesium-134 europium-152 europium-154 europium-155 sodium sulfamate nickel-63
potassium borate	copper sulfate	sodium hydroxide

Reference 4, pages 2.11-2.13; Reference 6 (The numerous individual pages are not included in the package, please consult the reference 6 document using the list of individual sites listed on the cover sheet of this package for verification of which individual sites contain which substances); Reference 19

Compound with highest score:

Several of these substances results in a score of 18. Uranium, lead, mercury sodium dichromate and plutonium are among those having scores of 18.

Substance	Toxicity Score	Persistence <u>Score</u>	TOTAL Score
uranium	3	3	18
lead	3	3	18
mercury	, 3	3	18
dichromate	3	3	18
plutonium	3	3	18

Reference 7, pages 794-797; Reference 20.

<u>Hazardous Waste Quantity</u>

Total quantity of hazardous substances at the facility, excluding those with a containment score of 0 (Give a reasonable estimate even if quantity is above maximum):

The total quantity of hazardous substances associated with the aggregate 100

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Area Site is estimated to be 4.3 billion cubic yards. Table 1 presents the individual sites and associated data for the 100 Area Liquid Waste Sites that were used in creating the aggregate 100 Area Site. Table 2 presents the individual sites and associated data for the 100 Area Solid Waste Sites that were used in creating the aggregate 100 Area Site.

References are listed by waste site in Table 1.

Basis of estimating and/or computing waste quantity:

The volume of wastes disposed of to waste sites at the 100 Area was calculated as the sum of the volumes received by individual waste sites. The basis for this estimate of liquid waste volume is given in Table 1. For purposes of developing the inventories, individual waste sites were aggregated into groups that received similar wastes. These groups, and the individual sites comprising each group, are identified in Table 1. Also included in this table are a description of the general types of waste received by the group of sites, the period of operation covered by the group, the estimated waste volume received by the group, and the references for the waste volumes received by the individual sites. In some cases, an individual site received more than one type of waste. If so, the site was assigned to more than one group and its waste volume was divided into the volumes received by each particular site group. Similar information for 100 Area sites receiving solid wastes is presented in Table 2.

In compiling the inventory, since groundwater seeps into the Columbia River, it is assumed that the containment associated with the surface water route is considered not to be 0, and all the waste quantities cited were available for migration.

Reference 12; Reference 19.

5 TARGETS

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Surface Water Use

Use(s) of surface water within 3 miles downstream of the hazardous substance:

The Columbia River is used as the source of drinking and process water for the 100 and 200 areas. The intake at 100-B Pumphouse supplies water to 100-B/C, 100-D, 100-N, 100-K, and the 200 areas. The intake at 100-D is used as a backup to 100-B intake. The 100-N intake supplies water to 100-N area and Washington Public Power Supply System Generation Station attached to the 100-N reactor. The 100-K Pumphouse supplies water to 100-K area.

The Columbia River both near and down stream of the 100 Area is used for recreation.

Reference 9, pages II.1-64-II.1-65; Reference 13, page 3; Reference 17.

TABLE 1. 100 AREA LIQUID WASTE SITES

Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Decontamination Waste Cribs (116-B-4, 116-B-6, 116-C-2, 116-D-1A, 116-D-1B, 116-F-5, 116-F-10, 116-H-3, 116-K-2)	Decontamination solutions consisting of chromic, oxalic, and/or sulfuric acid solutions neutralized with soda ash. Solutions were used to decontaminate dummy fuel elements and reactor hardware.	1947-1970	2.3 x 106 L	3,000	Reference 6, pages 13-14, 17- 20, 45-48, 59- 60, 61-62, 113- 114, 121-122, 151-152, 173-174
100 Area Contaminated Reactor Effluent Cribs and Trenches (116-B-1, 116-B-3, 116-C-1, 116-D-2, 116-DR-1, 116-DR-2, 116-F-2, 116-F-3, 116-F-4, 116-H-1, 116-H-4, 116-K-1, 116-K-2)	Reactor effluent contaminated when fuel element cladding failed. Radiological contamination consisted chiefly of mixed fission products. Chemical contamination consisted of sodium dichromate added to reactor coolant as oxidant.	1946-1971	3.0 x 10 ¹¹ L	3.9 x 10 ⁸	Reference 6, pages 7-8, 11-12, 43-44, 63-64, 87-88, 89-90, 93-94, 105-106, 107-108, 109-110, 111-112, 147-148, 153-154, 171-172, 173-174

Area No. and Name	Waste Types	<u>Years</u>	Estimated Volume	No. of Cubic Yds	References
100 Area Laboratory Waste Cribs and Trenches (116-B-5, 116-B-10, 116-C-2, 116-D-3, 116-D-4, 116-F-9, 116-F-13)	Wastes associated with labs and shops in the 108 Buildings, including 108-B Tritium Lab and Tube Examination Facility, 108-C Metal Examination Facility and 100 F Biology Labs. Contaminants consisted mainly of tritium and low levels of mixed fission and activation products. Negligible chemical contamination except Biology Lab facilities, which received nitrate from animal wastes.	1950~1976	3.2 x 10 ⁸ L	4.2 x 10 ⁵	Reference 6, pages 15-16, 23- 24, 45-48, 65- 66, 67-68, 119- 120, 127-128
100 Area Fuel Storage Basia Cribs and Trenches (116-8-2, 116-9-1A, 116-8-1B, 116-98-3, 116-F-3, 116-KE-3, 116-KH-2)	Storage basin cooling water contaminated with low levels of mixed fission products due to failure of fuel element cladding.	1946-1971	2.0 x 10 ⁷ l.	2.6 x 10 ⁴	Reference 6, pages 9-10, 59- 60, 61-62, 91- 92, 109-110, 179-180, 195-196

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Area No. and Name 100 Area Reactor Decontamination and Maintenance Effluents Cribs and Trenches (116-DR-6, 116-F-1, 116-F-6, 116-F-12, 116-H-2, 116-K-2)	Reactor effluents during reactor maintenance activities. Much of the waste volume was uncontaminated. Radionuclides present were mainly low levels of mixed fission products and activation products. The main chemical contaminant was sulfuric acid, which	Years 1945-1971	Estimated Volume 6.1 x 108 L	No. of Cubic Yds 8.0 x 105	Reference 6, pages 95-96, 105-106, 115-116, 125-126, 149-150, 173-174
100 Area Reactor Confinement Seal Pit Drainage Cribs (116-DR-8, 116-F-7, 117-B, 117-C, 117-D, 117-H)	was used for reactor decontamination. Sodium dichromate was also present in some of the effluents. Drainage from 117 Building confinement system seal pits. Very low levels of contamination.	1960-1969	2.2 x 10 ⁶ L	2,900	Reference 6, pages 42-44, 49- 50, 85-86, 99- 100, 117-118, 167-168
100 Area Sulfuric Acid Sludge Disposal Sites (100KE*1, 100KE*2, 100KE*3, 100KW*1, 100KW*2)	Mercury-contaminated sludge from sulfuric acid storage tanks.	1955-1971	3.7 x 10 ⁴ L	48	Reference 6, pages 181-183, 185-187, 189- 191, 197-199, 201-203



Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Miscellaneous Floor and Sink Drain Cribs (116-B-9, 116-D-6, 116-F-11)	Drainage from floor and sink drains. Very low levels of radioactive contamination. No chemical contamination.	1952-1967	3.4 x 105 L	440	Reference 6, pages 21-22, 69- 70, 123-124
100 Area Potassium Borate Crib (116-DR-7)	Liquid potassium borate solution. No radioactive contamination.	1953	4 x 10 ³ L	5.2	Reference 6, pages 97-98
100 Area Gas Purification Condensate Cribs (116-KE-1, 116-KW-1)	Condensate from reactor gas purification system. Contaminated with tritium and carbon-14 and small amounts of fission products.	1955-1971	1.6 x 106 L	2.1 x 10 ³	Reference 6, pages 175-176, 193-194:
100 Area Ion Exchange Regenerant Crib (116-KE-2)	Wastes from regeneration of ion exchange columns. Contained sulfuric acid, caustic, and mixed fission products and activation products.	1955-1971	3.0 x 10 ⁶ L	3.9 x 10 ³	Reference 6, pages 177-178



Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Retention Basin Leakage (107-B, 107-C, 107-D, 107-DR, 107-F, 107-H, 107-KE, 107-KW)	Reactor effluent containing very small amounts of fission products and activation products. Chemical contamination limited to small amounts of sodium dichromate added to reactor coolant as oxidant.	1944-1971	3.0 x 10 ¹² L	3.9 x 10 ⁹	Reference 16
100 Area White Bluffs Pickling Acid Crib	Waste nitric and hydrofluoric acid used to pickle galvanized piping. Acid may or may not have been neutralized.	1943-1945	7 x 10 ⁵ L	900	Reference 16
		TOTAL	3.3 x 10 ¹² L	4.3 x 10 ⁹	

TABLE 2. 100 AREA SOLID WASTE SITES

Area No. and Name	Waste Types	Years	Estimated Volume	No. of Cubic Yds	References
100 Area Metallic Waste Burial Grounds (118-B-1, 118-B-4, 118-B-5, 118-C-2, 118-D-1, 118-D-3, 118-D-5, 118-DR-1, 118-F-1, 118-F-3, 118-F-7, 118-H-1, 118-H-2, 118-H-3, 118-H-4, 118-H-5, 118-K)	Metallic waste consisting mainly of discarded reactor hardware, including such items as aluminum tubes and spacers; lead and cadmium containing slugs, graphite, desiccant, aluminum and boron splines; lead; and mercury. Radioactive contaminants, primarily activation products.	1944-1975	1,000 m ³	1,300	Reference 6, pages 25-27, 33- 34, 35-36, 55- 56, 71-72, 77- 79, 83-84, 101- 102, 129-131, 135-136, 143- 144, 155-157, 159-160, 161- 162, 163-164, 165-166, 205-207
100 Area Construction Haste Burial Grounds (118-B-2, 118-B-3, 118-D-4, 118-F-1, 118-H-3)	Construction waste, demolition debris. Small amounts of radioactive contamination.	1953-1967	48,000 m ³	63,000	Reference 6, pages 29-30, 31- 32, 81-82, 129- 131, 161-162
100 Area Lab Waste Burial Grounds (118-8-6, 118-F-5, 118-F-6)	Low-level radioactive wastes from 100 Area laboratories, in-cluding tritium wastes and animal wastes.	1950-1975	20,000 m ³	26,000	Reference 6, pages 37-38, 139-140, 141-142



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Area No. and Name	Waste Types	<u>Years</u>	Estimated Volume	No. of Cubic Yds	References
100 Area Miscellaneous Solid Waste Burial Grounds (118-B-1, 118-B-7, 118-C-1, 118-D-1, 118-D-2, 118-F-2, 118-F-4, 118-H-1, 118-K)	Radioactivity contaminated trash and miscellaneous solid wastes from 100 Area facilities.	1944-1975	61,000 m ³	80,000	Reference 6, pages 25-27, 39- 40, 51-53, 71- 72, 73-75, 133- 134, 137-138, 155-157, 205-207
100 Area Burn Pits (100 B/C Burn Pit, 100/DR Burn Pit, 100F Burn Pit, 100N Burn Pit, 100K Burn Pit)	Combustible trash from 100 Area buildings and offices.	1943-1971	130,000 m ³	170,000	Reference 6, pages 57-58, 103-104, 145- 146, 169-170, 209-210
		TOTAL.	260,000 m ³	340.000 vd3	



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Is there tidal influence?

The site is located upstream of four dams, therefore, there is not tidal influence at the site.

Reference 5, page 2.1

Distance to a Sensitive Environment

Distance to 5-acre (minimum) coastal wetland, if 2 miles or less: Not applicable.

Distance to 5-acre (minimum) fresh-water wetland, if 1 mile or less:

No fresh-water wetlands were found near the Hanford Site.

Reference 8

Course

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Distance to critical habitat of an endangered species or national wildlife refuge, if 1 mile or less:

Although there are several sensitive and threatened species that are residents of Hanford Site (for at least part of the year), no endangered species are known to be residents of the site. Two threatened species, the bald eagle and the ferruginous hawk, are residents of the site (for at least part of the year). Because there are no endangered species (federal listing) that reside at the site, there is no critical habitat to be considered in the ranking of the 300 Area Site.

The Draft Phase I Installation Assessment of Inactive Waste-Disposal Sites at Hanford, Volume 1, July 1986 listed the Merriam's Shrew as being on the State Endangered Species List. This information was taken from a list published in a preliminary draft of an environmental impact statement, and, since the referenced February 1987 list shows the Merriam's Shrew as only a proposed sensitive species, it is assumed that the previous list taken from the preliminary draft is in error.

Reference 4: Reference 14

Population Served by Surface Water

Location(s) of water-supply intake(s) within 3 miles (free-flowing bodies) or 1 mile (static water bodies) downstream of the hazardous substance and population served by each intake:

The 100-B Pumphouse (at 100-B/C Area) supplies water to 100-B/C, 100-D, 100-N, 100-K and 200 areas. The 100-D Pumphouse (at 100 D/DR Area) is a backup to the 100-B Pumphouse. The water plant at 100-N supplies water to both 100-N and the Washington Public Power Supply System Generating Station which is attached to the 100-N reactor. The 100-K Pumphouse supplies water to the 100-K area. The worker population of the 100 Area is 760 and the population of the 200 Area is 2,355.

Reference 9, pages II.1-64-II.1-65; Reference 13, page 3; Reference 15, page 1; Reference 4; Reference 8

Computation of land area irrigated by above-cited intake(s) and conversion to population (1.5 people per acre):

There are no crops irrigated by the 100 Area intakes, and there are no documented irrigation intakes drawing from the columbia river within 3 miles of the 100 Area Site.

Total population served:

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The total population served is calculated by summing all the respective population estimates:

$$760 + 2,355 = 3115$$

Name/description of nearest of the above water intakes:

The intake at the 100-D Area is the 181-D Pumphouse. The intake at the 100-N Area is the 181-N Pumphouse.

Reference 4: Reference 8

Distance to above-cited intakes, measured in stream miles.

The distance to the above intakes is zero because the intakes have documented observed releases.

Reference 4; Reference 8; Reference 11; Reference 12

AIR ROUTE

1 OBSERVED RELEASE

Contaminants detected:

Even though air concentrations of some of the constituents of interest can be detected above background offsite, no air monitoring data were found sufficient for HRS scoring of the Hanford CERCLA sites. These constituents of interest detected above background offsite are present in the routine gaseous effluents from operating facilities at Hanford. Therefore, the air route rating factors were not scored.

Date and location of detection of contaminants:

Not Applicable

Methods used to detect the contaminants:

Not Applicable

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Rationale for attributing the contaminants to the site:

Not Applicable

2 WASTE CHARACTERISTICS

Reactivity and Incompatibility

Most reactive compound:

Not Applicable

Most incompatible pair of compounds:

Not Applicable

<u>Toxicity</u>

Most toxic compound:

Not Applicable

Hazardous Waste Quantity

Total quantity of hazardous waste:

Not Applicable

Basis of estimating and/or computing waste quantity:

Not Applicable

3 TARGETS

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Population Within 4-Mile Radius

Circle radius used, give population, and indicate how determined:

0 to 4 mi

0 to 1 mi

0 to 1/2 mi

0 to 1/4 mi

Not applicable

Distance to a Sensitive Environment

Distance to 5-acre (minimum) coastal wetland, if 2 miles or less: Not applicable

Distance to 5-acre (minimum) fresh-water wetland, if 1 mile or less:
Not Applicable

Distance to critical habitat of an endangered species, if 1 mile or less: Not Applicable

Land Use

Distance to commercial/industrial area, if 1 mile or less:

Not Applicable

Distance to national or state park, forest, or wildlife reserve, if 2 miles or less:

Not Applicable

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Distance to residential area, if 2 miles or less:

Not Applicable

Distance to agricultural land in production within past 5 years, if 1 mile or less:

Not Applicable

Distance to prime agricultural land in production within past 5 years, if 2 miles or less:

Not Applicable

Is a historic or landmark site (National Register or Historic Places and National Natural Landmarks) within the view of the site?

Not Applicable

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	FATION LOG SHEET SITE NAME <u>U.S. DOE Hanford - 100 Area</u> CITY <u>Benton County</u> STATE <u>WA</u> IDENTIFICATION NUMBER
REFERENCE NUMBER	DESCRIPTION OF THE REFERENCE
1	Strontium Determination ALL Matrices, UST-RD-PM-9-80
2	Computer Printouts from Hanford Ground Water Data Base
3	Hanford Wells, PNL-5397, February 1985
4	Draft Phase I Installation Assessment of Inactive Waste-
,	Disposal Sites at Hanford, Volume 1, July 1986
, 5	Environmental Monitoring at Hanford for 1986, PNL-6120,
	May 1987
6	Draft Phase I Installation Assessment of Inactive Waste-
	Disposal Site at Hanford, Volume 2, July 1986
, 7	Uncontrolled Hazardous Waste Site Ranking System; A Users
	Manual, 40 CFR 300, Appendix A
8	U.S.G.S. Maps of the Area Around 100 Area
9	Waste Management Operations, Volume 1, ERDA-1538
10	Inductively Coupled Plasma Method, Method 6010
11	Hanford Stream data base printout showing Priest Rapids Dam
12	Investigation of Ground-Water Seepage from Hanford Shoreline
	of the Columbia River, PNL-5289, November 1984
13	Hanford Reservation Area Worker Census, BNWL-2298, July 1977
14	Endangered and Threatened Wildlife and Plants, 50 CFR,
	Part 17, Subpart B, October 86

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HRS DOCUMEN	TATION LOG SHEET SITE NAME <u>U.S. DOE Hanford - 100 Area</u> CITY Benton County STATE <u>WA</u> IDENTIFICATION NUMBER
REFERENCE NUMBER	DESCRIPTION OF THE REFERENCE
15	Hanford Environmental Health Foundation Drinking Water
	Report, HEHF-45
16	Draft Phase I Installation Assessment of Inactive Waste-
	Disposal Sites at Hanford, Volume 3
17	Memo to file regarding recreational use of the Columbia River
	from DR Sherwood, August 26, 1987
18	Letter from RD Stenner to DM Bennett regarding Ground Water
	Contaminant Plume, October 14, 1987
19	Letter from RD Stenner to DM Bennett regarding Liquid Waste
	Sites and Burning Pits, October 26, 1987
20	Memo to Sandy Crystall, EPA, from Kathleen Galloway,
	MITRE. December 29, 1987.
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REFERENCE 1

Strontium Determination ALL Matrices, UST-RD-PM-9-80

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STRONTIUM DETERMINATION ALL MATRICES

Principle

Strontium is precipitated sequentially first as the nitrate, and then as the carbonate. The fuming nitric acid separation removes most of the other interfering ions and concentrated nitric precipitations remove calcium. Radioisotopes of barium, radium, and lead are co-precipitated with barium chromate. Iron and final traces of fission products are removed by a hydroxide scavenge using mixed hold back carrier. An additional barium and hydroxide scavenge is done on difficult matrices. Following a final carbonate precipitation, gravimetric yield and Strontium-89'plus Strontium-90 activity is determined or just Sr-85 tracer yield is established. Yttrium-90 is permitted to grow into equilibrium with the strontium-90, then separated from the strontium by successive hydroxide and oxalate precipitations. The yttrium oxalate is transferred to a counting dish, dried under a heat lamp, ignited to Yttrium Oxide, weighed for chemical recovery, and counted on a low background Beta Proportional Counter. Decay counts are made to check the purity of the Yttrium-90.

Scope: Procedures are described for the following matrices: milk, water, urine, feces, filter, soil, vegetation, and ion exchange resin. Food, bone, tissue and grains are treated as vegetation.

The different sample matrices analyzed as described in the scope require slightly different preparations. Refer to the Appropriate Preparation Procedure; 30-19-01, for Bioassay Samples and 30-20-01 for Environmental Matrices.

For Strontium analysis, varying sample types require even preliminary separations to bring them to a common point for the Strontium Determination steps summarized below.

Milk

Strontium is separated from milk and concentrated by equilibration with cation exchange resin using a batch technique. The resin after equilibration is washed with water to remove milk solids. Strontium is desorbed from the resin by equilibrating with 4M NaCl solution. Strontium is separated from the NaCl solution by precipitating as the carbonate.

Soil

Following leaching the sample is dissolved in 9M HCl and the majority of iron is removed using an isopropyl ether extraction. The HCl solution is converted back to the nitric form and a fuming nitric acid precipitation is performed. The nitric acid supernate is reserved for actinide (usually plutonium) chemistry.

Vegetation

In the case of vegetation (especially alfalfa) and produce, the sample residue may contain some radioactivity, depending on the sulfate content of the sample. Hence, following sample preparation and a fuming nitric acid precipation, the residue is dissolved in water and any insoluble material is boiled with a concentrated solution of Na₂CO₃ to convert the insoluble sulfate residue to the acid soluble carbonate (metathesis). The carbonate is then refluxed with nitric acid and combined with the soluble water fraction.

Water

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For large volume sample, the sample is acidified and concentrated by evaporation, neutralized with ammonium hydroxide, and strontium then is precipitated as the carbonate. Strontium is precipitated directly from small volume samples, one liter or less, by neutralizing with ammonium hydroxide to the red end point of phenolphthalein and then adding about 25 mL saturated Na₂CO₃.

Urine and Feces

Following ashing, the sample salts are dissolved in nitric acid (aliquot of a Fecal sample is analyzed), then made basic with hydroxide and carbonate heated and precipitated as the carbonate.

Ion Exchange Resin Samples

The resin is transferred to a beaker and dried and muffled for up to I week or more. Sodium carbonate (Na₂CO₃) is added and a fusion is performed. The fusion "cake" is dissolved in water and a small amount of hydrochloric acid and then reprecipitated as the carbonate.

Sequential Analyses

All samples requiring other analysis (usually plutonium analysis) in addition to strontium must first have a fuming nitric separation is performed on the sample residue instead of a carbonate.

Literature References

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Krieger, H. L. and E. L. Whittaker, <u>Prescribed Procedures for Measurement of Radioactivity in Drinking Water</u>, EPA-600/4-80-032 August 1930.

EML - Procedure Manual, HASL 300 Edited by Herbert L. Volchok and Gail de Planque, 26th edition, 1983.

Urine and Feces Sample Preparation Procedure, 30-BA-SP.

Sample Preparation Procedure For Environmental Matrices, 30-ENV-SP.

Stevenson, P. C. and Nervik, W. E., The Radiochemistry of the Rare Earths: Scandium, Yttrium, and Actinium. Subcommittee on Radiochemistry, National Academy of Science - National Research Council, Office of Technical Services, Department of Commerce, Washington, D.C.

Harley, J. P., Editor. Manual of Standard Procedures. Health and Safety Laboratory, United States Atomic Energy Commission, New York, 1976.

Reagents: All chemicals are of "reagent grade" and deionized water is used throughout this procedure.

Sr carrier: 59.4 mg Sr/mL, 143.5 g Sr(NO₃) $_2$ dissolved and diluted to 1 liter with 2M HNO₃. Calibrated to 100.0 mg SrCO₃ per mL.

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Y carrier: 20 mg Y per mL, 25.4 g YzO3 dissolved in 125 mL conc. HNO3 and diluted to 1 liter with D.I. H2O. Calibrated to 25.4 mg YzO3 per mL.

Ba carrier: 10 mg Ba $^{+2}$ /mL, 20 g Ba(NO₃) dissolved and diluted to 1 liter with D.I. water:

HNO3: (nitric acid). Fuming; 90% (or 24M HNO3) Concentrated; 15M HNO3.

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2M HNO3; 125 mL of concentrated HNO3 diluted to 1 liter with D.I. water.

HCl: (hydrochloric acid) concentrated or 12M HCl. 9M HCl; 750 mL of concentrated HCl diluted to 1 liter with D.I. water. 6M HCl; 500 mL of concentrated HCl diluted to 1 liter with D.I. water.

CH3COOH: (glacial acetic), 17.4M. 1.5M acetic acid, 86 mL glacial acetic acid diluted to 1 liter with D.I. water.

NH₄C₂H₃O₂: (ammonium acetate) 3M. 231 g of NH₄C₂H₃O₂ dissolved and diluted to 1 liter with D.I. water.

Na₂CrO₄: (sodium chromate) 1.5M. 125 g Na₂CrO₄ dissolved and diluted to 1 liter with D.I. water.

NH₄OH: (ammonium hydroxide) 14.7M or concentrated.

Na₂CO₃: (sodium carbonate) saturated aqueous - 225 g Na₂CO₃ diluted to 1 liter with D.I. water.

NaOH: (sodium hydroxide) 12M or concentrated. 540 g of sodium hydroxide dissolved and diluted to 1 liter with D.I. water.

6M NaOH; 270 g of sodium hydroxide diluted to 1 liter with D.I. water.

 $\rm H_2C_2O_4.2H_2O:$ (oxalic acid) 13% solution, 150 g of $\rm H_2C_2O_4$ dissolved and diluted to 1 liter with D.I. water.

Collodion: 10% solution, (50 mL of collodion, 300 mL of diethyl ether, 150 mL of ethyl alcohol)

Methyl red indicator, 0.1%: Dissolve 0.1 g methyl red in 100 mL D.I. water.

Phenylphthalein indicator, 0.1%: 1 g phenylphthalein dissolved in 700 mL of 95% ethanol and 300 mL D.I. water.

Mixed Co, Ce, Cs, Mn, Ni, Fe Hold Back Carrier:

This solution when prepared as below will contain 2 mg/mL Co and Ni, 1.0 mg/mL Cs, Mn, Fe, and Ce. Transfer the following into a 250 mL beaker:

- 9.91 g Ni(NO₃)₂.8H₂O (Nickel nitrate), and
- 7.24 g Fe(NO₃)_{3.942}O (Ferric nitrate), and
- 3.60 g MnCl₂.4H₂O (Manganese chloride), and
- 8.07 g CoCl₂.6H₂O (Cobalt chloride), and
- 2.34 g $Ce(NO_3)_3.6H_2O$ (Cerous nitrate)
- 1.27 g CsCl (Cesium chloride)

Add fifty mL of water and 3 mL of 8 M nitric acid. Warm gently until everything has dissolved, then transfer quantitatively to a 1 liter volumetric flask. Bring the volume to 1000.0 mL mark, mix well, then pour into a storage bottle without rinsing the volumetric flask.

Reagents for milk only:

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Cation Exchange Resin - Dowex 50W-X8 (Na form), 50-100 mesh size, available from Bio-Rad Laboratories. (The commercially available H+ form is converted into the Na+ form by mixing 1.51 of 4M NaCl and 200 mL of resin on a magnetic stirrer for 30 minutes. The resin is filtered in a large Buchner funnel and washed with deionized water until the wash water is Cl-free when tested with 10% AqNO3. If a white AqCl precipitated forms, additional water washings are necessary. It is advantageous to prepare enough resin for several samples ahead of time.)

4M NaCl: 234 g NaCl dissolved in a l liter graduated mixing cylinder in distilled water, then diluted to l liter.

10% AgNO3 Silver Nitrate: 50 grams of AgNO3 dissolved in 300 mL of deionized water. Store in dark bottle, away from light.

Equipment and Materials

Beakers, 2 liter, 1 liter, 250 mL Centrifuge Centrifuge tubes, 100 mL plastic and/or glass, 30 mL conical bottom Hot plate Heat lamp Meeker burner Magnetic stirrer and stirring bar
Filter paper 7 cm GFA, Whatman #42 12.5 cm
Buchner funnel, 7 cm
Funnel, 600
Side arm flask, 1000 mL
125 mL separatory funnel and stopper
Wrist action shaker
Platinum crucible, 150 mL
Muffle furnace 500°C, 1000°C
Stainless steel planchet 1 1/2", 1" and cardboard holders
Plastic jar 250 mL or plastic vial, 25 mL
Balance, 100 q

Calculation of Strontium Activities in Urine, Feces, and Environmental Matrices

The calculations for manual mode or computer mode (CALCTL for ERA samples, PERCTL for BA samples) are basically the same except that in the manual mode, results obtained should be considered as tentative, as only the uncertainty from counting statistics are considered.

If the program is not available, use the following calculation:

Sr-89 or Sr-90 decay corrected value =

(Sr-89 or Sr-90 reference value) x e-K t

where:

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- t = difference in reference date and current date
- "K" = decay constant for Sr-89 and Sr-90, expressed in the same time units as "t" = $.693/t_{1/2}$.
- Total strontium activity of the sample and the corresponding counting uncertainty are calculated as follows:

Sr (Total) dpm per sample or pCi per unit =

Counting Uncertainty dpm or [pCi] per unit or [sample] =

[2] $\times SQR (C_1/T_1 + B/T_B) \times (dpc)_{Tot} \times [V \text{ or } W]$ [2.22] $\times Y_S \times (V_A \text{ or } W_A)$

where:

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- [] = The square brackets indicate parameters exclusively used for environmental or bioassay analyses only.
- (dpc)_{Tot} = Disintegrations per count value corresponding to the weight of Sr(NO₃)₂ in mg from the dpc table. Sr-89 and Sr-90 activities are assumed to contribute equally to the total activity.
- C1 = Gross cpm from the first strontium count
- T; = Sample counting time in minutes
- B = Background cpm
- Ta = Background counting time in minutes
- Ys = Chemical recovery of strontium expressed as a decimal
- [V. or W] = Volume or weight of the sample in the same units as V_{Δ} or W_{Δ} (applicable for bioassay samples only)
- V_A or W_A = Volume or weight of the aliquot in the required unit
- [2] = Two sigma uncertainty reported for environmental samples only
- 2. Sr-90 activity of the sample and the corresponding counting uncertainty from the Y-90 milking is calculated as follows:

Sr-90 dpm per sample or pCi per unit =

$$(C_{Y_1} - B) \times e^{K_2 + 2} \times (dpc)_{Y_2} \times [V \text{ or } W]$$

$$[2.22] \times I \times Y_S \times Y_V \times V_A \text{ or } W_A$$

. Counting Uncertainty dpm per sample or pCi per unit =

[2] x SQR $(C_Y/T_Y + B/T_B)$ x e^{KZtZ} x $(dpc)_{Y_Z}$ x [V or W] [2.22] x I x Y_S x Y_Y x V_A or W_A

where:

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- [] = parameters exclusively used for environmental or bioassay analysis
- Cy = Gross Y-90 cpm
- Ty = Time of Y-90 counting in minutes
- B = Background cpm
- TB = Length of background counting in minutes
- eK2t2 = Correction for the decay of Y-90 during the time (t2 hours) elapsed between Y-90 milking and counting. "K" is the decay constant for Y-90 = 0.693/64.1 hours.
- (dpc)Y2 = Disintegrations per count of Y-90 corresponding to the weight of yttrium oxide and the counter used
- [V or W] \Rightarrow Volume or weight of the total sample in the same units as V_A or W_A for Bioassay only
- V_A or W_A = Volume or weight of the sample aliquot in the units required
- [2.22] * Factor, dpm or pCi (Applicable for environmental samples only)
- Correction for the fractional ingrowth of Y-90 in the strontium sample at the time of milking (from the table)
- Y_s = Gravimetric recovery of strontium as strontium carbonate
- Yy = Gravimetric recovery of yttrium oxide (weight of final Y2O3 precipitated (mg)/25.4 mg expected).
- 3. Sr-89 Determination: Sr-89 activity is determined by subtracting Sr-90 activity and Y-90 ingrowth, which are estimated from Y-90 milking and determination, from the total strontium activity observed during the first strontium count with appropriate corrections for self absorption, scattering, and counting efficiency.

Sr-89 dpm per sample or pCi per unit =

$$(C_1 - B) - \frac{(dpc)_{S_{90}}}{(dpc)_{S_{90}}} - \frac{(dpc)_{Y_1}}{(dpc)_{Y_1}} * (1 - e^{-K_1t_1})$$

$$\times$$
 (dpc)_{S89} \times $e^{K_3t_3}$ \times V or W [2.22] \times Y_s \times V_A or W_A

where:

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= Parameters exclusive for environmental or bioassay analyses

(dpm)₅₉₀

= Disintegrations per minute of Sr-90 nominal value before correcting for factors, such as 2.22, Ys, volume and weight terms as shown below:

 $(dpc)_{Sqn}$, $(dpc)_{Y1}$, $(dpc)_{S89}$ = Disintegration per count of Sr-90, Y-90, and Sr-89 respectively corresponding to the counter used and the weight of strontium carbonate.

 $(1 - e^{-K}l^{t}l)$

= Factor expressing Y-90 ingrowch in the strontium sample during the time (t₁ in hours) elapsed between the time "to" of the first Y-90 removal from the strontium sample and the first (strontium) count. "K;" is the decay constant for Sr-90 = .693/28.8 years.

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= Factor, dpm per pCi - for environmental samples only.

[V or W]

= Volume or weight of the sample . in the same units as V_A or W_A only for Bioassay analysis.

eKztz

= Correction for the decay of Sr-89 from the date of first strontium counting to the sample date. "K3" is the decay constant for Sr-89 = 0.693/50.5 days.

Counting Uncertainty dpm per sample or pCi per unit =

where:

All the known uncertainties are propagated using a computer program.

PRE-SEPARATION:

Milk

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- I(M). Add 100 cc of Dowex 50W-X8 resin (Na-form)
 equilibrated as described in the reagent section, to
 the milk sample aliquots and stir for 3 hours using a
 magnetic stirrer. Allow the resin to settle for 5-10
 minutes. Decant as much of the milk as possible into
 a separate beaker without losing any resin.
 - 2(M). Add about 500 mL of D.T. Water to the resin beaker and stir for about one minute using magnetic stirrer. Allow the resin to settle, decant and discard the supernate. Repeat this process until visible milk solids (cloudiness) have been removed from the resin.
 - 3(M). Add 400 mL of 4M NaCl to the resin sample and stir for 15 minutes. Allow the resin to settle and filter through a 7-Cm GFA filter and collect the filtrate in a 1000 mL Erlynmeyer flask. Transfer the NaCl solution from the flask to a clean 1000 mL beaker. Transfer the resin from the filter reservoir back to the resin beaker, add 200 mL of 4M NaCl, stir for 15

min., filter under is before vacuum and rinse the resin with H₂O, collect the filtrate in the 1000 mL Erlynmeyer flask. Add the second NaCl solution and rinse from the flask to the 1000 mL beaker containing the first NaCl solution as wash solution. Discard the resin to waste.

- 4(M). Add 2-3 drops of phenolphthalein indicator to the NaCl solution, then add conc NH4OH while stirring until the solution turns pink (end point). Add 70-100 mL of saturated Na2CO3, and heat for 10 minutes, then cool to room temperature. Decant most of the supernate to waste. Centrifuge to collect the precipitate remaining.
- 5(M). Proceed to the Strontium Determination Section.

KEYPOINT: Exercise extreme caution when performing the fuming nitric acid step since a vigorous reaction is possible if any resin is remaining.

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- 1(S). Evaporate the 8M HNO₃ sample leachate to dryness, then add about 5 mL conc. HCl and carefully evaporate to near dryness at least two times.
- 2(5). Dissolve the sample in 9M HCl, transfer to a separatory funnel, add about 20 mL of isopropyl ether and shake with frequent venting for 1-2 minutes. Reject the ether (upper phase). Repeat the isopropyl ether extraction two more times or until the yellowish tint indicating iron does not appear in the ether phase.
- 3(S). Transfer 9M HCl phase back to the original beaker and evaporate to dryness. Convert the sample back to the nitric media using 2-10 mL additions of conc. nitric acid and cooperating to dryness after each addition.
- 4(S). Proceed to step one of the <u>Strontium Determination</u>
 Section.

Water

1(W). Add saturated Na₂CO₃ directly to the water sample or to the water sample that has been concentrated by evaporating and allow the precipitate to settle overnight. Decant most of the supernate to waste and collect the precipitate in a centrifuge tube. Dissolve the precipitate in about 5 mL of 2M HNO₃, dilute to 20 mL with D.I. H₂O and proceed to step 4

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of the Strontium Determination Section.

Urine and Feces

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- 1(U). Dissolve sample salts by adding about 50 mL of D.I. water, warm and add 5 to 10 mL of conc. HNO3. Add phenolphthalein indicator, and adjust to the red end point using concentrated NH40H. Add 25 mL of saturated Na2CO3, heat for 2-5 min. (metathesis), then cool for 10 minutes.
- 2(U). Transfer, using D.I. water to 1 or 2 100 mL plastic centrifuge tubes and allow the sample to cool for at least 1/2 hour. Centrifuge and decant the supernate to waste. Add about 30-40 mL of D.I. Water. Stir, centrifuge, and decant the supernate to waste.
- 3(U). Proceed to the Strontium Determination Section.

Ion Exchange Resin Samples

- 1(R). A prepared amount (see 30-20-01) of strontium-85 and strontium carrier is added to the sample in its original container. Add a small amount of deionized water, cap the container and shake vigorously. Transfer the sample quantitatively to a 600 mL beaker (labeled with a Tech pen) with DI water followed by a rinse with about 5 mL of concentrated nitric acid. The entire sample is again mixed with a stirrer and dried under a heat lamp.
- 2(R). The sample is ashed to a pinkish white residue at 500-550°C in a muffle furnace (process may take up to a week or more) and then transferred quantitatively to a platinum crucible. Na₂CO₃ is added in an amount equal to 4-5 times that of the sample weight and mixed in thoroughly. The mixture is fused at 900-1000°C for approximately 1-2 hours.
- 3(R). The platinum crucible is placed in a 600 mL beaker and the fusion mixture is dissolved with approximately 200 mL of hot water. The platinum crucible is removed and rinsed with a minimum of 6M HCl.
- 4(R). Approximately 25 mL of sat. Na₂CO₃ is added to the sample with stirring. After allowing the precipitate to settle the majority of the supernate is decanted to waste and the remaining is centrifuged. Proceed with the strontium carbonate precipitate to the Strontium Determination Section.

STRONTIUM DETERMINATION

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SPECIAL SAFETY PRECAUTION: Fuming nitric acid is used in this procedure. It can be hazardous; especially when brought in contact with organics, therefore exercise extreme care.

All waste fractions are generally saved to the end of the procedure; after acceptable recoveries are received they may be discarded.

1. Dissolve the sample residue or carbonate precipitate in a minimum amount of HNO₃ while stirring. Add 50-70 mL of <u>fuming</u> nitric acid and cool in an ice bath for 20-30 minutes. Centrifuge and <u>save</u> the supernate for <u>Sequential Analysis</u> if requested; proceed to the plutonium procedure (20-Pu-01) with this fraction.

The fuming nitric acid addition may be done in a beaker first, then transferred to centrifuge tubes, especially in the case of soil and vegetation.

KEY POINT: The <u>fuming</u> nitric precipitation is an excellent separation point for sequential chemistry and is usually included in the sample preparation chemistry (30-ENV-SP).

- 2. To the precipitate, add 30-40 mL of conc. KNO₃ to remove excess calcium. Stir, centrifuge and decant to waste. Repeat one or more times depending on the calcium content of the particular matrix.
- 3. Add 20-30 mL of warm D.T. water. Stir to dissolve completely. If any insoluble material is present separate this precipitate by centrifuging and decanting the supernate to a clean centrifuge tube.

For Vegetation Samples (or any other sample that has been muffled) when insoluble residue remains after dry washing: (metathesis) Transfer the insoluble material to a beaker with ZM HNO3, add excess Na2CO3 and boil for 15 minutes. Transfer back to the tube, centrifuge and decant supernate to waste. Dissolve precipitate in 5-15 mL conc. HNO3, warm, stir, centrifuge and combine this supernate with the water soluble fraction.

KEY POINT: Insoluble material not undergoing metathesis should be retained until an acceptable final precipitate weight

recovery or tracer recovery is established, then it can be discarded.

- 4. Add 1-2 drops of phenolphthalein indicator and adjust dropwise just to Basic (pink endpoint) using conc.

 NH4OH, then add 5 drops in excess. Add 1 or 2 drops of mixed (hold back) carrier, allow iron and other metal hydroxides to flocculate. Centrifuge and transfer the supernate to a clean centrifuge tube. Rinse the precipitate with 10 mL of D.I. water centrifuge and combine the supernates.
- Add 10 mL of freshly prepared buffer solution prepared by mixing equal amounts of 1.5M acetic acid and 3M ammonium acetate. Add 2-3 drops of methyl red indicator and adjust the pH to about 5.0, straw color of methyl red, using glacial acetic acid or conc. NH4OH. Heat for 5-10 min. then add 1 mL of barium carrier and 1 mL of 1.5M Na₂CrO₄ and continue heating for 5-10 min.

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6. Centrifuge and/or filter the BaCrO₄ through a 2" 600 funnel (using Whatman #42 filter paper) into a clean centrifuge tube. Rinse the filter with D.I water. Fill the centrifuge tube or tubes with saturated Na₂CO₃, centrifuge, and decant the supernate to waste.

KEY POINT: Centrifuge only if the BaCrO₄ precipitate appears to be heavy, otherwise filter.

For vegetation, soil and filter samples. Dissolve the precipitate in a minimum of 2M HNO₃ and repeat steps 4, 5, and 6.

7. Add fuming HNO3 dropwise to the precipitate until the effervescence has ceased, then add 50 mL of fuming HNO3 and cool in the ice water bath for 5-10 minutes. Centrifuge and decant the supernate to waste.

KEY POINT: Add the fuming nitric very slowly at first to avoid sample loss due to the effervescense.

8. Wash the sample one or more times using 20-30 mL of conc. HNO₃ depending on sample matrix. High calcium content matrices may require up to 3 nitric acid washes. Record the Sr-Y separation time on the counting request card.

KEY POINT: 1. If an unusually large precipitate

does not appear to diminish in size after two conc HNO₃ washes discontinue further washes and check with your supervisor.

- The time of the last nitric acid wash should be considered the separation time.
- 9. If sr-85 tracer was added, dissolve the sample precipitate in D.I. H₂O and transfer to 25 mL volume in a styrene jar or 20 mL volume in a plastic (scintillation) vial and submit for a gamma count for sr-85 recovery.
- If only Sr (cold) carrier was added, dissolve the precipitate in 10 mL of H₂O, add 25-50 mb of saturated Na₂CO₃, warm for 5-10 minutes, centrifuge, and discard the supernate. Wash the precipitate twice with 20-30 mL D.I. H₂O, centrifuge and discard the wash. Slurry the strontium carbonate precipitate with 1-2 mL of D.I. H₂O and transfer quantitatively to a tared 1" or 1 1/2" stainless steel planchet. Dry under infrared lamp. Cool, weigh, and record on the request card and beta count on either a Low Beta (plated on 1 1/2" planchet) or a Quad Low Background Beta proportional counter (plated on 1" or 1 1/2" planchet).

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11. Allow up to 14 days to permit Y-90 to grow into equilibrum with the Sr-90 (the ingrowth time may be shortened if necessary provided a growth correction is made).

STRONTIUM-90 (BY YTTRIUM-90)

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- 12. Using 2M HNO₃, carefully dissolve and rinse the strontium carbonate precipitate from the planchet into a 50-mL centrifuge tube. If Sr-85 yield determination was performed, transfer from the Styrene jar or plastic and to a 100-mL tube. Transfer one Yttrium carrier vial containing exactly 1 mL of Yttrium carrier (20.0 mg Y/mL). Rinse the vial with D.I. water, stir and let stand for 10 minutes.
- 13. Add 12M NaOH until a permanent clear gelatinous precipitate forms, then add 2 to 3 mL excess and decant supernate to a Styrene jar for possible re-milk. Stir, centrifuge for 5 minutes. Note the Yttrium separation date and time on the analysis request card.
- 14. Dissolve the precipitate with (1-2 mL) of conc. HNO₃, dilute to approximately 25-30 mL with water, then add conc. NH₄0H (approximately 15 mL) until a permanent precipitate forms. Add 2 to 3 mL excess NH₄0H, stir, centrifuge for 5 minutes and decant the supernate to waste.
- 15. Repeat Step \$14 then wash the precipitate with 25 mL of water, centrifuge for 5 minutes and decant the supernate to waste.
- 16. Dissolve the precipitate using 1 mL of conc. HCl, then add 40 mL of boiling hot 15% oxalic acid solution and let stand for 5 to 10 minutes. Centrifuge for 5 minutes and decant and discard the supernate. Wash the oxalate precipitate twice with 10 mL of water, centrifuge for 5 minutes and decant and discard the supernate.
- 17. Transfer the precipitate with water to a previously flamed, labeled, and tared 1" or 1 1/2" stainless steel planchet (depending on the beta counter to be used) with a plastic transfer pipet and dry under a heat lamp. Ignite slowly over a Meker burner to convert the oxalate to the oxide. Cool, weigh and record weight on the request card, fix the sample to the planchet with collodion (2 or 3 drops), and submit for a beta proportional count.

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1-01-5	4/10/87	1,1,1-7	<	1.00098+01	Н
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1-44-5	4/10/87	ALUMBIA	<	1.5000E+02	н
1-111-5	4/10/87	UIDOMNA	<	5.00016+01	ii
1 = 11 = 5	4/19/87	AHTIO IY	<	1,00002+02	H
1-44-5	4/10/87	ARSEUIC	<	5.0003E+00	н
1=44=3	4/10/87	BARIUT		5.070VE+01	H
111-5	1/10/87	BERYLAH	<	5.000JE+00	H
1-11-1-5	4/10/07	8ETA		8.7100E+00	H
1-111-3	4/10/87	CADITUM	<	2.000)E+00	Н
1-44-3	4/10/07	CALCIUM		4.0300E+04	H
1-44-9	4/10/87	CHLFORM	•	2.0003E+01	H
1-44-5 1-44-6	1/10/87	CHLORID		6.5300E+03	H H
1=44=5	4/10/87	CHRON IN	٠.	1.7700E+02 -1.6004E+00	8
1-41-5	1/10/87	C.) 40 C IL I Filh	₹'	2.2000E+00) H
1-41-5	4/19/87 4/10/87	COMBUCT	•	4.1400E+02	;; ii
1-44-5	4/10/87	COPPER	<	1.000002+01	н
1-114-5	4/19/87	C8-137	ζ.	3.6600E+00	i.
1-111-5	4/10/67	FAL PITH	3	1.5000E+02	ü
1-4-1-5	1/10/87	FAUT 1 III	3	1_0000E+02	ii
1-111-5	4/10/87	FARGE II	ξ.	5.0000E+00	H
1-11-1-5	4/10/87	FBARTIN		5.5000E+01	Н
1-44-5	4/10/87	FIDENYLL	<	5.000UE+00	łı
1-114-5	4/10/07	FCADILI	<	2.0001E+00	H
1-114-5	1/10/87	FCALCIU		5.7000E+01	11
1-11-5	4/19/07	FCHIO II		1.6000E+02	H
1-04-5	4/10/87	FCOPPER	4	1.0000E+01	H
1-111-5	4/10/87	FIRIN	4	5.0000E+01	11
1-111-5	4/19/47	FLEAD	<	5.0009E+00	H
1-44-5	4/10/07	FLUIRID	<	5.0000£+02	
1-111-5	4/10/87	FIAGRES		1.0300E+04	
1-114-5	1/10/87	F IA IGAN	<	5.00016+00	
ق ماراره 1	4/10/87	PIERCUR	<	1,0009E-01	H
1-111-5	4/10/47	FIICKEL	\$	1.0000E+01	H H
1=41=5	4/10/87	FUSAL 14	4	3.0009E+02 4.6004E+03	
1-19-5	4/10/87	FPOTASS	<	5.000UE+00	
1-44-5	4/10/87 3/10/87	FOCLETI FOILVER	3	1.0000E+01	H
1-114-3 1-11-5	4/10/87 1/10/87	F303111	•	1.0300E+04	
1-11-5	1/10/07		\$	3.9 (090-02	
* = 11 1 = 3	47 1 17 34	1.00000		. Jan 1975. 196	.н.

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. 1-41-5	1/13/8/	F /A IADI		5.000 1E400 \$	il ş	9	9	7	***	0	~	~
1-41-5	4/10/97	FZI-IC		4.20006401	НĒ	-	¥		ş	Ç.	**	. 19
1-41-5	4/10/87	100-1		5.00000+01	H							
1-41-5	4/10/87	LEADGE	<	5.0000E+00	н							
1-1:1-5	4/10/87	LOALPHA		2.3400E+00	н							
1=84=5	1/10/87	H-XYLE	<	1.0000E+01	- II				•			
114-5	1/10/87	HAGHET	•	1.0400E+04	н							
1-01-5	1/19/97	HAMBERE	<	5.0JB0E+00	H							
1-01-5	1/10/87	HERCHIY	<	1.000JE-01	H							
1-01-5	4/10/87	IET-10 IE	<	1.00008+01	H							
1-111-5	1/10/47	HETHYCH	<	1.0000E+01	H							
1-44-5	1/10/87	HICKEL	4	1,0909E+01	Ħ							
1-01-5	1/10/87	HITHATE		2.8709E+04	H							
1-11-5	4/10/87	OPXYLE	<	1.0900E+01	म							
1-41-5	4/10/87	034104	<	3.0000E+02	d							
144145	4/13/87	PERÇE IE	<	1.0000E+01	H							
1-111-3	1/13/87	Pil		7.370JE+00	H							
1-44-5	4/10/87	P. 103P 1A	<	1.000JE+03	H							
1-11-3	4/10/87	POTASUM		4.5500E+43	H							
1-11-5	1/10/87	RADIUI	<	3.4709E-02	H							
1-01-5	1/10/87	(f:) 10a	<	-1.9300E+01	В							
1-11-5	1/10/07	SELEDIA	<	5.000UE+00	i							
1 -4-1-1	1/1/1/87	SILVER	<	1.0000E+01	H							
1-:11-5	1/10/67	2-10 Liv-i		1.1000E+04	H		_	^		90.		
1-11-1-3	4/10/87	50 90	<u> </u>	7,53005-01		BK	9	for	-	905,	•	
عهطناتل	41774	مرابا ليامم	_ئے	3.0000E+03 5.736 E+04	~						-	
deller.	1/10/37	SILFATE	•		. т. П							
1-01-3	4/10/87	TETIM IE	<	1.00008+01	H							
1-114-5	4/19/87	ŢġÇ	•	0.6400E+02	н Н							
1-114-5	4/10/97	TOX	< <	1.0900E+02	H							
1-114-5	4/10/07 1/10/07	Taice ie Taition	•	1.0000E+01 1.0700E+03	ä							
1-114-5		11077.174		2.07002+00	B							
1-111-5	1/10/87	•		5,9000E+00	H							
1=111=5 1=111=5	9/10/87 9/10/87	AVAD 14		4_840UE+01	H							
fwil (a.)	47.10/41	ZINC		4403005401	••							

-	HELL HAME	COLLECTION	CONSTITUE IT	LESS THA I Flag	AHALYSIS VALUE	ANALYSIS SPONSOR
	6-77-54	6/12/87	1,1,1-T	<	1.0900E+01	н
	6-77-54	6/12/97	T-5,1,1	<	1.0000E+01	H
	6-77-54	6/12/47	1,1-014	<	3.00008+03	H
,	6-77-54	6/12/07	NIG-5,1	<	3.000)E+03	H
	6-77-51	6/12/87	12- Iban	<	1,0000E+01	н
	6-77-54	u/12/87	1234TE	4	1.0000E+01	Ħ
	6-77-54	6/12/47	153510	<	1.00001E+01	н
	6-77-34	u/12/87	123TRI	<	1.000008+01	н
	6-77-54	6/12/87	13- Iban	<	1.0000E+01	н
	6-77-54	6/12/87	135TRI	<	1.0000000101	H
	6-77-54	ú/12/87	14= ibun	<	1.0000E+01	Н
	6-77-54	6/12/07	ACETILE	<	3,0000E+03	н
Ų.	6-77-54	6/12/47	ACRYIJE	<	3.0000E+03	н
	6-77-34	6/12/07	ALKALIN		1.100E+05	H
\boldsymbol{z}	6-17-54	112/87ئ	ALLYLAL	< −	3.0000E+03	Ħ
Ref.	6-77-94	14\21\ن	UINCHEA	<	5.0000E+01	н
. "	6-17-54	6/12/87	BETA		0.5100E+00	Ħ
	6-77-54	6/12/87	CHLACET	<	3.0000E+03	H
.~	6-77-j4	6/12/87	C ILFO-UI	4	1.040 16+01	н
W	6-77-34	6/12/07	CHLARAL	<	3.0001E+03	H

•	6-77-3-	J/12/87	CHETRID		5.7109E+03	11		
	-		=	4	-			
	6-77-14	J/12/07	CHERRIP	<	3.03098+03	H		
_	6-77-54	0/12/87	соноист		2.77036+02	Н		
	6-77-54	U/15/87	CAY 10:40	<	3.0000E+03	H		
	6-77-54	6/12/87	CAYAC IT	<	3,0049E+03	H		
_	6-77-54	0/12/07	CAYJOCH	<	3.0000E+03	н		
	6-77-51	5/12/07	DICPR)P	<	3.000 IE+03	н		
	6-77-54	6/12/87	ETHCARD	4	3,00076+03	н		
<u> </u>	6-77-54	6/12/87	ETHCY40	<	3.000026103	11		
_	6-77-51	0/12/87	CIXI HT3	<	3.0000E+03	H	•	
	6-77-54	4/12/87	FALOMIN	<	1.50008+02	H		
	6-77-54	6/12/87	FAIRTE 10	<	1.09008+02	Ħ		
-	6-77-14	4/12/87	FARSE II	<	5.0300E+03	ží		
	6-77-54	6/12/87	PHART JIS		2.49008+01	21		
	6-77-34	0/12/87	FREHYLL	<	5.0000E+03	н		
•	6-77-51	6/12/67	FCAUTIN	<	2.00036+00	н		
	6-77-34	6/12/07	FCALCIU	•	3.5000E+04	н		_
	6-77-34	0/12/67	FC-130 II	<	1.000JE+01	н	AKG For C	hours.
٠.			FCOPPER	``	1.0701E+01	Н		THE WATER
	6-77-31	6/12/07			•			
	6-77-34	6/12/87	FIRE	۲	5.0)00E+01	ri		
•	6-77-54	4/12/87	FLUIRIA	_	5.0400E+02	н		
	6-77-54	6/12/47	FLU:III/IA	4	3.0000E+03	H		
	4-77-00	. /12/07	FHAGHES		1.1600E+04	н		
~	6-77-34	0/12/07			-			
	6-77-54	6/12/87	FIRALGAN	<	5.0000E+00	H		
	6-77-54	6/12/87	FIERC JR	<	1.0000E-01	H		
\	6-77-57	0/12/07	FHICKEL	<	1.0000E+01	н		
	6-77-54	u/12/97	FOS II JU	<	3.0000E+02	H		
	6-77-51	6/12/87	FPOTAGE		4.01006+03	Н		
1	6-77-54	6/12/87	FSELETI	<	5.0703E+00	H		
	6-77-54	0/12/07	FOILVER	<	1.0000E+01	н		
	6-77-57	u/12/87	F80.)I 18		1.5400E+04	н		
****	6-77-jl	o/12/87	FSTRO IT	< −	3.0000E+02	н		
	6-11-51	6/12/47	FYA IA'II		2.7000E+01	н		
	6-77-54	6/12/87	FZI iC		1.0900E+01	Н		
	6-77-34	0/12/87	GLYCIDY	<	3.0000E+03	H		
•	6-77-54	0/12/87	HEXACIL	<	1.0009E+01	H		
	6-77-54	0/12/87	HEXCHEN	<	1.00076+01	11		
	6-77-51	0/12/47	HYDRAZI	<	3.0000E+03	H		
	6-77-54	6/12/87	THORUTY	<	3.09096+03	H		
	6-77-54	u/12/87	KERASEA	έ.	1.0000E+04	H		
	6-77-54	6/12/87	LIALINIA		1.0900E+00	H		
*-	6-77-54	0/12/97	H-XYLE	4	1.000002+01	н		
	6-77-54	0/12/07	HET IN IE	₹	1.0000E+01	11		
	6-77-54	6/12/87	HET-IYC-I	3	1.0900E+01	н		
~	6-77-54	6/12/47	IETZI IE	4	3.0000E+03	11		
		6/12/87	HAPIJI IA	3	1.0000E+01	н		
	6-77-54			•	7.9000E+03	H		
$\overline{}$	6-77-54	6/12/87	HITRATE		1.0000E+01	Ä		
	6r77-54	u/12/87	OLXAFE	<		н		
	br77-54	0/12/87	PARALIE	<	3.0000E+03			
$\overline{}$	61-77-51	4/12/07	PENTCHA	<	1.0000E+01	H		
	6-77-54	6/12/07	PERCE IE	<	1.000UE+01	н		
	6-77-54	6/12/87	P.I		7.3000E+00	11		
	1-77-54	i/12/87	P.1		7.4400E±00	Н		
	1-77-54	6/12/07	PHE IOL	<	1.0000E+01	H		
	k-77-54	6/12/87	PHOSP-IA	<	1.0000E+03	11		
	6-77-54	o/12/87	PROPYLA	<	3.07008+03	н		
~	6-77-54	u/12/87	իժառ ա	<	3.0909E+03	Ħ		
	6-77-54	6/12/07	SILFATE		4.1500E+04	H		
\ _ሞ	6-77-54	u/12/07	TETRAILE	<	1.0000E+01	Н		
' μ	6-77-54	0/12/87	TETHCHE	₹	1.0900E+01	Ħ		
•"	6-17-54	5/12/47	TIC	4	4.4500E+02	H	·	
٠,٧٧		3/12/97	T 141,01.	ς.	1.63008+43	d		
•	6-77-54	J/12/97	THICE IE	<	1.09008+91	4		
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REFERENCE 3

Hanford Wells, PNL-5397, February 1985

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	WELL DESIGNATION EHA NO.	COORDINATES	CASING RLEV. IFT-MEL)	DEPTH	INIT. DEPTH (FT) TO WATER	DIA. (IN)	DEPTH TO BOTTOM (FT)	MIN-HAX PERFORATED DEPTH (FT)	DATE COMP. (M-Y)	PORMER DESIGNATION	СОММЕНТЯ
	199 H3 1	N 894994 P H 848852	421.98	75	45	8.8	75	29 - 74	860		SAMPLE PUMP
	199 H4 1 1875	N 895768 P W 838488	417.75	75		6.8	38	25 - 50	3-52	197-H-1	CASING REMOVED
	199 H4 2 1876	N 095200 P W 038565	428.43	386	3	6.8	311	25 - 59	5-52	167-B-2	CONFINED AQUIFER
	199 H4 3 1877	N 896373 P W 83988	420.35	55	39	6.0	55	34 - 55	5-74		SAMPLE PUMP
	199 H4 4	P # 038685	413.76	55	36	6.0	59		6-83		SCREEN 33-43 GROUTED SAMPLE PUMP
19	199 114 5	> N 096639 N 039065	416.26	68	36	6.8			5-83		SCREEN 32-42 GROUTED SAMPLE PUMP
	199 H4 6 1874	N #96473 P H 848245	419.58	55	41	6.0	54		5-83		SCREEN 39-49 GROUTED SAMPLE PUMP

	WELL DESIGNATI EMA NO		COORDINATES	Casing Elev. (FT-HSL)	DEPTH	INIT. DEPTH (FT) TO WATER	DIA. (IN)	DEPTH TO BOTTOH (PT)	MIN-MAX PERFORATED DEPTH (FT)	DATE COMP. (H-Y)	FORMER DESIGNATION	Сомнента
	199 N 1908	1	N 886157 P W 868593	456.19	108	54	8.8	62	34 - 95	5-64		
		C)	456.37	68	50	1.59		48 - 68	19-64		
		E	•	456.38	98	55	1.59		88 - 98	10-64		
		(}	456.36	74	55	1.50		72 - 74	10-64		
	199 -N 1984	2	N 066577 P W 060306	459.83	125	57	8.9	95	35-120	6-64		PLUG AT 95
		C)	468.18	76	57	1.50		68 - 78	18-64		
		I	•	469.10	122	57	1.50	132	112-122	18-64		
		•)	460.10	197	61	1.59		105-107	18-64		
			ì	460.18	79		1.50		77 - 79	10-64		
23	199 N 1896	3	N #86365 P H #6#828	459.45	125	63	8.8	70	34 - 95	6-64		SAMPLE PUMP
		C		459.72	68		1.59		68 - 68	19-64		REHOVED
		F	•	459.71	122		1.5#		112-122	18-64		REHOVED
		C)	459.72	. 98		1.50		96 - 98	10-64		REHOVED
		F	l	459.71	88		1.50		78 - 88	19-64		REHOVED
	199 N		N 085921 P W 060042	458.73	159	52	8.9	73	50 - 68	6-64		CEMENT BRIDGE 673 FT SAMPLE PUMP
		0	•	457.90	63	54	1.50		50 - 60	4-64		REHOVED 6-74
		F	•	457.98	138		1.50		128-138	19-64		REHOVED 6-74
		0)	457.98	123		1.50		113-123	16-64		REHOVED 6-74
		F	l	457.90	185		1.50		193-195	18-64		REHOVED 6-74
		8	i	457.90	78		1.50		76 - 78	18-64		REHOVED 6-74

DESIGNATION COORDINATES	Casing Elev. (FT-HSL)	DRILL Depth	(FT) TO	DIA. (IN)	DEPTH TO BOTTOM (FT)	MIN-MAX PERFORATED DEPTH (FT)	DATE COMP. (H-Y)	FORMER DESIGNATION	COMMENTS
N 874598	379.07	ر د		6.0	35		5~43	HR-11	PILLED IN WITH SILT
N 874698 653 73 218 P W 823378	380.88	36		6.8	36	•	5-43	ur-12	FILLED IN WITH SILT
659 76 34	375.01	19	2	4.0	19		0- 0	REF.2 NO.79	BITTED IN
655 74 38	414.89	41	3	16.0			9- 8	13/25-3D1 REP.7	DUG WELL
699 77 14 8 H 834275	397.24	21	7	12.5	21		0- 0	T14H R27E 32-Q1	FILLED IN
M 876788 699 77 36 8 4588 M 836158	412.28	159	35	6.9	82	32 - 82	4-57		CEMENT PLUG AT 82FT SAMPLE PUMP
№ 876688 699 77 43 Р Н 842588	441.37	44	7	72.8	44		9- 6	REF.2 NO.74 S1618	FILLED IN .
699 77 54 P 4512 H 854188	488.59	159	84	8.8	116	78 -128	5-57		CEMENT PLUG AT 120FT SAMPLE PUMP
699 78 36 ·4489	465.00	38	•	48.9			e- e	REP.2 NO.28 N.RUN NO.4	PILLED IN FARM WELL
N 878859 699 78 45 P H 844625	432.98	36	(69.9			8- 0	REF.2 NO.111	PILLED IN
N Ø7775Ø 699 78 62 P 4511 W Ø6230Ø	469.88	159	76	8.8	189	79 -128	5-57		#15 SCREEN 67-187 PT SAMPLE PUMP
699 79 184	774.08	699	378	16.0	783	429-678	2-53	PSN 515 14/25-31M1	ARHY CAMP WELL

大學者不可以有其可不然 不得時見一點接近在一個接有 以不為軍軍各人軍人以一下一人有關不關問題時本監

Ref. 3.3

EMA NO.	- c	COORDINATES	CASING ELEV. (FT-HSL)	DRILL E DEPTH ((FT) h	FT) T	O DIA. (IN)	DEPTH TO BOTTOM (FT)	HIN-HAX PERFORATED DEPTH (FT)	DATE COMP. (H-Y)	FORMER DESIGNATION	соннентв
<u>199 вз</u> 1851	D	N 871888 P W 879838	439.79	63	46	8.8	63	28 - 68	3-53	187-B-1	SAMPLE PUMP
199 B3 1852	2	N 071752 P W 070818	442.59	799	26	8.5	768	635-645	8-53	167-B-2	
1856	P		443.36	778		1.58		758-778	3-70		
1857	Q		442.81	642 `	42	1.50	429	632-642	3-76		
199 B4 1853	1	N 878889 P W 888659	461.88	98	65	8.6	83	58 - 98	2-49	166-8-1	SAMPLE PUMP
199 B4 1854	2	N 869933 P W 888672	461.99	9#	63	6.5	86	62' - 98	2-49	188-8-2	PLUG AT 86
199 B4 1855	3	N 069933 P W 080636	461.75	91	63	8.9	86	68 - 98	2-49	188-B-3	PLUG AT 86
199 B4 1891	4	N 868978 P W 888367	472.14	185	76	8.6	96	49 -182	9-69		SAMPLE PUMP
199 B5 1895	1	N 869934 P W 882888	455.56	151	58	8.9	180	40 -140	8-62		PLUG AT 188
199 B8	1	N 967375 P W 980490	494.59	75		8.5	76	нонв .	3-51	C431-1	NOT ACCESSIBLE
199 BB	2	N 967499 P W 989896	495.59	25		8.6	26	ноне	4-51	C431-2	CASING REMOVED
199 BB	3	N 867455 P W 881118	488.59	25		8.9	26	None	4-51	C431-3	CASING REMOVED
199 BB	4	N 967519 P W 881428	488.59	25		8.6	26	none	4-51	C431-4	CASING REMOVED

	WEILL DESIGNATION EHA NO.			CASING ELEV. (PT-HSL)	DRILL	(FT) 7	TO DIA.	DEPTH TO BOTTOM (FT)	HIN-HAX PERFORATED DEPTH (FT)	DATE COMP. (H-Y)	PORMER DESIGNATION	COMMENTS	
	199	D5	7	N #9352# P W #5214#	465.57	35		8.5	36		3-49	187-DR-4	CASING REHOVED
	199	D5	8	N 893528 P W 852090	465.57	31		8.6	32		3-49	187-DR-5	CASING REMOVED
	199	D5	9	N 893528 P W 851998	462.57	35		8.9	36		3-49	197-DR-6	CASING REHOVED
	199	D 5	10		464.57	28	DRY	8.8	29	14 - 27	4-57	199-DR-1	
	199	D5	11		464.57	27	ĐRY	4.8	28	14 - 27	4-57	198-DR-2	
<u></u>	189	D5 2	12	N 092125 P W 052546	469.67	91	85	8.8	91	35 - 98	8-66		SAMPLE PUMP
	199	D7	1		429.88	51		48.9	51		0- 0	T14N R26E 14-F1	PILLED IN
	199	D7	2		420.88	39		84.9	192		0- 0	T14N R26E 14-P2	FILLED IN
	199	DB	1	N #9523# P H #5314#	488.88	37		6.8	37		5-43	HR-9	BATTED IN
	199 1861		2	N 894725 P W 853818	444.61	77	DRY	0.9	44	38 - 75	6-52	187-D-1	
	199 1867		3	N #9472# P W #522#5	449.06	81	62	6.0	81	35 - 79	6-52	167-D-2	SAMPLE PUMP

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	199	K	16	H 976388 P W 867888	484.88	50		8.5	5.0		2-53	105-XE-1	
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	199 1879		18		489.66	68	21	8.8	48		16-54	187-KB-3	
	199 1884		19	N 878898 P W 867888	422.17	51	39	8.8	51	18 - 59	4-55	187-KE-4	SCREEN 26-46 PT. SAMPLE PUMP
	199	K	28	N 079500 P W 066125	422.57	50	31	8.8	48	10 - 59	5-55	107-KE-5	SAMPLE PUMP
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	199 1887		22	N 661600 P W 665000	421.68	59	30	8.8	49	10 - 50	5-55	197-KE-7	#15 SCREEN 29-49 FT. SAMPLE PUMP
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<u>Draft Phase I Installation Assessment of Inactive Waste-Disposal</u>

<u>Sites at Hanford</u>, Volume 1, July 1986

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2.0 DESCRIPTION OF HANFORD SITE

This section provides a summary of environmental conditions at the Hanford Site and a brief discussion of the Site's purpose and history. It also describes specific environmental features and the the process history of each operational area (i.e., the 100, 200, 300, 400, and 600 Areas).

2.1 ENVIRONMENTAL SUMMARY

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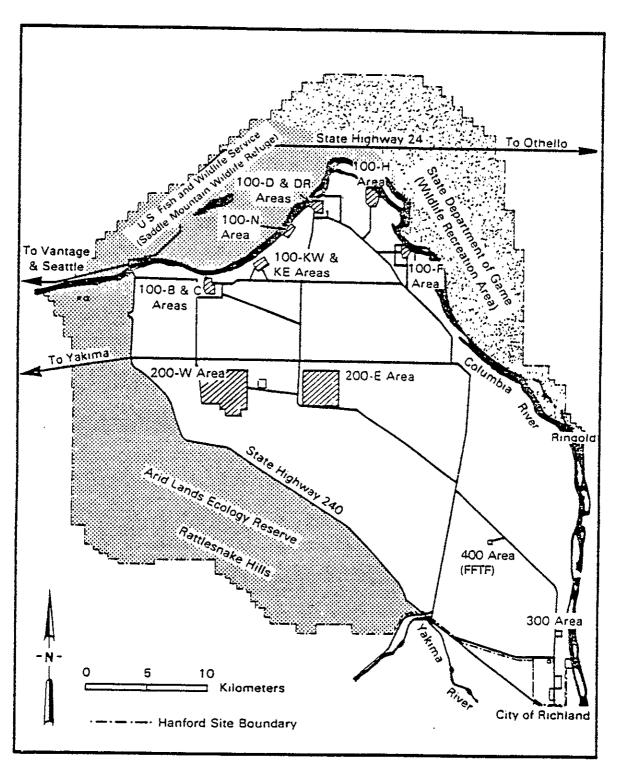
The semiarid Hanford Site, operated by the DOE, occupies about 1,476 square kilometers (570 sq mi) of the southeastern part of Washington State north of where the Yakima River flows into the Columbia (see Figure 2.1). The Site lies about 320 kilometers (200 mi) east of Portland, Oregon, 270 kilometers (170 mi) southeast of Seattle, Washington, and 200 kilometers (125 mi) southwest of Spokane, Washington.

Environmental conditions common to all areas at Hanford Site are summarized below. Descriptions of these environmental aspects are based on several recent reports (U.S. DOE 1984; Sommer et al. 1981; Yandon 1977; U.S. ERDA 1975).

2.1.1 Geology and Soils

The Hanford Site lies in the Pasco Basin, a structural and topographic basin of eastern Washington and the Columbia River Basalt Plateau. The region is underlain by three geologic units. In ascending order these are: 1) the sequential beds of basaltic lavas and interbed sediments of the Columbia River Basalt Group at the base; 2) the Pliocene-age Ringold Formation (lacustrine formation), consisting of well-rounded pebbles and cobbles with interstitial spaces filled with medium sand; and 3) the Hanford Formation, consisting of the Pasco (glaciofluvial) gravels and associated sediments of the late Pleistocene age lying at the surface.

The surface geology of the Site is characterized by a surface layer of light brown, fine, slightly silty, wind-deposited sand, sparsely covered by vegetation. Although the surface soil is fertile, it has little agricultural value without irrigation. Underlying the surface sands is a mixture of sand



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FIGURE 2.1. Features of the Hanford Site.

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and gravel extending to a depth of about 60 meters (200 ft). Basaltic rock starts at that depth and extends downward over 3000 meters (1.9 mi).

Elevations range from a low of about 105 meters (345 ft) above mean sea level (MSL) in the southeastern part of the Hanford Site to a maximum of 1,091 meters (3,579 ft) at the crest of Rattlesnake Mountain to the west. (See Section 2.3 for a discussion of geologic features peculiar to each operational area.)

2.1.2 Meteorology

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The Site lies east of the Cascade Mountains and, as a result, has a semiarid climate reflecting the rainshadow effect of the mountains. The average annual precipitation for the Site is about 160 millimeters (6.3 in.). Ten percent of this amount falls from July through September, and 42% falls from November through January. The greatest amount of rainfall recorded in a 12-hour period was 47.8 millimeters (1.9 in.).

Because of the limited rainfall, surface runoff from the Hanford Site is minimal. The annual precipitation mostly evaporates, resulting in small amounts of water available for runoff or infiltration.

2.1.3 Hydrology and Hydrogeology

The Columbia River (the fifth largest river by volume in North America) is the dominant aquatic ecosystem on the Hanford Site. Numerous dams have been built on the river. The only free-flowing section in the United States is between Priest Rapids Dam and McNary Reservoir, along the Hanford Site. No significant tributaries enter the stream in this section.

The Columbia has a long-term annual average flow of about 3,600 cubic meters per second (127,000 cfs). [The Yakima River, by comparison, flows an average of about 90 cubic meters per second (3,180 cfs).] The flow rates of the Columbia are influenced by water usage and upstream reservoir projects. The reservoirs provide active storage of more than 4.6×10^{10} cubic meters (37,000,000 acre-feet) of water.

The uppermost aquifer in the Pasco Basin is an unconfined system within the Hanford and Ringold Formations. The elevation of this aquifer ranges from

excluded from public use by the DOE and are used as a wildlife refuge and for DOE environmental research. The land north of the Columbia River is controlled by the Washington State Department of Game and the U.S. Fish and Wildlife Service as controlled hunting areas and a game refuge.

Land use within a 50-kilometer (30-mi) radius of the Site includes residential, suburban, corporate city, agricultural, industrial and commercial, scenic, recreational, and general use areas. The predominant use of lands within the 50-kilometer (30-mi) radius is agricultural, with farms located along or near all the Site boundaries.

2.1.6 Population

Population in the area surrounding the Hanford Site is sparse, consisting primarily of farms and farming communities to the north, east, and west of the Site. The Tri-Cities (Kennewick, Pasco, and Richland), located to the south and southeast of the Hanford Site, represent the major population concentration in the area (Sommer et al. 1981).

In 1980, an estimated 341,000 people were living within an 80-kilometer (50-mi) radius of the Hanford Meterological Station (HMS) (see Figure 2.4); it is estimated that the number will grow to 417,000 by 1990.

2.1.7 Air Quality

Air quality in the vicinity of the Hanford Site is generally quite good. Wind-eroded dust is a problem in the area, and the dust storms that occur in the region can produce high total-suspended particulate concentrations. However, on both an annual and a short-term basis, the region is in compliance with the National Ambient Air Quality Standards (NAAQS) for particulates. All other pollutant levels also satisfy the federal and State of Washington standards (U.S. DOE 1984).

2.2 PURPOSE AND HISTORY

In 1943, after the Fermi experiment showed that nuclear fission could be controlled in a small reactor, the U.S. Army Corps of Engineers selected "anford as the location to build larger versions of the Fermi reactor to produce plutonium for possible use in military weapons. Construction started

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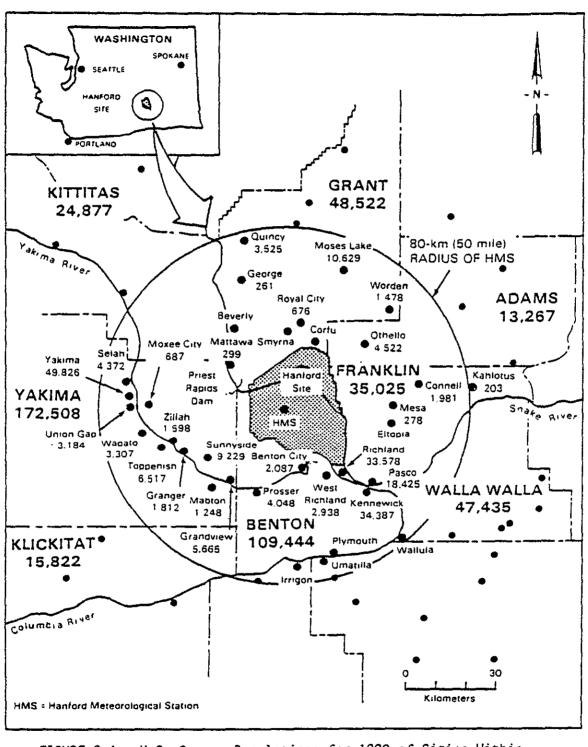
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FIGURE 2.4. U.S. Census Populations for 1980 of Cities Within 80 Kilometers of the Hanford Meteorological Station (U.S. DOE 1986)

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in March 1943 on three reactor facilities and three chemical processing facilities. The first of the reactors went into operation about 18 months after the start of construction, and the first plutonium was available some 4 months later.

After World War II, five reactors similar to those built during the war were constructed. A total of eight graphite-moderated reactors used the Columbia River for once-through cooling (i.e., water circulated through the reactors only once before being released back to the river).

Early in the 1950s construction began on the research and development facilities known as the Hanford Laboratories. This marked the first diversification of Hanford from a purely defense-materials production facility to one heavily involved in peacetime uses of the atom.

In 1963 the N Reactor was built. The N Reactor is different from the other eight reactors in that it generates steam as a by-product of the plutonium production and does not use river water as a once-through coolant. Since 1966 the Washington Public Power Supply System has used the steam to generate electricity.

A presidential decision was made in early 1964 to begin shutting down the older Hanford reactors. This decision resulted in the closing down of all eight of the older reactors by the end of 1971, leaving the N Reactor as the only operational production reactor.

2.3 DESCRIPTION OF OPERATIONAL AREAS

Environmental features specific to each operational area are described below; the waste-processing history of each area is also discussed. Each area is identified by number (i.e., 100, 200, 300, 400, and 600) and by letter (e.g., the 100-F Area is the location of the 100-Area F Reactor). Appendix B provides further information on waste-disposal site locations and types of waste-processing facilities in Hanford's operational areas.

2.3.1 100 Areas

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The nine 100 Areas (8, C, D, DR, KE, KW, F, H, N) border the Columbia River in the northernmost part of the Hanford Site. Each of the nine areas

has one production reactor. Eight of these reactors have been shut down; only the N Reactor, used for both plutonium and electricity production, is still operating. Because some of the areas are contiguous (B/C, D/DR, KE/KW), the Hanford Site map shows only six 100 Areas (Figure 2.1).

The 100 Areas are generally flat with no major surface features. The Hanford Formation lies near the surface of the 100 Areas, covered by a thin layer of wind-deposited silt and fine sand. The water table is found in these sediments at a depth of about 20 meters (66 ft), except in the F and H Areas where the depth to the water table is about 35 meters (115 ft) and 40 meters (131 ft), respectively. The depth to the Ringold Formation is about 25 meters (82 ft); the top of the basalt bedrock is approximately 240 meters (790 ft) below the surface.

Because the water table occurs within the highly permeable sandy gravels of the Hanford Formation, it fluctuates as the river level rises and falls. The ground water generally flows from the 100 Areas and toward the river. When active, each of the 100 Areas included support facilities such as powerhouses. Except for 100-N, these powerhouses produced process steam from coal-fired boilers; 100-N has oil-fired boilers. Adjacent to each area's powerhouse were large storage areas that received railroad carloads of coal and disposal areas for flyash/clinker disposal. Most areas also included water-treatment plants, water-storage tanks, subsurface sewage-disposal systems, raw-water intake structures, and process sewers.

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<u>B and C Areas</u>. The 8 and C reactors are located adjacent to each other on a 2.6-square-kilometer (650-acre) site (the 100 B/C Area) and are the farthest upstream of the 100 Areas. The B Reactor was operated from 1944 to 1968, and the C Reactor was operated from 1952 to 1969. Virtually all the facilities in the area are inactive, with the exception of the B/C export water system, which continues to provide the raw water supply to the 200 Area and some 100 Areas. An electrical substation in the area taps power for the pumps providing the 200-Area water. Fewer than 100 people work in this area (Yandon 1977).

When the reactors were operational, cooling water was drawn from the river and treated with alum, sulfuric acid, and chlorine. Excess sulfuric

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acid was used to maintain the pH of the water within a desired range. To control oxidation of aluminum parts in the reactor, sodium dichromate was used to maintain an oxidation coating on aluminum parts. The chlorine was added for algae control in the settling basins; at times copper sulfate was added for additional algae control. Chromic acid, oxalic acid, and nitric acid were used for dummy fuel-element decontamination.

In addition to vertical safety rods for emergency reactor shutdown, the reactors were equipped with hoppers of nickel-plated boron steel balls, nickel-plated carbon steel balls, and stainless steel balls that would drop into the vertical safety rod channels for emergency shutdown. This system required no supplementary power source. Although it was never used, a third safety system, one involving the use of a potassium borate solution, was in place at the reactors.

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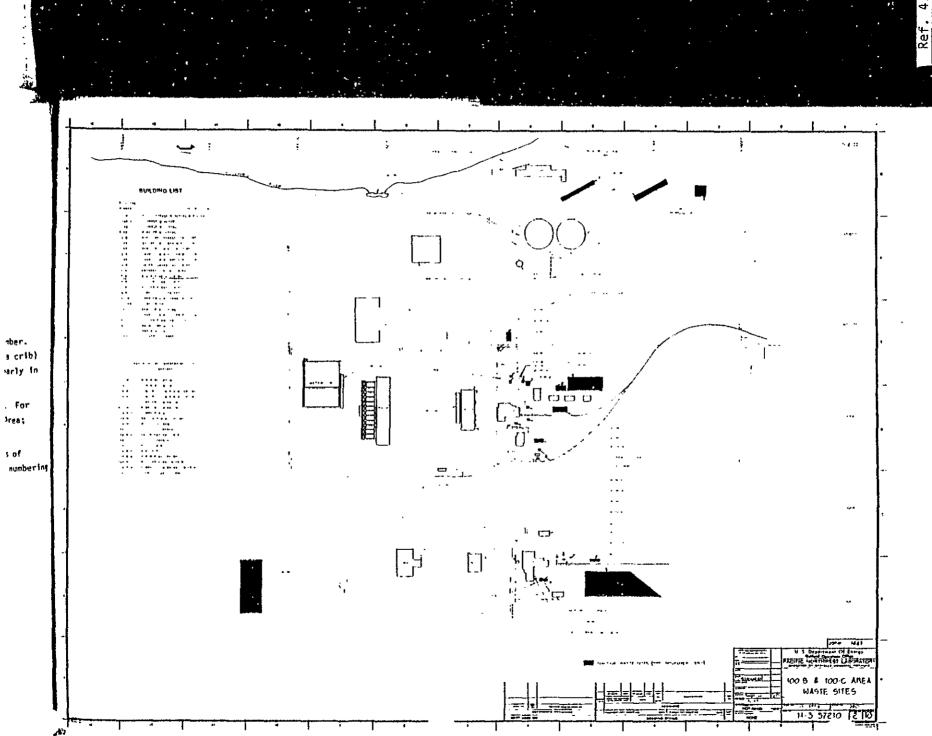
A supplementary control system, in addition to the normal horizontal control rods, was incorporated into the reactors. This supplementary control system consisted of a Poison^(a) Column Control Facility that could charge selected process tubes with a lead-cadmium poison to absorb neutrons. Boron-carbide aluminum poison splines were also used for supplementary control.

The coolant water system and backup control and shutdown systems at the other 7 once-through-cooled reactors were similar to the those in 100 B/C Area.

D and DR Areas. The 100-D/DR Areas, covering about 3.9 square kilometers (970 acres), are located 11 kilometers (7 mi) downriver of the 100-B/C Area. The D Reactor was operated from 1944 to 1967 and the DR Reactor from 1950 to 1965. These areas are extensively used, and their utilities and services are still in operation. The electrical substation serves as a backup supply for the 100-N Area. The water system is a backup system for the 100-B water system, which supplies water to the 200 Areas. The UNC Nuclear Industries engineering laboratory here is operated in support of the N Reactor.

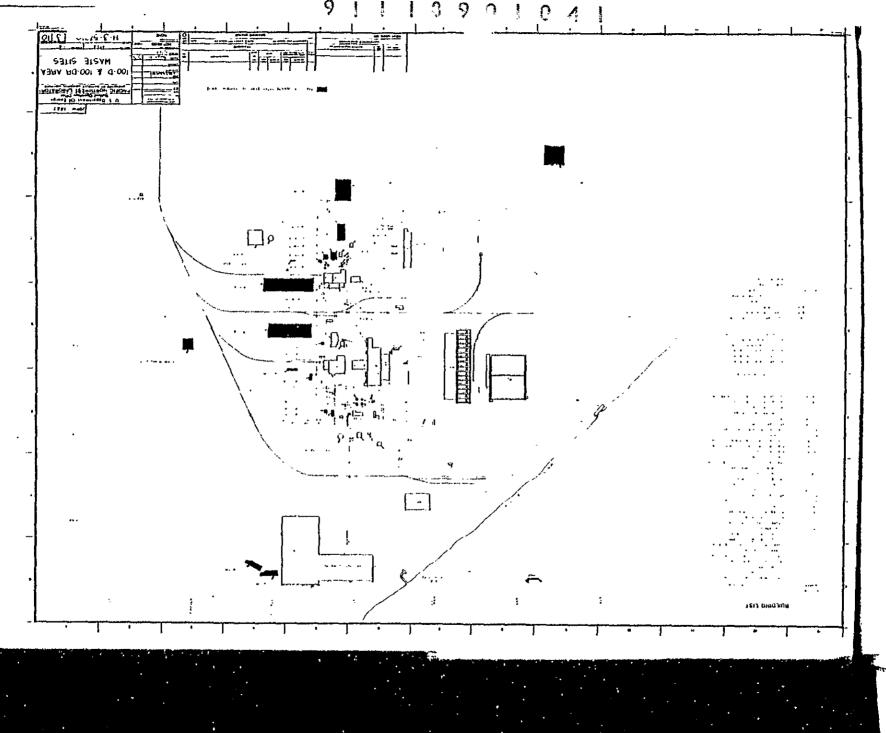
Approximately 20 people are employed in the D and DR Areas (Yandon 1977).

⁽⁴⁾ The term poison refers to a material's ability to absorb neutrons and thus control the rate of fission.



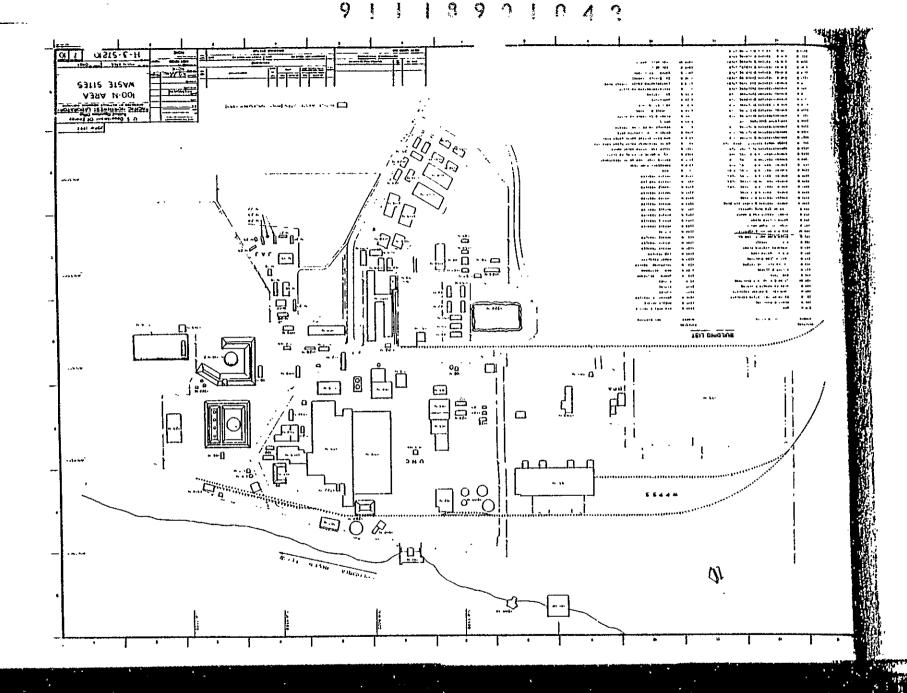
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Environmental Monitoring at Hanford for 1986, PNL-6120, May 1987

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2.0. BACKGROUND INFORMATION

2.1. DESCRIPTION OF THE HANFORD SITE

K. R. Price, P. J. Mitchell, and M. D. Freshley

The U.S. Department of Energy's Hanford Site is located in a rural region of south-eastern Washington and occupies an area of 1,500 km². The Site (shown in Figure 2.1) lies about 320 km northeast of Portland, Oregon, 270 km southeast of Seattle, Washington, and 200 km southwest of Spokane, Washington. The Columbia River flows through the northern edge of the Hanford Site and forms part of the eastern boundary. The southern boundary of the Site includes the Rattlesnake Hills, which exceed 1000 m in elevation. Both confined and unconfined aquifers are present beneath the Site. The main geologic units are the Columbia River Basait Group, the Ringold Formation, and a series of glaciofluvial sediments. The Hanford Project was established in 1943 and was originally designed, built, and operated to produce plutonium for nuclear weapons.

SURFACE CHARACTERISTICS OF THE SITE

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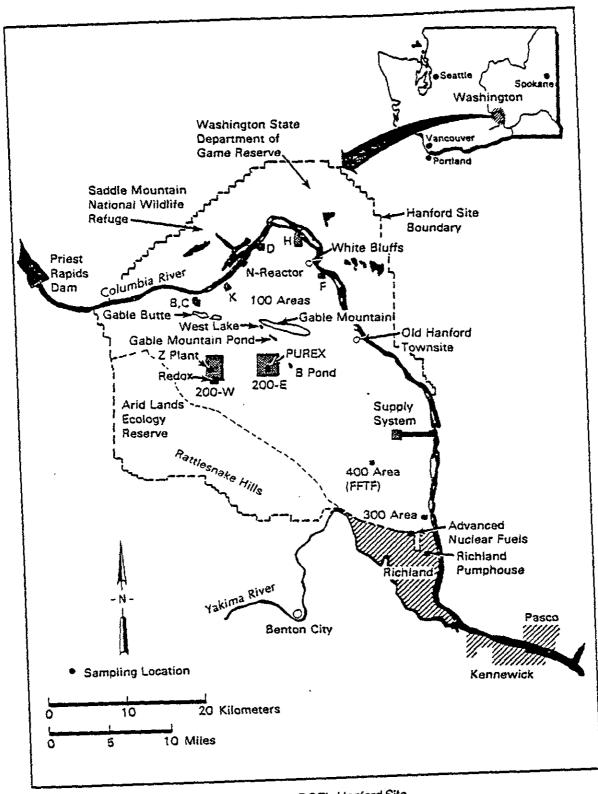
The semiarid land on which the Hanford Site is located has a sparse covering of desert shrubs and drought-resistant grasses. The most broadly distributed type of vegetation on the Site is the sagebrush/cheatgrass/bluegrass community. Most abundant of the mammals is the Great Basin pocket mouse. Of the big-game animals, the mule deer is the most abundant, while the cottontail rabbit is the most abundant of the small-game animals. Coyotes are also abundant. The bald eagle is a regular winter visitor to the relatively large areas of uninhabited land comprising the Hanford Site.

The Columbia River, which originates in the mountains of eastern British Columbia, Canada, flows through the northern edge of the Hanford Site and forms part of the Hanford Site's eastern The river drains a total area of approximately 70,800 km² enroute to the Pacific The flow of the Columbia River is regulated by 11 dams within the United States, 7 upstream and 4 downstream of the Site. Priest Rapids Dam is the nearest impoundment upstream of the Site, and McNary Dam is the nearest dam downstream. (The Hanford reach of the Columbia River extends from Priest Rapids Dam to the head of Lake Wallula, which is created by McNary Dam.) This is the only stretch of the Columbia River within the U.S. that is not impounded by a dam. The width of the river

varies from approximately 300 m to about 1000 m. The flow through this stretch of the river is relatively swift, with numerous bends and several islands present throughout the reach...

The flow rate of the Columbia River in this region is regulated primarily by Priest Rapids Dam. Hanford reach flows fluctuate significantly because of the relatively small storage capacity and operational practices of the nearby upstream dams. A minimum flow rate of 1,000 cubic meters per second (m³/s) (36,000 cubic feet per second (cfs)] has been established at Priest Rapids. Typical daily flows range from 1,000 m³/s (36,000 cfs) to 7,000 m³/s (250,000 cfs) with peak spring runoff flows of up to 12,600 m³/s (450,000 cfs) being recorded. annual average flows at Priest Rapids Dam are $3,100 \text{ m}^3/\text{s}$ (110,000 cfs) to $3,400 \text{ m}^3/\text{s}$ (120,000 cfs). Monthly mean flows typically peak from April through June and are at the lowest levels from September through October.

The temperature of the Columbia River varies seasonally. Minimum temperatures are observed during January and February while maximum temperatures typically occur during August and September. Monthly temperatures for the river range from approximately 3°C to about 20°C during the season a year. Water storage management practices at upstream dams and the flow rate of the river dictate, to a large extent, the thermal characteristics of the Columbia River along the Hanford reach.



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FIGURE 2.1. DOE's Hanford Site

The Columbia River system has been developed extensively for hydroelectric power, flood control, navigation, irrigation, and municipal and industrial water supplies. In addition, the Hanford reach is used for a variety of recreational activities including fishing, hunting, boating, water skiing, and swimming. The State of Washington has classified the stretch of the Columbia River from the Washington-Oregon border to Grand Coulee Dam (which includes the Hanford reach) as Class A and established water quality criteria and water use guidelines for this class designation. Because these criteria do not include specific limits for radionuclides, Environmental Protection Agency (EPA) and State of Washington drinking water limits were used for comparison. Other surface water on the Site consists of West Lake (a small, natural pond) and a number of ditches and artificial ponds created for routine disposal of waste water.

Hanford's climate is dry and mild; the area receives approximately 16 cm of precipitation annually. About 40% of the total precipitation occurs during November, December, and January; only 10% falls in July, August, and September. Approximately 45% of all precipitation from December through February is snow. The average minimum and maximum temperatures in July are 16°C and 32°C. For January, the average temperatures are 3°C and -6°C.

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Monthly average wind speeds range from about 10 km/h in the summer to 14 km/h in the winter. The prevailing regional winds are from the northwest, with occasional cold-air drainage into valleys and occurrences of strong crosswinds. The region is a typical desert area with frequent strong inversions that occur at night and break during the day, resulting in unstable and turbulent wind conditions.

Land near the Hanford Site is primarily used for agriculture and for livestock grazing. Agricultural lands are found north and east or the Columbia River and south of the Yakima River. These areas contain orchards, vineyards, and fields of alfalfa, wheat, and vegetables. The Hanford Site north of the Columbia River is shared between a state wildlife management area and a federal wildlife refuge. The northeast slope of the Rattle-snake Hills along the southwestern boundary of the Site is designated as the Arid Lands Ecology Reserve (ALE) and is used for ecological research by DOE.

The major population center nearest to the Hanford Site is the Tri-Cities area (Richland, Pasco, and Kennewick), which is situated on the Columbia River downstream from the Site and has a population of approximately 90,000. Approximately 340,000 people live within an 80-km radius of the Hanford Site. This number includes people living in the Tri-Cities, the Yakima area, several small communities, and the surrounding agricultural area. More detail on Site characteristics and activities is available in "The Final Environmental Statement, Waste Management Operations, Hanford Reservation" (ERDA 1975).

SUBSURFACE CHARACTERISTICS OF THE SITE

The DOE operations on the Site have resulted in the production of large volumes of waste water that have historically been discharged to the ground through cribs, ditches, and ponds. These discharges greatly influence the physics and chemistry of the subsurface. Approximately 25 billion liters of liquid effluent in the 200 Areas and 2.6 billion liters of liquid effluent in the 100N Area were disposed to the ground during 1986, including process cooling water and water containing low-level radioactive wastes. The discharge of waste water to the ground at the Hanford Site began in the mid-forties and reached a peak in 1955. After 1955, discharge to cribs declined because of improved treatment of waste streams and the deactivation of various facilities (Graham et al. 1981). Since the restart of the Plutonium and Uranium Extraction (PUREX) Plant and related facilities in late 1983, discharge of PUREX-related effluents has resumed.

Subsurface structures, such as cribs, have primarily been used for the disposal of water containing radioactive wastes, while surface ponds and ditches have primarily been used for the disposal of uncontaminated cooling water (Graham et al. 1981). Sanitary wastes are discharged to the ground via tile fields. The majority of liquid disposal occurred in the Separations Area, which includes the 200-East (200E) and 200-West (200W) Areas (Figure 2.1). Smaller amounts of waste water were disposed in the 100 and 300 Areas. Discharges of waste water to the ground in the 400 Area were minimal.

the Ringold formation has been removed. These sediments were deposited by the ancestral Columbia River when it was swollen by glacial meltwater. The glaciofluvial sediments consist primarily of gravels and sands, with some sitts (Newcomb, Strand and Frank 1972).

Hydrology

Both confined and unconfined aquifers are present beneath the Hanford Site. The confined aquifers, in which the ground water is under pressure greater than that of the atmosphere, are found primarily within the Columbia River basalts. In general, the unconfined or water-table aquifer is located in the Ringold Formation and glaciofluvial sediments, as well as some more recent alluvial sediments in areas adjacent to the Columbia River (Gephart et al. 1979). This relatively shallow aquifer has been affected by wastewater disposal at Hanford more than the confined aquifers (Graham et al. 1981). Therefore, the unconfined aquifer is the most thoroughly monitored aquifer beneath the Site.

The unconfined aquifer is bounded below by either the basait surface or, in places, the relatively impervious clays and silts of the lower unit of the Ringold Formation. Laterally, the unconfined aquifer is bounded by the anticlinal basalt ridges that ring the basin and by the Yakima and Columbia rivers. The basalt ridges above the water table have a low permeability and act as a barrier to lateral flow of the ground water (Gephart et al. 1979). The saturated thickness of the unconfined aquifer is greater than 61 m in some areas of the Hanford Site and pinches out along the flanks of the basalt anticlines. The depth from the ground surface to the water table ranges from less than 0.3 m near the Columbia River to over 106 m in the center of the Site. The elevation of the water table above mean sea level for June of 1986 is shown in Figure 2.3.

Recharge to the unconfined aquifer originates from several sources (Graham et al. 1981). Natural recharge occurs from precipitation at higher elevations and runoff from ephemeral streams to the west, such as Cold Creek and Dry Creek. The Yakima River recharges the unconfined aquifer as it flows along the southwest boundary of the Hanford Site. The Columbia River recharges the unconfined aquifer during high stages when river water is transferred to the aquifer along the river bank. The unconfined aquifer receives little, if any, recharge from pre-

cipitation directly on the Hanford Site because of a high rate of evapotranspiration under native soil and vegetation conditions. However, present studies, such as those described by Heller, Gee, and Meyers (1985), suggest that precipitation may contribute more recharge to the ground water than was originally thought.

Large scale artificial recharge occurs from offsite agricultural irrigation and liquid-waste disposal in the operating areas at Hanford. Recharge from imigation in the Cold Creek Valley enters the Hanford Site as ground-water flow across the western boundary. Artificial recharge from wastewater disposal at Hanford occurs principally in the It was estimated that re-Separations Area. charge to the ground water from facilities in the Separations Area (including B Pond and Gable Mountain Pond, as well as the various cribs and trenches in the 200W and 200E Areas) adds ten times as great an annual volume of water to the unconfined aquifer as is contributed by natural inflow to the area from precipitation and imigation waters to the west (Graham et al. 1981).

The operational discharge of water has created ground-water mounds near each of the major waste-water disposal facilities in the Separations Area and in the 100 and 300 Areas (Figure 2.3). These mounds have altered the local flow pattern in the aquifer, which is generally from the recharge areas in the west to the discharge areas (primarily the Columbia River) in the east. Water levels in the unconfined aquifer have changed continuously during Site operations because of variations in the volume of waste water discharged. Consequently, the movement of ground water and its associated constituents has also changed with time.

In addition to the Separations Area, groundwater mounding also occurs in the 100 and 300 Areas. Ground-water mounding in these areas is not as significant as in the Separations Area because of differences in discharge volumes and subsurface geology. However, in the 100 and 300 Areas, water levels are also greatly influenced by river stage.

Liquid Effluent Movement

If significant quantities of liquid effluents are discharged to the ground at the Hanford Site waste disposal facilities, then these effluents would percolate downward through the unsaturated zone to the water table. As

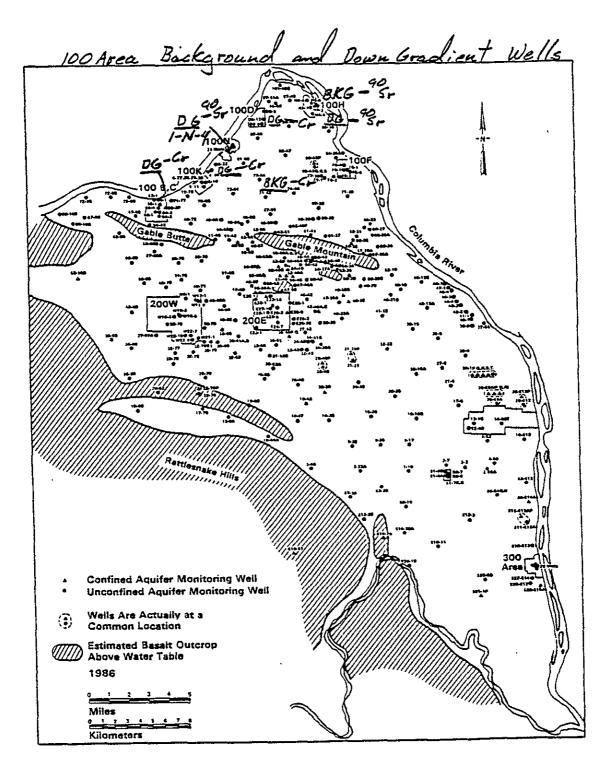


FIGURE 3.10. Location of Ground-Water Monitoring Wells Sampled in 1986 (first digit of well number has been dropped)

<u>Draft Phase I Installation Assessment of Inactive</u>

<u>Waste-Disposal Site at Hanford</u>, Volume 2, July 1986

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This site contains numerous sheets from the document; reference is made to the document instead of attaching all of the individual sheets.

Uncontrolled Hazardous Waste Site Ranking System; A Users Manual,
40 CFR 300, Appendix A

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Part 300, App. A

Physical state refers to the state of the hazadous substances at the time of disposations except that gasas perfected by the hazadous substances in a disposal sees should be considered in rating this factor. Each of the hazadous substances being evaluated to sestified a value as follows:

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Table 3—Comawment Value for Ground WATER ROUTE

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Table 3—Containment Value for Ground

WATER ROUTE -- Continued

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Toricity and Persistence have been com-bined in the matrix below because of their important relationship. To defermine the overall value for this combined action, evaluate each factor individually as disevaluate each factor indiciously as dis-cused below, March the indicidual values assigned with the values in the marine for the combined rating factor. Evaluate several

Environmental Protection Agency

of the most hazardous subatances at the fa-cility independently and enter only the highest score in the mateix on the work sheet.

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Persistence of each hazardous substance is evaluated on its blodegradability as follows:

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More specific information is given Tables 4 and 9.

FABLE 4-WASTE CHARACTERISTICS VALUES FOR SOME COMMON CHEMICALS

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Part 300, App. A

TABLE 4-WASTE CHARACTERISTICS VALUES FOR SOME COMMON CHEMICALS—Conlined

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TABLE 5-PERSISTENCE (BIODEGRADABLITY) OF SOME ORGANIC COMPOUNDS*	Value - 3 Highly Pershelent Compounds
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OF SOME DRAWING COMPOUNDS -- Continued TABLE 6-PERBISTENCE (BIODEORADABLITY)

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Environmental Protection Agency

TABLE 6-SAX TOXICITY RATINGS

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TABLE 7-NFPA TOXICITY BATINGS.

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Part 300, App. A

"Hatonal fee froischon Association Mattonal fee Chiling Vol 13, No. 49, 1873. TABLE 6—SAX TOXICITY RATINGS—Configured

hazardous substances at a facility (see received except that with a containment
that of D. Do not include amounts of contaminated soil or water in such care, the
amount of contaminating hazardous substance may be estimated.

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miles of the hazardous substance, including the georgaphical extent of the incesurable concentration in the extent for Azign a value using the following guidance: drawn from the aquifer of concern within 3 Ground water use indicates the nature of the use made of ground water

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Ground water use	Unacide (e.g., expressly peaks action, or groundy but yield (etc.). Commercial industrial implicit water source presently practicely, not used but	photological and purisped votes from the production of the production of the presence of the production of the productio

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U.S.G.S. Maps of the area around the 100 Area:

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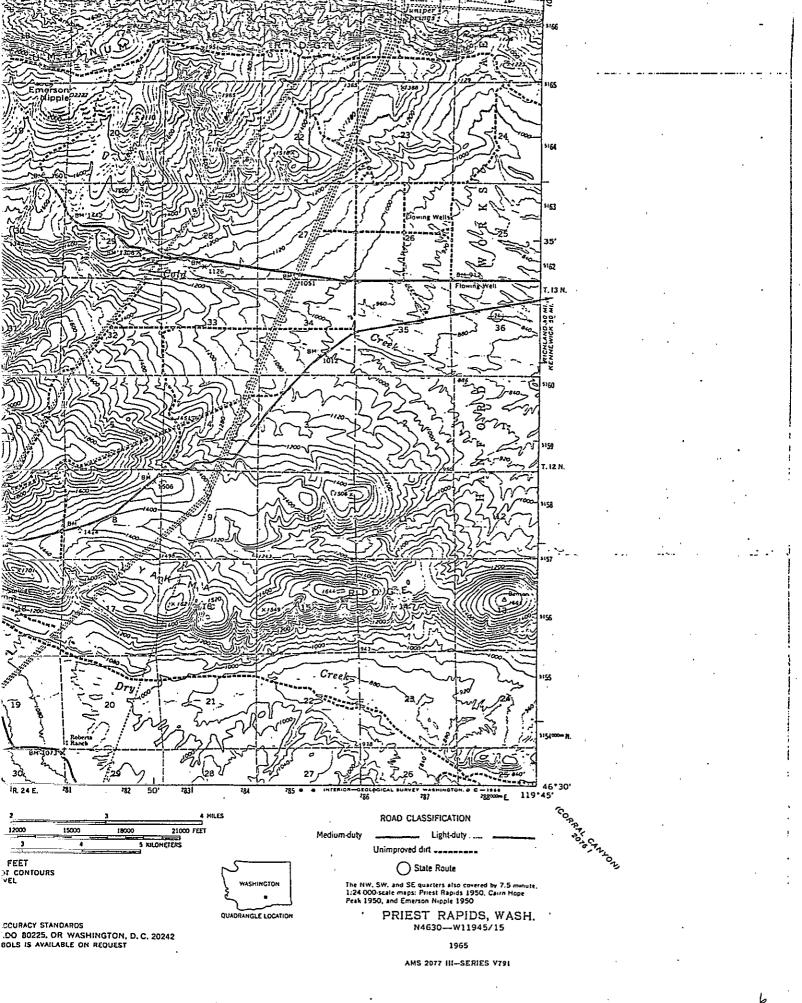
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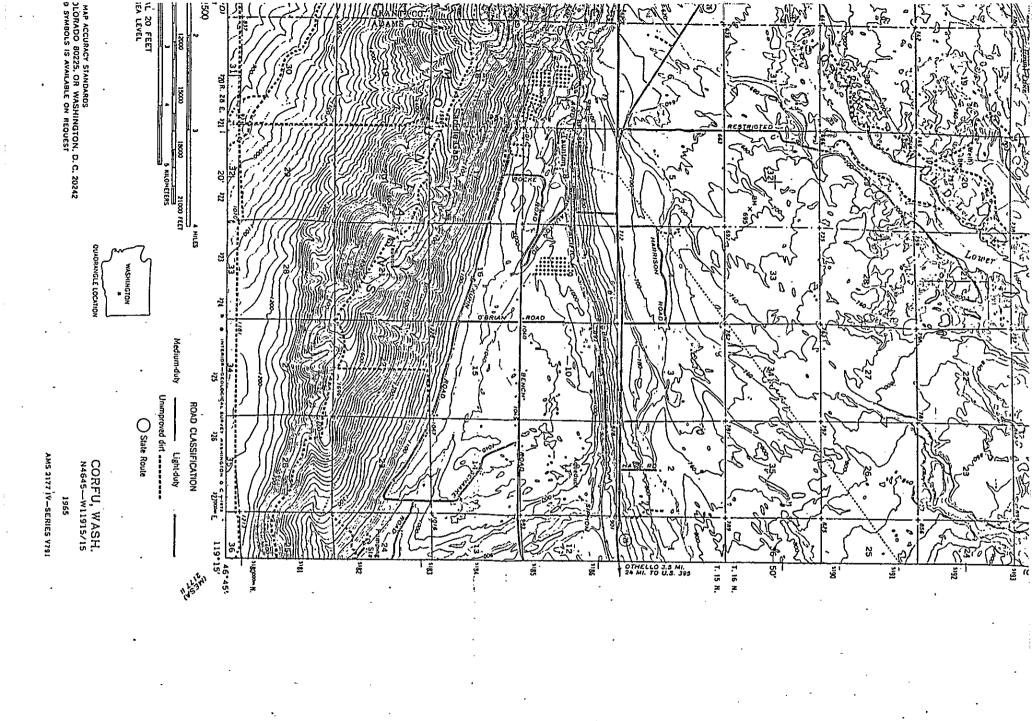
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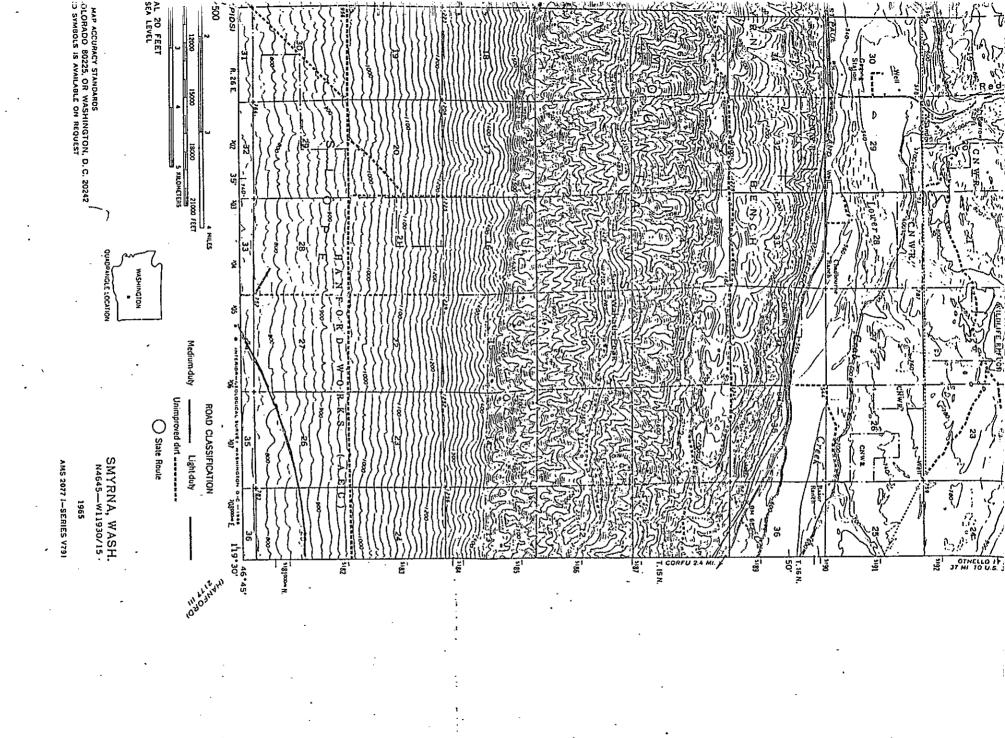
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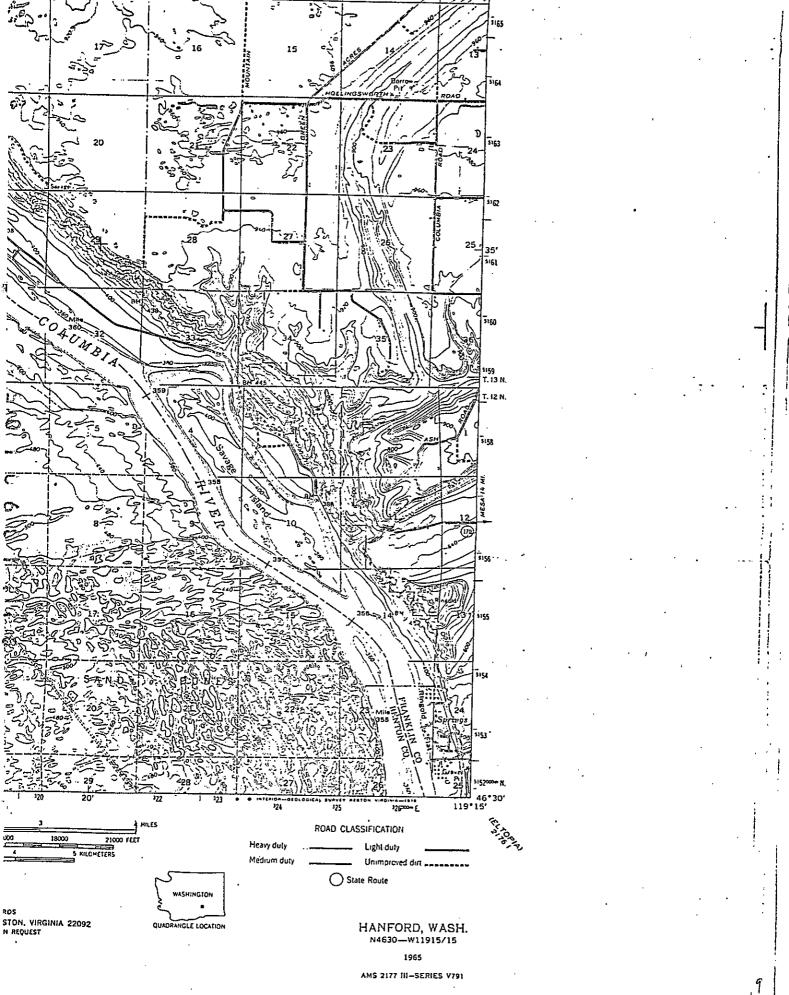
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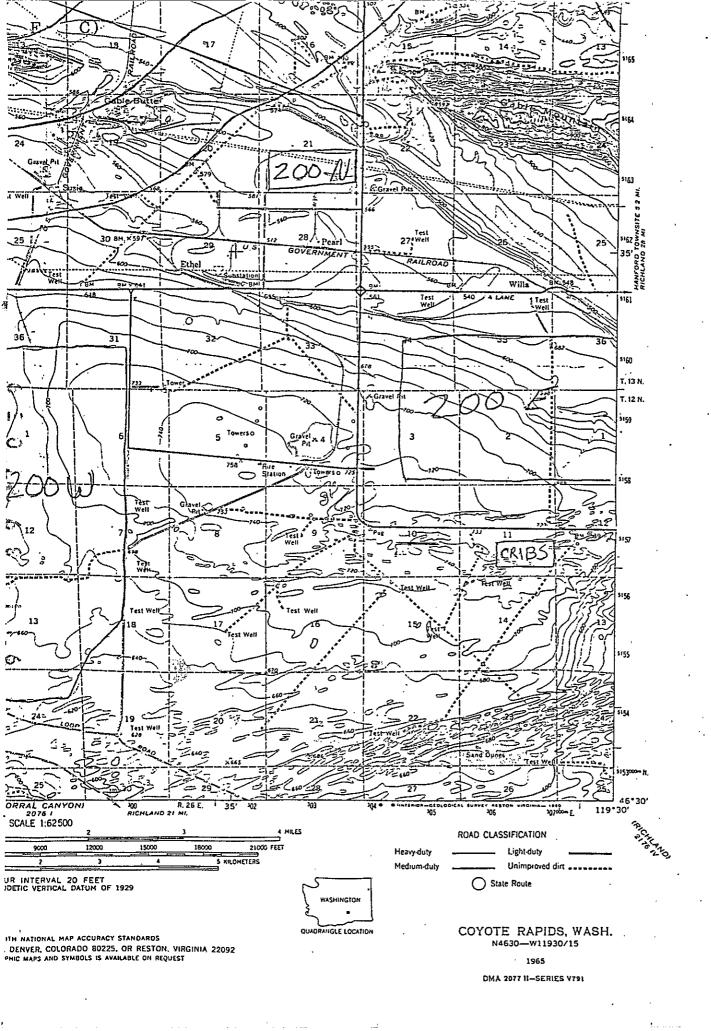
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Waste Management Operations, Volume 1, ERDA-1538

TABLE II.1-5a (Continued)

CHARACTERIZATION OF 300 NORTH AND WYE BURIAL GROUNDS

- Field survey of waste disposal sites begun late FY-1975
- Drilling of current series of sampling wells completed late FY-1976
- Estimate of any migration of radionuclides based on available data completed early FY-1977
- Final report and disposal site maps completed and issued late FY-1977

PLUTONIUM RECOVERY

 Detailed schedules will be developed based on the results of the 216-Z-9 Crib plutonium recovery program.

FISSION PRODUCT RECOVERY

No schedule for recovery has been established.

II.1.1.3 Plant Water Usage

8.2.J.L.C.8

Sanitary and process water for the Hanford Reservation operations are supplied from a number of different locations (Table II.1-6). The 100-8 pumphouse, with the 100-D as backup, supplies water to 100-B-C, 100-B, 100-H, 100-F, and 200 Areas. (All of these areas are shown on Figure II.1-1.) A water plant at 100-N Area supplies sanitary and process water to both 100-N and the WPPSS generating station located adjacent to 100-N. The 300 Area pumphouse supplies all water needs for the 300 Area. The City of Richland water supply system is intertied to the 300 Area system as an emergency backup. Various wells in the 600 Areas on the site supply both sanitary and process water.

An average of 440 cfs (0.4% of annual average flow) of Columbia River water is continuously withdrawn, mainly for cooling purposes. Most of this water is discharged directly back to the river as nonradioactive disposal. In addition, approximately 34.7 cfs (less than 0.03% of the total annual average flow) of Columbia River water is withdrawn and discharged to the ground as sanitary and industrial wastes. Wells supply approximately 73,000 gal/day for sanitary and minor irrigation use which is almost entirely discharged to the ground.

II.1.1.4 Waste Inventories [X.18]

Multiple disposal sites for radioactive waste have been used at Hanford during its 30-yr history. The choice of disposal site is based on many factors including half-life and toxicity of the elements involved, quantity of material to be discarded, proximity to water table and/or the Columbia River, and optimum personnel radiation exposure and contamination control. The use of many of these sites has been terminated.

The waste inventory data reported subsequently and in greater detail in Appendix II.1-C for cribs, burial grounds, ponds, ditches, specific retention sites, and unplanned releases are subject to variations in sample collection and analysis.

High-level waste streams are analyzed for plutonium, uranium and neptunium. Fission product content is calculated from irradiation history. The total contents of all waste tanks are known with fair precision, although the contents of individual tanks and, to some extent, of individual tank farms are less certain because of transfers among tanks and farms.

Discharges to cribs, specific retention sites, ponds and ditches are based on line samples. Since the concentration of radionuclides in these streams is often low, accuracy is low and considerable uncertainty is involved. In some cases, particularly for discharges to ponds and ditches, the radioactivity is below limit of detection and the results are therefore reported as "less-than" figures.

Unplanned releases of liquids to grounds are based on estimates of volumes involved and of concentrations of the radionuclides in the stream. Gaseous releases are based on stack samples.

TABLE II.1-6 PLANT WATER USE

Plant or Facility	Source of Supply	Primary Use
100-8	Columbia River 100-B Pumphouse	Process water
100-D	Columbia River 100-D Pumphouse	Process water
100-K	From 100-K Pumphouse	Process water
א-001	Columbia River-100-N Pumphouse	Process, sanitary and cooling water
100-F	From 100-8 Pumphouse	Process water
200 Areas	100-8 Pumphouse Well 299-E26-6 Wells 299-E28-11 and 15	Process and cooling water Emergency cooling water supply Emergency process supply
300 Area	Columbia River 300 Area Pumphouse	Process and sanitary 331 Fish Ponds
FFTF	Wells 699-S0-7 699-S0-8 699-S1-78	Construction and Sanitary
Atm. Physics	100-8 Pumphouse	Sanitary
609 Fire Sta. (100 Area Central)	100-8 Pumphouse	Sanitary
609 A fire Sta. -(200 Area Central)	100-8 Area	Sanitary
6652C (Aeronomy)	Rattlesnake Springs	Sanitary
6652I (Ale Hq)	Well 699-26-89	Sanitary and Irrigation
Emergency Relocation Ctr.	Well 699-S18-51	Sanitary
BY Telephone Exchange	Well 699-50-28A	Sanitary
Hanford Road Maintenance	Well 699-Han-9	Industrial and Sanitary
Vernita Park	Wells 699-72-101A, B and C	Sanitary and Irrigation

Reported quantities of radioactivity in burial grounds or storage tunnels are estimates based on operating history and radiation levels of the equipment or waste buried. Generally, these data are of greater uncertainty than liquid and gaseous streams since there is no practical method by which a representative sample, particularly of large equipment pieces, can be obtained.

Overall, however, the accuracy of the inventory total is believed to be within a factor of two, probably within 50% or better. The reported inventories in this final environmental impact statement have been adjusted to a single significant figure to reflect the uncertainties in the data.

II.1.1.4.1 Contained Solids and Liquids

II.1.1.4.1.1 Disposal Sites

The approximate inventories of solid waste disposal in the 100 Areas through 1972 summarized below are shown in Appendix II.1-8, Part 3. These inventories are corrected for decay through the end of 1972.

Inductively Coupled Plasma Method, Method 6010

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Chromium Method of Analysis.

METHOD 6010

INDUCTIVELY COUPLED PLASHA METHOD

1.0 Scope and Application

- 1.1 Method 6010 is approved for determining the concentration of trace elements in mobility procedure extracts (exclusive of arsenic and selenium), wastes, and soils.
- 1.2 This method may be used for the determination of total elements in groundwater if the detection limit is less than primary interim drinking water standards or less than background levels.
- 1.3 All samples must be subjected to an appropriate dissolution step prior to analysis. (Refer to Section 2.1.)
- 1.4 Precaution must be taken to ensure that all spectral, chemical and physical interferences are detected and the appropriate corrective measures taken.
 - 1.5 Use of this method is restricted to experienced spectroscopists.
 - 1.6 Elements for which Method 6010 is an approved procedure are:

Aluminum	Cobalt	Potassium
Antimony	Copper	Silica
8ari um	Iron	Silver
8eryllium	Lead	Sactium
Baran	Magnesium	Thallium
Cadmi um	Manganese	Vanadium
Calcium	Molybdenum	Zinc
Chromium_	Nickel	

2.0 Summary of Method

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- 2.1 Prior to analysis by Method 6010, samples must be prepared for sample introduction. The method of sample preparation will vary according to the sample matrix. Aqueous samples are subjected to the acid digestion procedure described in this method. Sludge-type samples are prepared using the procedure described in Method 3050. For samples containing oils, greases or waxes, the procedures described in Methods 3030 and 3040 may be applicable.
- 2.2 Method 6010 describes a technique for the simultaneous or sequential multielement determination of trace elements in solution. The basis of the method is the measurement of atomic emission by an optical spectroscopic technique. Samples are nebulized and the aerosol that is produced is transported to the plasma torch where excitation occurs. Characteristic atomic-line emission spectra are produced by a radio-frequency inductively coupled plasma

2 / MULTIELEMENT INORGANIC ANALYTICAL TECHNIQUES

(ICP). The spectra are dispersed by a grating spectrometer and the intensities of the lines are monitored by photomultiplier tubes. The photocurrents from the photomultiplier tubes are processed and controlled by a computer system. A background correction technique is required to compensate for variable background contribution to the determination of trace elements. Background must be measured adjacent to analyte lines on samples during analysis. The position selected for the background intensity measurement, on either or both sides of the analytical line, will be determined by the complexity of the spectrum adjacent to the analyte line. The position used must be free of spectral interference and reflect the same change in background intensity as occurs at the analyte wavelength measured. Background correction is not required in cases of line broadening where a background correction measurement would actually degrade the analytical result. The possibility of additional interferences named in 3.1 (and tests for their presence as described in 3.2) should also be recognized and appropriate corrections made.

3.0 Interferences

- 3.1 Several types of interference effects may contribute to inaccuracies in the determination of trace elements. They can be summarized as follows:
 - 3.1.1 Spectral interferences can be categorized as (1) overlap of a spectral line from another element; (2) unresolved overlap of molecular band spectra; (3) background contribution from continuous or recombination phenomena; and (4) background contribution from stray light from the line emission of high concentration elements. The first of these effects can be compensated by utilizing a computer correction of the raw data, requiring the monitoring and measurement of the interfering element. The second effect may require selection of an alternate wavelength. The third and fourth effects can usually be compensated by a background correction adjacent to the analyte line. In addition, users of simultaneous multielement instrumentation must assume the responsibility of verifying the absence of spectral interference for an element that could occur in a sample but for which there is no channel in the instrument array. Listed in Table 2 are some interference effects for the recommended wavelengths given in Table 1. The data in Table 2 are intended for use only as a rudimentary guide for indicating potential spectral interferences. For this purpose, linear relations between concentration and intensity for the analytes and the interferents can be assumed.
 - 3.1.1.1 The interference information, which was collected at the Ames Laboratory, is expressed as analyte concentration equivalents (i.e., false analyte concentrations) arising from 100 mg/liter of the interference element. The suggested use of this

Ames Laboratory, U.S. DOE, Iowa State University, Ames, Iowa 50011.

TABLE 1. RECOMMENDED WAVELENGTHS AND ESTIMATED INSTRUMENTAL DETECTION LIMITS

Element	Wavelength ² (nm)	Estimated detection limit (µg/1)
Aluminum	308.215	45
Arsenic	193.696	536
Antimony.	206.833	32
Barium	455.403	2
Beryllium	313.042	0.3
Soron	249.773	5
Cadmi um	226.502	4
Calcium	317.933	10
Chromium	257.716	7 7
Cobalt	228.616	7
Copper	324.754	. 6 7
Iron	259.940	. 7
Lead	220.353	42
Magnesium	279.079	30
Manganese	257.610	2
Molybdenum	202.030	. 8
Nickel	231.504	15,
Potassium	766.491	seed
Selenium	196.026	750
Silica (SiO ₂)	288.158	58
Silver	328.068	7
Sodium	588.995	29
Thallium	190.864	40
Vanadium	292.402	8 2
Zinc	213.856	2

aThe wavelengths listed are recommended because of their sensitivity and overall acceptance. Other wavelengths may be substituted if they can provide the needed sensitivity and are treated with the same corrective techniques for spectral interference (see 3.1.1). In time, other elements may be added as more information becomes available and as required.

bThe estimated instrumental detection limits as shown are taken from "Inductively Coupled Plasma-Atomic Emission Spectroscopy-Prominent Lines," EPA-600/4-79-017. They are given as a guide for an instrumental limit. The actual method detection limits are sample-dependent and may vary as the sample matrix varies.

CMethod may not be applicable for groundwater analysis for these metals.

dHighly dependent on operating conditions and plasma position.

TABLE 2. ANALYTE CONCENTRATION EQUIVALENTS ARISING FROM INTERFERENTS AT THE 100 MG/L LEVEL

						inte	rferent	(mg/1)				
Analyte	Wavelength (nm)	Al	Ca	Çr	Си	Fe	Hg	Mn	Ni	Ti	٧	
Aluminum	308,215							0.21		*=	1.4	
Ant imony	206.833	0.47		2.9		0.08				0.25	0.45	
Arsenic	193.696	1.3		0.44				**	+-	**	1,1	
Barium	455,403									*-		
Beryllium	313.042									0.04	0.05	
Boron	249.773	0.04				0.32						
Cadmium	226,502					0:03			0.02			
Calcium	317,933			0.08		0.01	0.01	0.04		0.03	0.0	
Chromium	267,716					0.003		0.04			0.0	
Cobalt	228,616			0.03		0.005			0.03	0.15		
Copper	324.754			~-		0.003				0.05	0.02	
lron	259.940							0.12		~-		
Lead	220.353	0.17							ap 400			
Hagnesium	279.079		0.02	0.11		0.13		0.25		0.07	0.12	
Hanganese	257.610	0.005		0.01		0.002	0.002					
Holybdenun	202.030	0.05				0.03						
Nickel	231.604											
Selenium	196.026	0.23				0.09						
Silicon	288,158		***	0.07							0.01	
Sodium	588.995									0.08		
Thallium	190.864	0.30										
Vanadium	292,402			0.05		0.005				0.02		
Zinc	213.856				0.14	*	ga ++•		0.29			

information is as follows: Assume that arsenic (at 193.696 nm) is to be determined in a sample containing approximately 10 mg/liter of aluminum. According to Table 2, 100 mg/liter of aluminum would yield a false signal for arsenic equivalent to approximately 1.3 mg/liter. Therefore, 10 mg/liter of aluminum would result in a false signal for arsenic equivalent to approximately 0.13 mg/liter. The reader is cautioned that other analytical systems may exhibit somewhat different levels of interference than those shown in Table 2, and that the interference effects must be evaluated for each individual system.

- 3.1.1.2 Only those interferents listed were investigated, and the blank spaces in Table 2 indicate that measurable interferences were not observed for the interferent concentrations listed in Table 3. Generally, interferences were discernible if they produced peaks or background shifts corresponding to 2-5% of the peaks generated by the analyte concentrations also listed in Table 3.
- 3.1.1.3 At present, information on the listed.silver and potassium wavelengths is not available but it has been reported that second order energy from the magnesium 383.231-nm wavelength interferes with the listed potassium line at 766.491 nm.

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- 3.1.2 Physical interferences are generally considered to be effects associated with the sample nebulization and transport processes. Such properties as change in viscosity and surface tension can cause significant inaccuracies especially in samples which may contain high dissolved solids or acid concentrations. The use of a peristaltic pump may lessen these interferences. If these types of interferences are operative, they must be reduced by diluting the sample and/or utilizing standard addition techniques. Another problem which can occur from high dissolved solids is salt buildup at the tip of the nebulizer. This affects aerosol flow rate, causing instrumental drift. Wetting the argon prior to nebulization, the use of a tip washer, or sample dilution have been used to control this problem. Also, the use of a high solid nebulizer can reduce sait build-up in the nebulizer and can also prevent drifting and loss of sensitivity in the instrument. In addition, it has been reported that better control of the argon flow rate improves instrument performance. This is accomplished with the use of mass flow controllers.
- 3.1.3 Chemical interferences are characterized by molecular compound formation, ionization effects and solute vaporization effects. Normally these effects are not pronounced with the ICP technique; however, if observed they can be minimized by careful selection of operating conditions (that is, incident power, observation position, and so forth), by buffering of the sample, by matrix matching, and by standard addition procedures. These types of interferences can be highly dependent on matrix type and the specific analyte element.

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TABLE 3. INTERFERENT AND ANALYTE ELEMENTAL CONCENTRATIONS USED FOR INTERFERENCE MEASUREMENTS IN TABLE 2

Analytes	mg/liter	Interferents	mg/liter
Al	10	Al	1,000
As	10	Ĉ.	1,000
8	10	Cr	200
8-a	ī	C ₁₂	200
8e	1	Ca Cr Cu Fe	1,000
Ca	1	Mg	1,000
Cd		Mn	200
Ca	i	N1	200
-Cr	10 1 1 1	Ti :	
Cu	1	A.	200
Fe	1		
Mg	1	•	
Mn	1		
· Mo	10 10		
Ма	10	•	
หร	10		
Pb	. 10		
Sb	10		
Se Si	10		
Si	1	•	
TI	10		
y Zn	10.		
Zn	10 .		

- 3.2 It is recommended that whenever a new or unusual sample matrix is encountered, a series of tests be performed prior to reporting concentration data for analyte elements. These tests, as outlined in 3.2.1 through 3.2.4, will ensure the analyst that neither positive nor negative interferences will have on any of the analyte elements and distort the accuracy of the reported values.
 - 3.2.1 Serial dilution: If the analyta concentration is sufficiently high (minimally a factor of 10 above the instrumental detection limit after dilution), an analysis of a dilution should agree within 5 percent of the original determination (or within some acceptable control limit that has been established for that matrix). If not, a chemical or physical interference effect should be suspected.
 - 3.2.2 Spike addition: A spike addition added at a minimum level of 10x the instrumental detection limit (maximum 100x) to the original determination should be recovered to within 90 to 110 percent or within the established control limit for that matrix. If not, a matrix effect should be suspected. The use of a standard addition analysis procedure can usually compensate for this effect. CAUTION: The standard addition technique does not detect coincident spectral overlap. If suspected, use of computerized compensation, an alternate wavelength, or comparison with an alternate method is recommended (see 3.2.3).
 - 3.2.3 Comparison with alternate method of analysis: When investigating a new sample matrix, comparison tests may be performed with other analytical techniques such as atomic absorption spectrometry or other approved methodology.
 - 3.2.4 Wavelength scanning of analyte line region: If the appropriate equipment is available, wavelength scanning can be performed to detect potential spectral interferences.

4.0 Apparatus and Materials

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- 4.1 Inductively coupled plasma-atomic emission spectrometer
- 4.1.1 Computer-controlled atomic emission spectrometer with background correction.
 - 4.1.2 Radiofrequency generator.
 - 4.1.3 Argon gas supply, welding grade or better.

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4.2 Operating conditions: Because of the differences between various makes and models of satisfactory instruments, no detailed operating instructions can be provided. Instead, the analyst should follow the instructions provided by the manufacturer of the particular instrument. Sensitivity, instrumental detection limit, precision, linear dynamic range, and interference effects must be investigated and established for each individual analyte line on that particular instrument. It is the responsibility of the analyst (1) to verify that the instrument configuration and operating conditions used satisfy the analytical requirements and (2) to maintain quality control data confirming instrument performance and analytical results.

5.0 Reagents

- 5.1 Acids used in the preparation of standards and for sample processing must be ultra-high purity grade or equivalent. Redistilled acids are acceptable.
 - 5.1.1 Acetic acid, conc.
 - 5.1.2 Hydrochloric acid, conc.
 - 5.1.3 Hydrochioric acid (1:1): Add 500 ml conc. HCl to 400 ml Type II water and dilute to 1 liter.
 - 5.1.4 Nitric acid, conc.
 - 5.1.5 Nitric acid (1:1): Add 500 ml conc. $\rm HNO_3$ to 400 ml Type II water and dilute to 1 liter.
- 5.2 Type II water: Prepare by passing distilled water through a mixed bed of cation and anion exchange resins. Use Type II water for the preparation of all reagents, calibration standards, and as dilution water. The purity of this water must be equivalent to or better than ASTM Type II reagent water of Specification D1193.
- 5.3 Standard stock solutions may be purchased or prepared from ultrahigh purity grade chemicals or metals. "All salts must be dried for 1 hr at 105" C unless otherwise specified. (CAUTION: Many metal salts are extremely toxic and may be fatal if swallowed. Wash hands thoroughly after handling.) Typical stock solution preparation procedures follow.
 - 5.3.1 Aluminum solution, stock, 1 ml=100 μg AI: Dissolve 0.100 g of aluminum metal in an acid mixture of 4 ml of (1:1) HCl and 1 ml of conc. HNO3 in a beaker. Warm gently to dissolve. When solution is complete, transfer quantitatively to a one liter flask and add an additional 10 ml of (1:1) HCl and dilute to 1,000 ml with Type II water.

- 5.3.2 Antimony solution, stock, I ml=100 μg Sb: Dissolve 0.2669 g K(Sb0)C₄H₄O₆ in Type II water, add 10 ml (1:1) HCl and dilute to 1.000 ml with Type II water.
- 5.3.3 Arsenic solution, stock, 1 ml=100 µg As: Dissolve 0.1320 g of As₂0₃ in 100 ml of Type II water containing 0.4 g NaOH. Acidify the solution with 2 ml conc. HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.4 Barium solution, stock, 1 mi=100 μ g Ba: Dissolve 0.1516 g BaCl₂ (dried at 250° C for 2 hr) in 10 ml Type II water with 1 ml (1:1) HCl. Add 10.0 ml (1:1) HCl and dilute to 1,000 ml with Type II water.
- 5.3.5 Beryllium solution, stock, 1 ml=100 µg 8e: 00 not dry. Dissolve 1.966 g BeSO4-4H2O in Type II water, add 10.0 ml conc. HNO3 and dilute to 1,000 ml with Type II water.

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- 5.3.6 Boron solution, stock, 1 ml=100 μg 8: Do not dry. Dissolve 0.5716 g anhydrous H₃BO₃ in Type II water and dilute to 1,000 ml. Keep in a tightly stoppered bottle, and store in a desiccator to prevent the entrance of atmospheric moisture.
- 5.3.7 Cadmium solution, stock, l ml=100 μg Cd: Dissolve 0.1142 g Cd0 in a minimum amount of of (1:1) HNO3. Heat to increase rate of dissolution. Add 10.0 ml conc. HNO3 and dilute to 1,000 ml with Type II water.
- 5.3.8 Calcium solution, stock, 1 mI=100 µg Ca: Suspend 0.2498 g CaCO₃ dried at 180° C for 1 hr before weighing in Type II water and dissolve cautiously with a minimum amount of (1:1) HNO₃. Add 10.0 ml conc. HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.9 Chromium solution, stock, 1 ml=100 µg Cr: Dissolve 0.1923 g CrO₃ in Type II water. When solution is complete, acidify with 10 ml conc. HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.10 Cobalt solution, stock, 1 ml=100 µg Co: Dissolve 0.1000 g of cobalt metal in a minimum amount of (1:1) HNO3. Add 10.0 ml (1:1) HCl and dilute to 1,000 ml with Type II water.
- 5.3.11 Copper solution, stock, 1 ml = 100 μg Cu: Dissolve 0.1252 g CuO in a minimum amount of (1+1) HNO3. Add 10.0 ml conc. HNO3 and dilute to 1,000 ml with deionized, distilled water.
- 5.3.12 Iron solution, stock, I ml=100 μ g Fe: Dissolve 0.1430 g Fe₂O₃ in a warm mixture of 20 ml (1:1) HCl and 2 ml of conc. HNO₃. Cool, add an additional 5.0 ml of conc. HNO₃ and dilute to 1,000 ml with Type II water.

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- 5.3.13 Lead solution, stock, 1 ml=100 μ g Pb: Dissolve 0.1599 g Pb(NO₃)₂ in a minimum amount of (1:1) HNO₃. Add 10.0 ml (1:1) HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.14 Magnesium solution, stock, I ml=100 µg Mg: Dissolve 0.1658 g MgO in a minimum amount of (1:1) HNO3. Add 10.0 ml (1:1) conc. HNO3 and dilute to 1,000 ml with Type II water.
- 5.3.15 Manganese solution, stock, 1 ml=100 μ g Mn: Dissolve 0.1000 g of manganese metal in the acid mixture (10 ml conc. HCl and 1 ml conc. HNO₃) and dilute to 1,000 ml with Type II water.
- 5.3.16 Molybdenum solution, stock, 1 ml=100 μg Mo: Dissolve 0.2043 g (NH₄)₂MoO₄ in Type II water and dilute to 1,000 ml with Type II water.
- 5.3.17 Nickel solution, stock, l ml=100 μg Ni: Dissolve 0.1000 g of nickel metal in 10.0 ml hot conc. HNO3, cool, and dilute to 1,000 ml with Type II water.
- 5.3.18 Potassium solution, stock, 1 ml=100 µg K: Dissolve 0.1907 g KCl, dried at 110° C, in Type II water and dilute to 1,000 ml.
- 5.3.19 Selenium solution, stock, 1 ml=100 µg Se: Do not dry. Dissolve 0.1727 g H₂SeO₃ in Type II water and dilute to 1,000 ml.
- 5.3.20 Silica solution, stock, 1 ml=100 μ g SiO₂: Do not dry. Dissolve 0.4730 g Na₂SiO₃·9H₂O in Type II water. Add 10.0 ml conc. HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.21 Silver solution, stock, 1 ml=100 μg Ag: Dissolve 0.1575 g AgNO3 in Type II water and 10 ml conc. HNO3. Dilute to 1,000 ml with Type II water.
- 5.3.22 Sodium solution, stock, l ml=100 μg Na: Dissolve 0.2542 g NaCl in Type II water. Add 10.0 ml conc. HNO3 and dilute to 1,000 ml with Type II water.
- 5.3.23 Thallium solution, stock, 1 ml=100 µg Tl: Dissolve 0.1303 g TlNO3 in Type II water. Add 10.0 ml conc. HNO3 and dilute to 1.000 ml with Type II water.
- 5.3.24 Vanadium solution, stock, 1 ml=100 μg V: Dissolve 0.2297 g NH₄VO₃ in a minimum amount of conc. HNO₃. Heat to increase rate of dissolution. Add 10.0 ml conc. HNO₃ and dilute to 1,000 ml with Type II water.
- 5.3.25 Zinc solution, stock, 1 ml=100 μ g Zn: Dissolve 0.1245 g ZnO in a minimum amount of dilute HNO3. Add 10.0 ml conc. HNO3 and dilute to 1,000 ml with Type II water.

5.4 Mixed calibration standard solutions: Prepare mixed calibration standard solutions by combining appropriate volumes of the stock solutions in volumetric flasks (see 5.4.1 through 5.4.5). Add 2 ml of (1:1) HNO3 and 10 ml of (1:1) HCl and dilute to 100 ml with Type II water (see Mote 1 in Section 5.4.5 and Note 6 in Section 7.3). Prior to preparing the mixed standards, each stock solution should be analyzed separately to determine possible spectral interference or the presence of impurities. Care should be taken when preparing the mixed standards that the elements be compatible and stable. Transfer the mixed standard solutions to a FEP fluorocarbon or unused polyethylene bottle for storage. Fresh mixed standards should be prepared as needed with the realization that concentration can change on aging. Calibration standards must be initially verified using a quality control sample and monitored weekly for stability (see 5.6.3). Although not specifically required, some typical calibration standard combinations follow when using those specific wavelengths listed in Table 1.

- 5.4.1 Mixed standard solution I: Manganese, beryllium, cadmium, lead. and zinc.
- 5.4.2 Mixed standard solution II: Barium, copper, iron, yanadium, and cobalt.
- 5.4.3 Mixed standard solution III: Molybdenum, silica, arsenic, and selenium.

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- 5.4.4 Mixed standard solution IV: Calcium, sodium, potassium, aluminum, chromium, and nickel.
- 5.4.5 Mixed standard solution V: Antimony, boron, magnesium, Silver, and thallium.

NUTE 1: If the addition of silver to the recommended acid combination results in an initial precipitation, add 15 ml of Type II water and warm the flask until the solution clears. Cool and dilute to 100 ml with Type II water. For this acid combination, the silver concentration should be limited to 2 mg/l. Silver under these conditions is stable in a tap water matrix for 30 days. Higher concentrations of silver require additional HCI.

- 5.5 Two types of blanks are required for the analysis. The calibration blank is used in establishing the analytical curve while the reagent blank is used to correct for possible contamination resulting from varying amounts of the acids used in the sample processing.
 - 5.5.1 The calibration blank is prepared by diluting 2 ml of (1:1) HNO3 and 10 ml of (1:1) HCI to 100 ml with Type II water (see note of in Section 7.3). Prepare a sufficient quantity to be used to flush the system between standards and samples.

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- 5.5.2 The reagent blank must contain all the reagents and in the same volumes as used in the processing of the samples. The reagent blank must be carried through the complete procedure and contain the same acid concentration in the final solution as the sample solution used for analysis.
- 5.6 Standards: In addition to the calibration standards, an instrument check standard, an interference check sample, and a quality control sample are also required for the analyses.
 - 5.6.1 The instrument check standard is prepared by the analyst by combining compatible elements at a concentration equivalent to the midpoint of their respective calibration curves (see 8.5.1).
 - 5.6.2 The interference check is prepared by the analyst in the following manner. Select a representative sample that Contains minimal concentrations of the analytes of interest but known concentrations of interfering elements that will provide an adequate test of the correction factors. Spike the sample with the elements of interest at the approximate concentration of either 100 µg/liter or 5 times the estimated detection limits given in Table 1. (For effluent samples of expected high concentrations, spike at an appropriate level.) If the type of samples analyzed are varied, a synthetically prepared sample may be used if the above criteria and intent are met. A limited supply of a synthetic interference check sample will be available from the Quality Assurance Branch of EMSL-Cincinnati.
 - 5.6.3 The quality control sample should be prepared in the same acid matrix as the calloration standards at a concentration near 1 mg/liter and in accordance with the instructions provided by the supplier.

6.0 Sample Collection, Handling, and Preservation

- 6.1 All samples must be collected using a sampling plan that addresses the considerations discussed in Section One of this manual.
- . 6.2 For the determination of trace elements, contamination and loss are of prime concern. Dust in the laboratory environment, impurities in reagents, and impurities on laboratory apparatus which the sample contacts are all sources of potential contamination. Samples containers can introduce either positive or negative errors in the measurement of trace elements by (a) contributing contaminants through leaching or surface desorption and (b) by depleting concentrations through adsorption. Thus the collection and treatment of the sample prior to analysis requires particular attention. Laboratory glassware including the sample bottle (whether polyethylene, polypropylene, or FEP-fluorocarbon) should be thoroughly washed with detargent and tap water; and rinsed with (1:1) nitric acid, tap water, (1:1) hydrochloric acid, tap water, and finally Type II water, in that order (see Notes 2 and 3).

- NOTE 2: Chromic acid may be useful to remove organic deposits from glassware; however, the analyst should be cautioned that the glassware must be thoroughly rinsed with water to remove the last traces of chromium. The use of chromic acid can cause a contamination problem for the determination of chromium if the glassware is not rinsed properly. A commercial product, NOCHROMIX, available from Godax Laboratories, 6 Varick Street, New York, NY: 10013, may be used in place of chromic acid. Chromic acid should not be used with plastic bottles.
- NOTE 3: If it can be documented through an active analytical quality control program using spiked samples and reagent blanks that certain steps in the cleaning procedure are not required for routine samples, those steps may be eliminated from the procedure.
- 6.3 Aqueous samples must be preserved and pretreated appropriately depending on whether data on dissolved, suspended, or total metals are desired.
- 6.4 Nonaqueous samples shall be refrigerated when possible, and analyzed as soon as possible.

7.0 Procedure

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- 7.1 The following definitions are intended to clarify the ensuing discussion.
 - 7.1.1 Instrumental detection limit: The concentration equivalent to a signal, due to the analyte, which is equal to three times the standard deviation of a series of twenty replicate measurements of a reagent blank signal at the same wavelength.
 - 7.1.2 Sensitivity: The slope of the analytical curve, i.e., functional relationship between emission intensity and concentration.
 - 7.1.3 Instrument check standard: A multielement standard of known concentration prepared by the analyst to monitor and verify instrument performance on a daily basis (see 5.6.1).
 - 7.1.4 Interference check sample: A solution containing both interfering and analyte elements of known concentration that can be used to verify background and interelement correction factors (see 5.6.2).
 - 7.1.5 Quality control sample: A solution obtained from an outside source having known concentration values to be used to verify the calibration standards (see 5.5.3).

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- 7.1.6 Calibration standards: A series of known standard solutions used by the analyst for calibration of the instrument (i.e., preparation of the the analytical curve; see 5.4).
- 7.1.7 Linear dynamic range: The concentration range over which the analytical curve remains linear.
- 7.1.8 Reagent blank: A volume of Type II water containing the same acid matrix as the calibration standards carried through the entire analytical scheme (see 5.5.2).
- 7.1.9 Calibration blank: A volume of Type II water acidified with HHO3 and HCI (see 5.5.1).
- 7.1.10 Method of standard addition: The standard addition technique involves the use of the unknown and unknown plus a known amount of standard (see 7.9.1).
- 7.2 Sample preparation: Aqueous samples should be prepared according to Section 7.3. Sludge-type samples should be prepared according to Method 3050 and samples containing oils, greases, or waxes may be prepared according to Methods 3030 and 3040. The applicability of a sample preparative technique to a new matrix type must be demonstrated by analyzing spiked sample or relevant standard reference materials.
- 7.3 Preparation of aqueous samples: For the determination of total elements, choose a measured volume of the well-mixed acid-preserved sample appropriate for the expected level of elements and transfer to a Griffin beaker (see note 5 below). Add 3 ml conc. HNO3. Place the beaker on a hot plate and evaporate to near dryness cautiously, making certain that the sample does not boil and that no area of the bottom of the beaker is allowed to go dry. Cool the beaker and add another 5-ml portion of conc. HNOq. Cover the beaker with a watch glass and return to the hot plate. Increase the temperature of the hot plate so that a gentle reflux action occurs. Continue heating, adding acid as necessary, until the digestion is complete (generally indicated when the digestate is light in color or does not change in appearance with continued refluxing). Again, evaporate to near dryness and cool the beaker. Add 10 ml of (1:1) HCl and 15 ml of Type II water per 100 ml final solution and warm the beaker gently for 15 min to dissolve any precipitate or residue resulting from evaporation. Allow to cool, wash down the beaker walls and watch glass with Type II water and filter the sample to remove insoluble material that could clog the nebulizer (see Note 4 below). Adjust the sample to a predetermined volume based on the expected concentrations of elements present. The sample is now ready for analysis (see Note 6 below). Concentrations so determined shall be reported as "total."
 - NOTE 4: In place of filtering, after diluting and mixing, the sample may be centrifuged or allowed to settle by gravity overnight to remove insoluble material.

NOTE 5: If low determinations of boron are critical, quartz or polypropylene glassware should be used. Polypropylene glassware is generally preferred since it is cheaper and just as accurate as quartz.

NOTE 6: If the sample analysis solution has a different acid concentration from that given in 7.3, but does not introduce a physical interference or affect the analytical result, the same calibration standards may be used.

- 7.4 Set up instrument with proper operating parameters established in Section 4.2. The instrument must be allowed to become thermally stable before beginning. This usually requires at least 30 min of operation prior to calibration.
 - 7.5 Initiate appropriate operating configuration of computer.
- 7.6 Profile and calibrate instrument according to instrument manufacturer's recommended procedures, using the typical mixed calibration standard solutions described in Section 5.4. Flush the system with the calibration blank (5.5.1) between each standard (see Nota 7 below). (The use of the average intensity of multiple exposures for both standardization and sample analysis has been found to reduce random error.)
 - NOTE 7: For boron concentrations greater than 500 µg/liter extended flush times of 1 to 2 min may be required.

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- 7.7 Before beginning the sample run, reanalyze the highest mixed calibration standard as if it were a sample. Concentration values obtained should not deviate from the actual values by more than ±5% (or the established control limits, whichever is lower). If they do, follow the recommendations fo the instrument manufacturer to correct for this condition.
- 7.8 Begin the sample run by flushing the system with the calibration blank solution (5.5.1) between each sample (see Note 7). Analyze the instrument check standard (5.6.1) and the calibration blank (5.5.1) after each 10 samples.
- 7.9 If it has been found that methods of standard addition are required, the following procedure is recommended.
 - 7.9.1 The standard addition technique involves preparing new standards in the sample matrix by adding known amounts of standard to one or more aliquots of the processed sample solution. This technique compensates for a sample constituent that enhances or depresses the analyte signal, thus producing a different slope from that of the calibration standards. It will not correct for additive interference which causes a baseline shift. The simplest version of this technique is the single-addition method. The procedure is as follows. Two identical aliquots of the sample solution, each of volume $V_{\rm x}$, are taken. To the first (labeled A) is added a small volume $V_{\rm x}$ of a standard analyte

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solution of concentration c_S . To the second (labeled B) is added the same volume V_S of the solvent. The analytical signals of A and B are measured and corrected for nonanalyte signals. The unknown sample concentration c_X is calculated:

$$c_x = \frac{S_B V_S c_S}{(S_A - S_B) V_X}$$

where S_A and S_B are the analytical signals (corrected for the blank) of solutions A and 8, respectively. V_S and c_S should be chosen so that S_A is roughly twice S_B on the average. It is best if V_S is made much less than V_X , and thus c_S is much greater than c_X , to avoid excess dilution of the sample matrix. If a separation or concentration step is used, the additions are best made first and carried through the entire procedure. For the results from this technique to be valid, the following limitations must be taken into consideration.

- The analytical response signals must be linear with respect to concentration.
- The chemical form of the analyte added must respond the same as the analyte in the sample.
- The interference effect most be constant over the working range of concern.
- 4. The signal must be corrected for any additive interference.

7.10 Calculations: Reagent blanks (5.5.2) should be subtracted from all samples. This is particularly important for digested samples requiring large quantities of acids to complete the digestion. If dilutions were performed, the appropriate factor must be applied to sample values. All results should be reported in mg/liter with up to three significant figures.

8.0 Quality Control

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- 8.1 All quality control data should be maintained and available for easy reference or inspection.
- 8.2 Dilute samples if they are more concentrated that the highest standard or if they fall on the plateau of a calibration curve.
- 8.3 Employ a minimum of one procedural blank per sample batch to determine if contamination or any memory effects are occurring.
- 8.4 Run one duplicate sample for every 10 samples. A duplicate sample is a sample brought through the whole sample preparation process.

- 8.5 Check the instrument standardization by analyzing appropriate quality control check standards as follows:
 - 8.5.1 Analyze an appropriate instrument check standard (5.6.1) containing the elements of interest at a frequency of 102. This check standard is used to determine instrument drift. If agreement is not within 15% of the expected values or within the established control limits, whichever is lower, the analysis is out of control. The analysis should be terminated, the problem corrected, and the instrument recalibrated.
 - 8.5.2 Analyze the calibration blank (5.5.1) at a frequency of 10%. The result should be within the established control limits of 2 standard deviations of the mean value, if not, repeat the analysis two more times and average the three results. If the average is not within the control limit, terminate the analysis, correct the problem, and recalibrate the instrument.
 - 8.5.3 To verify interelement and background correction factors, analyze the interference check sample (5.6.2) at the beginning, end, and at periodic intervals throughout the sample run. Results should fall within the established control limits of 1.5 times the standard deviation of the mean value. If not, terminate the analysis, correct the problem, and recalibrate the instrument.
 - 8.5.4 A quality control sample (5.6.3) obtained from an outside source must first be used for the initial verification of the calibration standards. A fresh dilution of this sample shall be analyzed every week thereafter to monitor their stability. If the results are not within +5% of the true value listed for the control sample, prepare a new calibration standard and recalibrate the instrument. If this does not correct the problem, prepare a new stock standard and a new calibration standard and repeat the calibration.
- 8.6 Spiked samples or standard reference materials shall be employed periodically to ensure that correct procedures are being followed and that all equipment is operating properly.

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- 8.7 The method of standard additions shall be used for the analysis of all EP extracts and whenever a sample suffers from matrix interferences.
- 8.8 The method detection limit (MBL) is defined as the minimum concentration of a substance that can be measured and reported with 99% confidence that the value is above zero. The MDL concentrations listed in Table 1 were obtained using reagent water. Similar results were achieved using representative wastewaters. The MDL actually achieved in a given analysis will vary depending on instrument sensitivity and matrix effects.
- 8.9 In an EPA round-robin phase I study, seven laboratories applied the ICP technique to acid-distilled water matrices that had been dosed with various metal concentrates. Table 4 lists the true value, the mean reported value and the mean % relative standard deviation.

REFERENCE 11

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Hanford Stream data base printout showing Priest Rapids Dam data as background data

ENVIRONMENTAL SAMPLE ANALYSIS RESULT REPORT

21 DEC 82 to 15 MAR 83

WATER

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RIVER WATER

CUMULATIVE

UNFILTERED

Samp #: 1265

PRIEST RAPIDS-RIVER

121/SR 90

Use this sample for back ground comments

Date Off	Date On	Result +/- 2 Sigma (PCI/L)	Comment
21 DEC 82	23 NOV 82	3.37E-01 +/- 1.83E-014	3 1/ C
18 JAN 83	21 DEC 82	1.89E-01 +/- 2.44E-02	
15 FEB 83	18 JAN 83	1.91E-01 +/- 3.18E-02	·
15 MAR 83	15 FEB 83	2.57E-01 +/- 4.11E-02	

Fraction of Results>DL: 4/ 4 Mean: 2.44E-01

Minimum: 1.89E-01 (18 JAN 83) Standard Error of Mean: 8.64E-02 Maximum: 3.37E-01 (21 DEC 82) Standard Deviation: 1.73E-01

Median: 1.91E-01

REFERENCE 12

Investigation of Ground-Water Seepage from Hanford Shoreline

of the Columbia River, PNL-5289, November 1984

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Investigation of Ground-Water Seepage from the Hanford Shoreline of the Columbia River

W. D. McCormack
J. M. V. Carlile

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November 1984

Prepared for the U.S. Department of Energy under Contract DE-AC06-76RLO 1830

Pacific Northwest Laboratory
Operated for the U.S. Department of Energy
by Battelle Memorial Institute



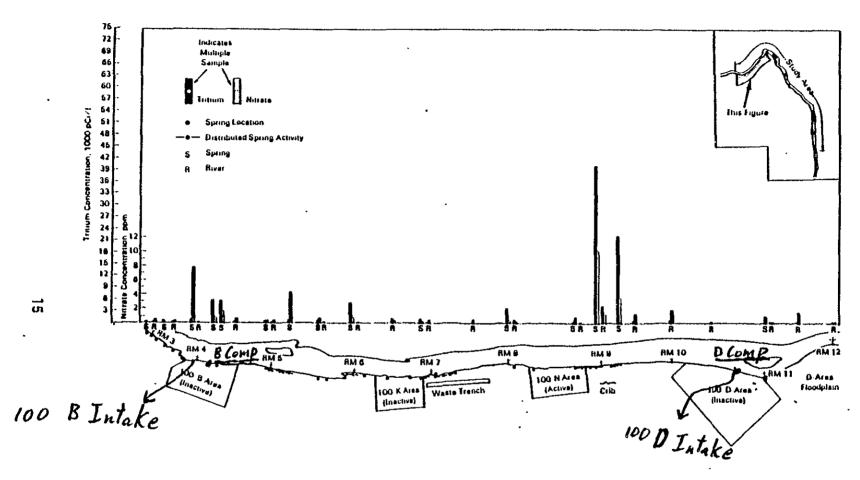


FIGURE 3. Locations and Analytical Results for Spring and River Samples from River Mile 3 through River Mile 12

in PNL.5289

TABLE B.1. Strontium-90 Analyses from Columbia River Samples

River Mile Location	,	• Sam II	<u>.</u>	Date Collected	Concentration, pCi/1 ±2σ	
3.0-5.0	В	Comp	RW(a)	01/22/83	0.55 ± 0.23	DG
5.5-7.5		Comp		12/18/82	0.18 ± 0.02	DE Jan grade
8.0-9.5	N	Comp	RM	12/18/82	28 ± 0.47	sample Con
10.0-12.0	D	Comp	RW	12/18/82	1.1 ± 0.05	DG our grast
14.0-17.5	Н	Comp	RW	01/22/83	0.50 ± 0.14	
18.0-22.0	F	Cemp	RW	01/22/83	0.93 ± 0.15	pecame
Upstream Columbia River Concentration (Average 1983)					0.18 ± 0.22	chase sampling date
DOE Concentration Guide (USDOE 1981)					300	magnitude
(a) Comp-RW denotes (aliquots from imm					omprised of	higher

TABLE B.2. Iodine-129 Analyses from Spring and Columbia River Samples

River Mile Location	Sample ID	Date Collected	Concentration, pCi/1 ±2σ
27.0	27.0 RW ^(a)	01/22/83	$3.3 \times 10^{-6} \pm 1.4 \times 10^{-6}$
27.0	27-1 Sp ^(b)	09/11/83	$1.6 \times 10^{-4} \pm 2.1 \times 10^{-5}$
28.0	28-2 Sp	09/11/83	$6.2 \times 10^{-2} \pm 6.8 \times 10^{-3}$
29.0	29.0 RW	01/22/83	$6.3 \times 10^{-5} \pm 5.0 \times 10^{-6}$
31.75	31-5 Sp	09/11/83	$3.0 \times 10^{-5} \pm 4.0 \times 10^{-6}$
32.5	32-0 Sp	09/11/83	$4.4 \times 10^{-5} \pm 2.7 \times 10^{-5}$
Upstream Columbia River Concentration (1983 Average)			$2.4 \times 10^{-5} \pm 2.6 \times 10^{-5}$
DOE Concentration Guide (USDOE 1981)			60

⁽a) RW denotes composite river water sample.(b) Sp denotes river bank spring sample.

(in PNL- 5289)

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REFERENCE 13

Hanford Reservation Area Worker Census, BNWL-2298, July 1977

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TABLE 1. Concentration of Hanford Reservation Workers by Site

Site	Number of Workers	Percent of Total
100	760	5
200 E&W	2,355	16
WPPSS #1,2,&4	2,905	20
FFTF	2,420	16
300	3,110	21
Battelle, et al.	3,345	22
TOTAL	14,895	100

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While the worker counts being reported by Reservation employers are usually shown concentrated around a designated site, in reality a substantial portion of these workers are likely to be distributed over the surrounding area. For convenience, however, they are credited to such particular sites as 200 East, 200 West, WPPSS 1, 2, & 4, etc.

Identification of shift workers posed some reporting difficulties since some firms run four shifts while most of the others conduct their operations in three. The 100 Area was a special problem since these workers operate over a wide area. Regardless, all workers have been accounted for in this census although some of the shift counts may be approximate.

DISTRIBUTION OF WORKERS BY RADII AND COMPASS DIRECTION

Figure 2 maps the distribution of Hanford Reservation workers by work shift over intervals of one-mile radii and 16 compass directions centered at the Purex Plant. These same worker distributions are repeated in Figure 3 without the maped Reservation Area as a background. As a tabulating convenience, sector parcel counts have been rounded to units of 5 and 10, but were adjusted to the total count for the separate companies. (Because of confidentiality, worker counts for the separate companies are not being presented here.) For better readability, sector counts within the first two mile radii from the Purex Plant center are presented separately at the bottom of the figure. Table 2 presents work distribution in detail including a cumulative count of workers and percent of total as distance and direction from the Purex Plant center increases.

REFERENCE 14

Endangered and Threatened Wildlife and Plants, 50 CFR,
Part 17, Subpart B, October 86

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populations of the same vertebrate

subspecies or species shall be identi-

fied by their differing geographic

boundaries, even though the other two

columns are identical. The term Entire" means that all populations

throughout the present range of a ver-

tebrate species are listed. Although

common names are included, they

cannot be relied upon for identifica-

tion of any specimen, since they may

vary greatly in local usage. The Serv-

ices shall use the most recently accept-

ed scientific name. In cases in which confusion might arise, a synonym(s)

will be provided in parentheses. The Services shall rely to the extent practicable on the International Code of Zo-

(c) In the "Status" column the following symbols are used: "E" for En-

dangered, "T" for Threatened, and "E

ological Nomenclature.

FEDERAL REGISTER document. That document, at least since 1973, includes a statement indicating the basis for the listing, as well as the effective

date(s) of said listing. (2) The "Special rules" and "Critical habitat" columns provide a cross reference to other sections in Parts 17, 222. 226, or 227. The "Special rules" column will also be used to cite the special rules that describe experimental populations and determine if they are essential or nonessential. Separate listing will be made for experimental populations, and the status column will include the following symbols: "XE" for an essential experimental population and "XN" for a nonessential experimental population. The term "NA" (not applicable) appearing in either of these two columns indicates that there are no special rules and/or critical habitat for that par-

ticular spec propriate through 2: that species other rules such wildl quirements. references column list two Service: species or t Federal age ernments.

(g) The li includes all example, th bons) is liste ticular species. However, all other appropriate rules in Parts 17, 217 through 227, and 402 still apply to that species. In addition, there may be other rules in this title that relate to such wildlife, e.g., port-of-entry requirements. It is not intended that the references in the "Special rules" column list all the regulations of the two Services which might apply to the species or to the regulations of other Federal agencies or State or local governments.

(g) The listing of a particular taxon includes all lower taxonomic units. For example, the genus *Hylobates* (gibbons) is listed as Endangered through-

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out its entire range (China, India, and SE Asia); consequently, all species subspecies, and populations of that genus are considered listed as Endan gered for the purposes of the Act. In 1978 (43 FR 6230-6233) the species Haliaeetus leucocephalus (bald eagle) was listed as Threatened in "USA (WAOR, MN, WI, MI)" rather than its entire population; thus, all individuals of the bald eagle found in those five States are considered listed as Threatened for the purposes of the Act.

(h) The "List of Endangered and Threatened Wildlife" is provided below:

Scientific rane	Patone range Production by a faired Patone range Production by a faired Production by		operate a		Vertebrate				
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		1	Sikkim.					
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Deer, Shanai sika		Chins (Shansi Province)		E	50	NA	NA	
Deer, South China sika		Southern China		E	50	NA	NA	
Dear, awamp (=-barasingha)		India, Nepal		, E	3	NA	NA	
Deer, Yarkand		China (Sinklang)		E	50	NA	NA	
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Elephant, Asian		South-central and Southeast Asia		Ė	15	NA	NA	
Farret, black-fooled		Western U.S.A., Western Canada			1.3	NA.	NA.	
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Fox, San Joaquin kil		USA (CA)		اع	ĭ	NA.	NA.	
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Gazelle, Clerk's (=Dibatag)		Somalia, Ethiopia		Ē	3	NA.	NA.	
Gazelle, Cavier's		Morocco, Algeria, Tunisia			3	NA	NA.	
Gazelle, Mhorr		Morocco			3	NA.	NA.	
Gazelle, Moroccan (-Dorcas)		Morocco, Algeria, Tunisla			3	NA.	NA NA	
Gazelle, Rio de Oro Dama		Western Sahara			3	NA NA	NA NA	
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nangaroo, eastern gray	Macropus pyantous (all aubapecies	Australia	_	-	: ^	ž	
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hangatoo, weatern gray	Macropus higmostas	Australia	æ	; p-		_	1707 60
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	Spe	cles		Vertebrate	Sta-	When	Critical	0	
	Common name	Scientific name	Historic range	endangered or population where	tua Gra	Nated	habitat	Special	7.11
	Lomius.	temuridae (Incl. Cherogaleidae, Lepile- muridae); all members of genera Lemur, Phener, Hapalemur, Lepilemur, Microcobus, Allocebus, Cherogaleus,	Malagasy Republic (≃Madagascar)		, E	3, 15	NA	NA	
	Leopard,,,,,,	Varocia. Panthera pardus	Africa, Asia	Wherever found, except where it is listed as Threatened as	E	3, 5, 114	NA	NA	
	Do			set forth below. In Akince, in the wild, south of, and including, the following countries: Gabon, Congo, Zake, Uganda, Kenya.	T	3, 5, 114	NA	17.40(I)	
	Leopard, clouded	Neorelis nebulosa	Southeast and south-central Asia, Talwan		E	3, 15	NA.	. NA	
72	Leopard, anow	Panthera uncia	Central Asia	do	E	5	NA	NA	
w	Linsang, spotled	Prioriodon pardicolor	Nepal, Assam, Vietnam, Cambodia, Laos, Burma.	do	E	15	NA) NA	
	Lion, Asiatic	Panthera leo persica	Turkey to India	do		3	NA NA	NA NA	
	Loris, lessor slow	Nyclicebus pygmaeus	Indochina	do		t 8	NA.	17.40(c)	
	Lynx, Spanish	Felis (= Lynx) perdine	Spain, Portugal	,,do,	E	3	NA.	NA.	
	Macaque, Formosan rock	Macaca cyclopis	Talwan	do	T	16	NA	17.40(c)	1 10
	Macaque, Japanese	Macaca hiscala		do	T	18	NA .	17.40(c)	50
	Macague, Ilon-tailed	Afacaca sienus	India	do		3	NA	NA.	CE
	Macaque, atimp-talled	Afacaca arctokles	India (Assam) to southern China	do		10	NA.	17.40(c)	X
	Macaque, Toque	Macaca sinica	Sri Lanka (- Ceylon)	do		16	NA	17 40(c)	_
	Manatee, Amazonian	Trichechus louoguis	South America (Amazon River Basin)	do		3	NA	NA.	Ü
	Manales, West African,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Trichechus senegalensis	West Coast of Africa from Senegal River to Cuanza River.	do	T	52	, NA	NA.	-
•	Manatee, West Indian (Florida)	Trichechus menetus	U.S.A. (southeastern), Caribbean Sea, South America.	do	Ε	1, 3	17.95(a)	NA.	(10
	Mandell	Papio sprint	Equatorial West Africa	ldo	E	16	NA	NA.	9
	Mangabey, Tana River	Carcocebus galaritus				3	NA :	NA.	Ī.
	Mangabey, white-collared	Cercocebus torquetus				16	NA.	NA.	
	Kargay		U.S.A. (TX), C. and S. America			5	NA :	NA.	
	Markhor, Kabal		Afghanistan, Pakistan			15	NA.	NA.	
	Markhor, straight-horned					15	NA.	NA.	=
	Marmosel, buff headed		Brasil	40	ΙĒ		NA I		7

Marmosel, builty lutted-ear Califfrix facchus aunita do E 233 NA NA SA Marmosel, Collon-top Saguinus oedigus Costa filica to Colombia do E 16 NA NA in Marmol, Vancouver Island Marmola vancouverensis Canada (Vancouver Island do E 13 NA NA Marsupial, eastern jectos Antechnomys langer Australia

idition)	U.S. Fish and Wildlife Serv., Interior	17.1
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- 00 - 00 - 00 - 00 - 00 - 00 - 00 - 00	66	do
Real	Coute Rice to Colombia Bazai, Colombia, Ecuador, Peru, Bohrie Candada (Vancouver tisind) do do do Congo, Camencon, Gabon. Cangro, Camencon, Gabon. Marko, Quatemata, Betze Costs West Africe Marko to South America Coute Rice, Nestague Costs Rice, Reague Costs Rice, Panama Marin Nigela. Marin Nigela. Marin Nigela. Andas of northern Peru. US A. (AL) Austrella.	Araban Pennauta Cameroon, Nigoria
Children farcept	Calibrara Jacchus susta Saguinus oedipus Calibrico gooksi Municals unincurvanias Saruthopsis psammophila Saruthopsis psammophila Saruthopsis psammophila Saruthopsis psammophila Calchous salunas Aktuatta pulisti (= vilipsis) Caccopilhecus dana. Aktuatta pulisti (= vilipsis) Caccopilhecus salunas Mustita pulisti (= vilipsis) Massis larvalua basisi Caccopilhecus sylthogasiar Maleis geofilioty duniatus Caccopilhecus sylthogasiar Aldisis geofilioty duniatus Caccopilhecus sylthogasiar Caccopilhecus sylthogasiar Maleis geofilioty duniatus Caccopilhecus sylthogasiar Caccopilhecus sylthogas	Oya bucoya. Aoya (Parsoya) conpica merodon
Markor, Kubalamanananananananananananananananananan	Marmossi, buity tufied-sar. Marmossi, Cotton-top. Marmossi, Goekd's. Marmossi, Goekd's. Marsupial-mouse, large disself. Marsupial-mouse, large disself. Marsupial-mouse, large disself. Marsupial-mouse, large disself. Monkey, balack Cokbus. Monkey, black Cokbus. Monkey, potosecs. Monkey, potosecs. Monkey, red ested nose-spotted. Monkey, red besided. Monkey, gelove-tailed woolly. Monkey, yellove-tailed woolly. Monkey, yellove-tailed woolly. Monkey, yellove-tailed woolly. Monkey, yellove-tailed woolly. Monkey, Alabama basch. Mouse, Alabama basch. Mouse, Choclawhatchee beach. Mouse, Gould's. Mouse, Gould's. Mouse, Gould's. Mouse, Shark Bay. Mouse, Shark May. Mouse, Shark Bay. Mouse, Shark Bay. Mouse, Shark May. Mouse, Shark Bay. Mouse, Shark May. Mouse, May. Muniac, M	Oryx, Authan Orier, Carnwoon clawless

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	Qp	9	Ş		do	9	ş	4		8			Q0	Ş	4		Qo	Ş	ş	9		90	dode		8	ş		4		00	dp		Q	Q	8	-8	Ą			į		00	db	do	9	do	Ş	46		99	9	qo	90	ch.	ş
	Costa Rica to Colombia	Brazil, Colombia, Ecuador, Peru, Bohria	Cunada Mencouver Jelandi	CEIRCLE CARROLL SIGNATURE TO STATE OF THE ST	Australia	90	9			Equational Gunea, People a Nepublic of	Conco. Cameroon, Galbon.		Mexico, Guelemele, Bezze	Coastal West Atrice	Liveloo to Couth America	THE PROPERTY OF THE PERSON OF	Upper Eastern Congo Basin, Cameroon	Cameroon	Borner	Court Dies Desemb	CORRESPONDE LA COMPANION CONTRACTOR CONTRACT	Weller Nigera	Nigeria, Cameroon, Fernando Po	Costs Rica, Niceragus	Costa Rica, Panama	Kanva			D144	Andes of northern Peru	Janzaria	U S.A. (AL)	Austrella	2	IISA (FI)	Attstration				•	Australia	U.S.A. (AL, PL)	U.S.A. (CA)	Austraka	do.	Ş	9	Aloghan The Hand D. one	Number of the second particle commenced to the commenced of the commenced	AUSURIA		U.S.A. (AZ, TX) to C. and S. America	Borneo, Sumetra	Arablan Pennaula	Cameroon Monda
	Supuinus cadipus	Californico cookial	flacing to concentrate of the	MARTINGLA VAINCHONDERINGS	Antechnomys langer	Snurthoosis osammonhile	Small Consist Sommer Sets	Charles de la ch	WHILE INTRACE LANGUAGE CONTINUES	Colobus satarias			Abustis plus	Carmonithacus datas	4.50 and a 10.00 d		Carcopithecus hoasii	Colobia herbut waissi	Monetie legentes	0.1 - 1.2 4.2 4.2 4.2 4.2 4.2 4.2 4.2 4.2 4.2 4	SHIM ON SIGHE CONTRACTOR OF THE PROPERTY.	Cercopinacus enthrogester	Cercopithecus anthrolis	Atalas geoffort honews	Atales ocotron cenamensis	Catching adomitating (hadais) cultons.	TOTAL COLUMN TO A		DESCRIPTION OF SECURITION OF S	Lagother Havisauda	Colobus Arki	Peromyscus potonofus ammobates	Przomys (= Notomys) pedwyczjałus	Notomes antilo	Paromychie antiophie altorhose	Peruchana fabri	Pendane south	Commence of the second	recompleting goalstein and another com-		Pseudomys novieholiandaee	PAROMYSCUS POLONOLUS LISSYROPSIS	Reuthrodontomys raviventns	Pseudomys precous.	Pseudomys shortridool	Psarchmys hamins	Peandoons occidentate	The first of the first	ALVINELLE TOUR	DESTRUCTION VANITABLE	Myrmacobus lascalul	Fets pardals	Pango pyametus	Chx bucong	Anna (Parana) monte monte
SELECTION OF PRINCIPAL PROPERTY OF THE PROPERT	armotel, cotton-log	fermonat Goald's	CLEORAL COUNTY Communications in the communication of the communication	larmot, Vancouver Bland	largupial, eastern jerboe	tersuible/moune taxon depart	and an order	The Contract of the Contract o	terten, romosen yezow-troeted	tookey, black colobus			lookey, black howler	Sates: Depart	COMPT CALLS ASSESSMENT TO THE PROPERTY OF THE PARTY OF TH	Konkey, Dowler	lookey L'hoest's	Inches, Brains, rad colohica			Konkey, red-pecked aquiriel	lookey, red belied	tonkey, red-sared nose-apotted	lankey, saider	online and a	coton: Test Diese and Adobus	Ulhay, talls there is concusting		DOKEY, WOOLY EDIGET	onkey, yellow-tailed woolly	onkey, Zanzibar red colobus	Ouse, Alabama beach	nuse. Australian nativo	Australian nativa	Checkenther heach	Cuse Bish's	Source Courte	CARD, GAMO Branches and the second se	Dute, ney Largo Collun	:	Ouse, New Housend	Ouse, Perdido Key beach	louse, sell marsh harvest	louse, Shark Bay	Cours Shorteidos	Come Smoke	Ches Lealing	DEPT. N CALONIA STATEMENT OF THE PROPERTY OF T	Unige, Tol S	sive-cal, satisformment	umbalal	Celot		nx, Author	Mor Camuroco chaudean

	Common name	Scientific name	Historic range	Vertebrate population where endangered or threatened	Sta- tus	When Sated	Critical habitat	Special rules	17.11
		_					NA	NA	
Ottor, of	ani	Pteronura brasiliensis	South America	do	E	2 4 6	NA.	NA.	
	ong-taked			do	E.	3, 15	NA NA	NA.	
	381118] Lutte felins		60	-	15	NA NA	NA.	
	outhern river	Lutra provocax		,00	Ê	15 21	NA AR	NA.	
Ottor, a	outhern aba	Enhydra lutris nerels	West coast U.S.A. (WA, OR, CA) south to Mexico (Baje California).		'		,,,,		
Onnda	glant	Aturopoda melanoleuca	. People's Republic of China	,ðo	E	139	NA	NA.	
	u (-scely eulosia)		.) Atrica	do	E	1 15	NA.	NA.	
	Florida	""1 <u>"</u> "	.) U.S.A. (LA and AFI east to SC and FL)	do	E) 1	NA.	NA.	
	is this amountaine and an annual and a second		Austraka	do	E	1	NA	NA.	
Lizinĝa	10 Riffe efterenenenenterrenenenenenenenenenen	subthissona)	[Ì.	l		
Charles .	s, southern	Planigala tenukostris	.[d0,			NA.	NA.	
	ine, thin-spiced	Chaelomys subspinosus	. Brazii	do	Ē	3	NA.	NA.	
	n, Leapheater's		4	,.do ,	E	233	NA	NA.	
	n, mountain pypmy	1 -	Attaliane	, d 0	E	1 1	NA.	NA.	
	n, scaly-talled			do	E	1 1	NA.	NA.	
	doo, Mexican		Hexico	40	Ę	3	NA	NA.	
	dog, Ulah	1 4 4	. U.S.A. (UT)(TU) A.B.U [1	8, 149	NA.	17 40(9)	
	om, peninsular		Maxico (Baja California)	do	Ę	- 10	NA.	I NA	
	orn, Sonoran		USA. (AZ), Mexico	da	€ ξ	1, 3	(NA	l MA	
	A		Southern South America	do	ĮΕ	15	NA.	NA NA	
	Costs Ricen		Niceragua, Panama, Costa Rica	do		15	NA.	NA NA	
			Australia	do		1 .	NA.) ñã	
	Arving		Japan (Flyukyu lalanda)	do			NA NA	NA NA	
	voicano		Maxico				I NA	I NA	
	lse water		Auskalia	40			,	NA NA	
	reano kangaroo	Dipodomys retratoides exilis] U.S.A. (CA)	40	1 _		,		
	prio Bay kangaroo							NA.	
	ck-past] Australia	00) NA	l NA	,
	ngaroo, brush-lailed	Bettonout peniculata	do	.) do) NA) NA	
	ngaroo, Gaimard's	Bettongia gairnardi		(b	} Ē		NA.	[NA	
	ngarge, Leaver's	Rattonoia lasueur	40	40			NA.	NA.	
	naeroo, piala	Caloprymnus campestris		do			NA.	NA.	
	ngaroo, Queensland	Bettongia tropica	,,[, do	do,			NA.	NA.	
	eros, black		Sub-Saharan Africa	do) NA	NA.	
	ceros, prest Indian		lode, Nepal	do		1 4	NA.	, WV	
	CAYDS, JAYAN		Indonesta, Indochina, Burma, Thaifand, Sikkim, Bangladesh, Malaysia.	do		3] NA	NA NA	
Rhinor	cerps, nombern white	Ceratotherium simum cottoni		do	٤	3	NA	NA NA	
Pitwood	cerps, Sumetran	Dicerorhinus (= Didermoceros) sumetren			. 6	1	NA NA	NA	
2.1.	Sheer and the desirement	Saiga tatarica mongolica			! ⊊	10	. I NA	NA.	

Seal, Hewailan monk		Caribbean See, Gull of Mexico	60	E E	233 3 1, 2, 46 18 3	AA AA AA AA	NA NA NA NA	U.S. Fish
	Capricornia sumatraensia		' do	Ë	3	NA Ì	NA	Ę

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7	AM	VN	91	3	ф	Cyprus	Ома тиятоп (= опелия) ортоп гмО	אושוואושי
602	VN	VN	E	3		Peru, Braza, Ecuador, Colombia, Venezu-	Caragao app. (all apacios)	Uakan (all species)
	VN	VN	E	Įā	ор	мівчегА	Liny Michigan Syrocophania	Juger, Teamanlan (-Thylacine)
	YN	٧N	9 €	ļá		Temperate and Tropical Asseminant	Penthole toline	
	(a)(b.\1)	ΥN	9L	í	ор	Physpinesseniggida	Tersus synchia	Tarset, Philippinennnn
				i	F	AleusanaV bna		
	VN	YN	E	3	ор	dox. Colombia, Ecuador and possibly Peru	enbaycod modej	Taps, mossilain
	AM	VΝ	£	э	ор	guay and Argantina Southern Maxico to Colombia and Ecua-	र्मिश्रम्य क्ष्मसम्	Tepu, Central AmericanTepu,
	YN	٧N	ε	3	ор	-ausig of rituos aleuxeneV bns aidmoloD.	รบุรคบุค รณช่อ]	Tapu, Brazilien
	YN	VN	Şί	3	ор	Burma, Laos, Cambodia, Vistnam, Malay- sla, indonesia, Thailand	Tapins infens	Tepx, Asten
	(2)OF (1	VΝ	91	۱ĩ		Colombia	Saguma hucopus	Termenin, whate-footedpeloot-environment
	YN	YN	91	Ιŝ			Saguma bicolor	Tamatin, pred
	***	'''	3"]		*	20024 110083	#91)*
		ļ		}	i		#poctal.	-omtafi noti-neblog = ;nnamaT bebaen
	YN .	YN	6	3			Me) .qqs (εωθούπου.) -> ευσκήκοιποθ.	-neblog=) begmm-neblog ,nasmaf
	VN	AN)	3		Physphoseseniqquid	Bubalus neinforents	WENTHER
	AN ;	VN	09	3	Ор	латО	Nountegus levent	Tahle, Arablan
	YN	VN	09	19	ор			Sunk, Zanasmadisnas, knus
	YN	VN	£	۱ã	ор	Kaahmir	Cervus eliphus hanghi	Slag, Kaahmit
	VN.	VN	č	13			Cervus elaphus berbarus	Slag, Barbary
	YN	VN	691	3	Entre			Squarel, Virginia northem flying
ö		1 "	00.] "	Sussex County).	(W W) VSU	Giaucomys sabinna hiscus	cold madica along langs
Ē	(#)>0 (L	ΨN	191	NX	-30) Y \$ G		,	
		i	99 L		Sussex Co., DE.	(Yd		
5	YN	VN	1,161,	3	โดยเล, except	lessifices of alusnined avaimled) A S U	этемия идок скижене	Squirrel, Delmarya Peninaula fox
_	YN	VN	SPL	3	ор	USA. (NG, 14)	стинсому выболния содочания	Squirel, Ceroline northern flying
	YN	YN	€	3	ор	Dominican Republic, Halb	enropened vopouelos	Solenodon, Hallan Solenodon,
ŗ,	YN	VN	6	3		Cuba	Colondon (Alopogale) cubanua	Solanodon, CubanSolanodon, Cuban
.g	VN	VN.	7 E	l a		Brazil	guegbha joidheine	Sloth, Brazillan three-toed
V1	VN	VN	1	ā	Ор	Майадазу Republic (= Мьфадаесаг)	Propinsors app. (eli epecies)	
Wildlife	YN	VN	SI	i i	ор	Malaysia, Indonésia	วิลัยบุคโลยนิยา syndaciylus	Gururis
Ξ	VN.	VN	518	1 i		ON W VS	Sover longworks fished	Sixew, Dismal Swamp southeastern
יש	VN.	YN	C	Ē		Tibet, Bhtsken	Cervus elephus wellichi	bous
Ξ	VN :	YN	91	Ĭ		Keehnik	O-18 หรือค่า เคยเล่น ราว	
5	VN	YN	É	Ιá			Feits serval constantina	Servel, Bathery
77	VN	YN	ğι	li		Esst Asla, Sumetra	Caphcoms aumananas	MOIS
and	VN	VN	έ	13		Bangladesh, Southeast Asia, India	goe denue	
0	***	· · ·	"	"		and Black Sea	•	
-5	YN	YN	E	9		Mediterrang, Northwell African Coast		Seal, Meditertanean monk
Ë	AN.	YN	91	13	ор	(H) Y S.U		
	AM	AN	1' 3' 12	Э	ор	Caribbean See, Gulf of Muxico	พลายเกาะ เกาะ เกาะ เกาะ เกาะ เกาะ เกาะ เกาะ	Seaf, Caribbean monk
ν,	YN.	VΝ	l c	3			Cintopoles sibiliseus	Sald, white-notedbeson-shife,
\Rightarrow	AH	AH	668	ja	ор	- v kx418	Cincpoles senanes selanas	Sald, southern beared

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<u> </u>	pecie•	Libraria ranna	Population where	Sta-	When	Critical	Special
Common name	Scientific name	Historic range	endangered or threatened	tus	Kaled	habilet	rule#
cuna		South America (Andes)	do	E	3	NA	NA
ole. Americana		U.S.A. (CA)		E	166	17 95(a)	NA
allaby, banded hare			do	E	4 }	NA	NA
atioby, brindled nell-falled	Onychogales treensis	do		E.	4 }	NA	NA
aliaby, crascent neil-telled	Onychogales lunals	do.,			4	NA.	NA
Haby, Parma			do		1	NA	NA
Haby, Western hare			to	E	4	NA.	NA
slieby, yellow-fooled rock	Petrogale xanthopus				8	NA.	NA
tele, blue		Oceanic		E	3	NA	NA
hale, bowhead		Oceanic (north letitudes only)		E	3	NA	NA
hale, finback		Oceanic	do	E	3	NA	NA
hale, gray	- L =	North Pacific Ocean: coastal and Bering	do	E	3	NA.	NA.
inia, Maj distribution		Sea					
hale, bumpback	Megaptera novarangkas	Oceanic	do	E	3	NA :	→ NA
hale, right			do	Ε	3	NA	NA.
hale, Sel			do	E	3	NA	NA
hale soors	Physeler calodon		do	, €	3	NA	NA
oil, gray		. Holarciic	. U S.A. (48	~ E	1, 5, 13,	17.05(a)	NA NA
00, 000			conterminous	١.,	15, 35	i i	
	}		States, except	} "		'	1
	i e		MN), Mexico-	1	•		
Do		dodo	USA (MN)	Ţ	35	17.95(a)	
loil, maned			Entre	E	4	NA.	NA
(4), [100 PF 2		QUAY.	1		l i	1	1
/oli. ted	Cania ruhra	USA (southeastern U.S.A. west to con-	do,	. E	1	NA.	NA.
WII, 1994		trat TX).		1	•	l	l
tombat belougoead I-Bernard's a	d Lasiorhinus kraftil (tormerly L. barnardi	Australia	do	E	4, 0	NA.	NA.
Queensland haky-nosed).	and L. gillespiel).	1	ì	1		}	Ì
/eodral, Key Largo		U.S.A. (FL)	,, ,do	ĮΕ	131E,	NA.	NA NA
today sal colo muminum			Į.	1	160	ļ	ļ
sk, wild	Bos grunniens	China (Tibet), India	do	E	3	NA.	NA.
bys, Grevy's			do	. τ	54	NA.	NA NA
ebra. Harimann's mountain		Namible, Angola	do) ī	54, 111	NA.	NA NA
ebra mountain	-, • · -	South Africa		E	15, 111	NA.) NA
U.S 157,000 100 11	2003 200		ì	ł	ł	[•	l
Bands		1	1	_			.
kepa, Hawaii (honeycreeper)	Laxoos coccineus coccineus	USA (HI)	da		2	NA.	NA.
teps Maus (honeycreeper)			60 [E) 2	NA.	NA.
kistos, Kausi (honeyomaper)			do	<u> </u>	1	NA.	NA.
kiepolaeu (honeycresper)				Į E	1	N.A	NA.
Ibaiross, short-leted		1	Entire, except	E	3	NA.	NA
na= ^44' 4: Mt. Isaa. A		LISA (AK CA RI OR WA).	USA	Į	ł	1	
ileabled selless shouldered	Agelakis xanthomus		Entre	, E	17	17.05(b)	l na

Brislabid, wastern	Western Pacific Ocean: U.S.A. (Guam) do E 158 NA Indian Ocean: Mauritius do E 3 NA Eastern Attente Ocean: Augustius Go E 3 NA	U.S. Fish C
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Dayonia bezopeirara kayvasina, data Ocean Chisinias island Dayonia brachonia siocisa Wasisian Pacific Ocean USA (Gaam) Hyaye faycove aproacas process Annas arrangana Gorous representational process Gorous representational proce
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	Species	Historic range	Vertebrata population where	Sta-	When	Critical	Special	:
Common name	Scientific name		endangered or threatened	tua	Nated	habitat	Unjes	
Eagle, bald	Haliaeetus leucocephalus	North America south to northern Mexico	U.S.A. (conterminous States, except	E	1, 34	NA	NA	
Do	40		WA. OR, MN. WI. MI). U.S.A. (WA. OR.	,	34	NA.	17.41(a)	
		ł	MN, WI, MI).	1	,	•		
Eagle, Greenland white-lailed		Greenland and adjacent Atlantic Islands			j 15 j	NA.	NA.	
Eagle, harpy		Mexico south to Argentine] 15	NA.	NA.	
Eagle, Philippine (≕monkey-eating)		Philippines			} 3	NA.	NA.	
Eagle, Spanish Imperial		. Spain, Morocco, Algeria			3	NA.	NA	
Egrat, Chkossa		China, Korea	do	{ E	} 3 '	NA 1	NA.	
Fatcon, American peregrine	Falco peregrinus anatum	Nests from central Alaska across north- central Canada to central Maxico, win- ters south to South America.	do	Ε	2, 3, 145	17.05(b)	NA	
Falcon, Arctic peregrine	Falco peregrinus tundrius	Nests from northern Alaske to Green- land; winters south to Central and South America	do	1	2, 3, 145	NA	NA	
Falcon, Eurasian peregine	Falco peregrinus peregrinus	Europe, Eurasia south to Africa and Mid- east	do	E	15	NA	NA	
Falcon, northern aptomado	Falco lumoralis septentilonalis	USA (AZ, NM, TX), Moxico, Guatemeta	do	ļΕ	216	NA	NA.	
Falcon, paregine		Worldwide, except Anterclics and most Pacific Islands	Wherever found in wild in the conterminous 48 States.	E(S/A)		NA	NA	
Finch, Laysan (honeycreeper)	Telespyza (= Psittirostra) cantans	. U.S A. (HI)	Entra	İΕ	1 11	NA.	NA	
Finch, Nihoa (honeycreeper)	Telespyza (= Psittirostra) uturna	do]do	E	1 1	NA.	NA.	. !
Flycatcher, Euler's	Empidonax autari johnstonat	. West Indies: Grenada	do	. €	3	NA	NA	
Flycatcher, Seychelles paradise	Terpsiphone corvina	Indian Ocean Seychelles		.) E) 3]	NA.	NA	. !
Flycalcher, Tabili		South Pacific Ocean: Tabili			1 3	NA.	NA	
Fody, Seychelles (weaver-linch)		Indian Ocean: Seychelles	do) E	} 3	NA	NA	
Fngatebird, Andrew's	Fregala andrewsi	East Indian Ocean	do	E	15	NA.	NA	•
Goose, Aleutian Canada				} E	1,3	NA.	NA.	. '
Goose, Hawakan (- nene)	Nesochen (= Branta) sandvicensis	USA (HI)		E	1 1	NA.	NA	
Goshawk, Christmas Island		indian Ocean: Christmas Island) Ē) 3	NA.	NA.	•
Grackle, stander-billed		Maxico] 3	NA	NA	
Grasswren, Eyrean (flycatcher)		Australia) 3	NA.	NA.	
Grebe, Alulan		Gualemala			3	NA	NA	
Greenshank, Nordmann's		U.S.S.R., Japan, south to Malaya, Bomeo			15	NA.	NA.	
Guan borned		Gualemala, Mexico			3	NA.	NA.	
Guit, Audouin's		Mediamanean Sea				NA.	NA	
	Larus relicius					NA.	NA.	
Mark Antoing island spectrus	Accipiter francesii pusilius	Inden Dones Compo Islanda	do			NA		

Honeyeater, helmeled tols, Japanese crested...

Rhinoplax vigil Meliphaga cassidix...

Ecuador (Galapagos Islanda) U.S.A. (HI)..... Brazil . ..do . USA (HI) Thelland, Malaysia Australia . da. ... do... China, Japan, U.S.S.R., Korea............ South Pacific Ocean' New Caladonia

NA NA NA NA NA NA HA HA HA HA NA HA 1 15 1 15 4 3 FEEFF Fish ar

Hawk, Galapagoa		Ecuador (Gatapagos Islands)		€ļ	3. {	NA (MA
Hawk, Hawalian (= lo)	Buteo solitarkie	USA (N),		£	1	NA	NA
Hermit, hook-billed (hummingbird)	Glaucia (= Ramphodon) dohrnil	Brazil		E	15	NA	NΑ
Honeycreeper, created (akohekohe)	Palmena dolel	USA (Hi)		Εį	1	NA	NA
Hornbill, helmeted	Rhinoplex vigil	Thalland, Malaysia		Εļ	15	NA .	NA
Honeyeater, hetmeted	Heliphaga cassida	Auskaha		ΕĮ	4 {	NA (NA
Ibia, Japanese crested	Nipponia nippon	China; Japan, U.S.S.R., Kotea		εį	3 }	NA .	NA
Kagu	Rhynochelos jubalus	South Pacific Ocean: New Catedonia		εį	3	NA	N۸
Kakapo (-owi-parrol)	Sungops habropulus	New Zealand	do,	E	3	NA	NA
Kestrel, Mauribus	Falco punctatus	Indian Ocean. Maumkia	do	E	3	NA	NA
Kestrel, Seychelles	Falco arasa	Indian Ocean: Seycheltes Islands	do	E	3	NA	NA
Kinglisher, Guern Micropesian	Haloyon connamomina cinnamomina	Wastern Pacific Ocean: U.S.A. (Guam)	do	E }	156	NA	NA
Kita, Cuba hook-billed,	Chonorphierax uncinatus wisonii,	West Indies: Cubs		E	3	NA	NA
Kite. Everplade anali	Rostrhamus sociabilis plumbaus	USA (FL), Cuba	Florida	E)	1	17 95(b)	NΑ
Kite, Grenada hook-buled	Chondrohierax uncinatus minus	West Indies. Grenada		ΕÌ	3	NA.	NA
Kokako (wattlebud)		New Zealand	do.,,,,	ĒΙ	3	NA	NA
Macaw, glaucous		Paraguay, Uroquay, Brazil		Ë	16	NA.	NA
Macaw, Indigo	Anodorhynchus lasti	Brezi		Ē	15	NA	NA
Macaw, kitle blue	Cyanopsitla spixa	40	do	E	15	NA	NA
Magple-robin, Saycheltes (thrush)		Indian Ocean: Seychottes Islands		ĒΙ	a l	NA	NA
Malkoha, red-faced (cuckoo)	Phienicopheeus pyrrhocephelus	Srl Lanka (Ceylon)	do	٤Ì	aí	NA I	NA
Mallard, Mariana		West Pacific Ocean: U.S.A. (Guarn, Mari-	do	ĒΙ	23	NA	NA
		ana Islands).		-]			•••
Megapode, Micronesian (-La Perouse's)	Mogapodus laperouse	West Pacific Ocean: U.S.A. (Palau Island,	do	εl	a Í	NA	NA
	· •	Manana Islanda).		1			
Megapode, Majeo	Macrocephalon maleo	Indonesia (Celebea)	do	Εl	3	NA	NA
		USA (HI)	do	Εļ	11	NA	NA
Monarch, Timen (old world flycalcher)	Monerche lekalsukesee	Western Pacific Ocean: U.S.A. (Mariana	do	ΕÌ	a i	NA I	NA
		febrafal		_ i	· I		•-
Moorhen (=gallinule), Hawaiian common	Gallinula chloropus sandvicensis	USA (HI)	do	Εİ	1	NA.	NA
Moorhen (-gallingle), Mariana common	Gallinula chloropus guarni	Western Pacific Ocean: U.S.A. (Guam.	40	ĒΙ	158	NA	NA
		Tinian, Saipan, Pegan).		- [[•••
Nightler (=whip-poor-will), Puerto Rico	Capamulgus nociriherus.	USA (PR)		Εİ	6	NA	NA
Nulupu'u (honeycreeper)	Hemignathus kicidus	U.S.A. (HI)	do	E	1, 2	NA	NA
	Moho braccatus		do	Ë	- "il	NA	NA
Ostrich, Arabian	Suvituo camelus syriacus	Jorden, Saudi Arabia	do	Ē	3	NA	NA
	Sirutho camelus spatzi	Spanish Sahera		Ē١	āi	NA	NA
'O'u (honeycreeper)	Psitrosus psittaces	USA (HI)	do	ĒÌ	- 1	NA	NA
Owl. Anjouan acops	Otus rulius capnodes		do	E	á	NA	NA
Owl, plant scops		Philippines: Marinduque and Mindanao		Εİ	15	NA	NA
-	-	laland		1	!	1	
Owl, Seychelies	Otus Insularis	Indian Ocean: Seychelles (stands	do	E	3 [NA	NA
Owlet, Morden's (-Sokoke)	Otus kenose	Kenya		E	3 \	NA.	NA
Pakia (honeycreeper)		USA (HI)		6	11	17 95(b)	NA
Parakeet, Forbes		New Zealand		Ë	3, 15	NA	NA
Parakeet, golden		Brazu		ΕÌ	4	NA	NA
		Australia		E	a l	NA I	NA.
		Indian Ocean; Maurilles		ĒÌ	3 1	NA.	NA

Accorder transceal publica

Hawk, Anjouan Island aparrow.

P 40/1

MA AM

NA

Common name	Cles Scientific name	Historic range	Vertebrate population where endangered or threatened	Sta- tus	When listed	Critical habitat	Special rules
Parakeel Ochre-marked	Pyritiura cruentata	Brazi.	do	Ε	3	NA NA	N.
Parakest, orange-belijed	Neophama chrysogaster	Australia	do	E	4	NA:	N.
erakest, paradise (= beautiful)	Psephotus pulchernmus	do	do,	E	4	- NA	N.
srakest, scarlet-chested (= splendid)	Necohema splendida	do	do	E	4	Î NA	N/
arakeel, turquoise	Naophema puichella	do	do	E] 3	NA	· N.
erol Australian	Geoositecus occidentalis		do	. 6	1 3	NA '	N.
arrot, Bahaman or Cuban	Amazona leucocophaia	Wost Indies: Cube, Sahamas, Caymana	do	E	3, 15	NA.	N
Periol, ground	Pozoporus wallicus	Australia	do	Ē	6	NA.	N.
arrol imperial	I commercial control of the control	West Indies: Dominics	do	Ë	l a	NA :	Ñ.
ariot, Puerto Rican	Amerone vittele	U.S.A. (PR)	00	Ë	i	NA NA	N.
arrot, red browed		Brazk	do	E	أ أ	NA.	N.
Parrol, red Gepped		do	do	Ë	15	NA I	N.
and, red-necked		West Indies: Dominics		l F	50	l NA	N.
arrol, red-spectacled	Amazona pretrei pretrei	Brazil, Argentina	do	E	15	NA.	N.
	Amazona versicolor	West Indies: St. Lucia	60	Ē	3	NA.	N
Parrol, St. Lucia	Amezona guidogi		do	Ē	3	NA I	N.
errot, St. Vincent	Rhynchopsitta pachyrhyncha	Mexico, U.S.A. (AZ, NM)	Mexico	Ē] 3	NA	N
Perrot, thick-billed		Brazil	Entre	E	15	NA I	N
errot, vinaceous-breasted	Amazona vinacea	USA (HI)	do	Ë	'	l IXA	N.
arrotbili, Maul (honeycreeper)	Pseudonestor xanthophrys	USA (Carolinas to TX, CA), West indies,	Entire, except U.S.	ءَ ا	2. 3. 171	NA NA	N.
Pelican, brown	Pelecunus occidentalis	G. and S. America: Coastal.	Allantic coast,	=	2, 3, 171	^^	1
enguin, Galapagos	Soheniscus mendiculus	Ecuador (Gelepagos Islands)		Ε	3	NA.	N-
etrel Hawallan dark-rumped	Prerodroma phaeopygia sandwichensis	U.S.A. (HI)		E	2, 4, 1	l NA	N.
hossant, bar-talled		Burne, Chine	do	E	a	NA	N
	1 25	Burna, China, India	do	ΙĚ	آ آ	NA NA	N.
heasant, Blyth's tragopan	,	Chka			1 3	NA NA	N.
heasani, brown eared		do		Ē	1 3	NA.	l ii
heasant, Cabot's trapopan			do	Ė	3	NA:	Ň
heasant, Chinese monal	Lophophorus Ihuysii		do	Ę	1 3	NA.	N
heasant, Edward's	Lophura edwardsi	Visinam		Ē	15	NA NA	N.
heasani, Elliot's	Symmeticus ellioti	China			3	NA NA	N N
hessent, Imperial		Vietnam	do	E		****	N
heasani, Mikado		Telwen	do	🖺	3	NA 1	
theasant, Palawan peacock		Philippines	do	E	3	NA:	N
treasant, Sciater's monal		Burma, China, India		E	3	NA.	N
heasant, Swinhoe's		Talwas	do	E	3	NA.	N
heasant, western tragopan	Tragopan melanocephalus	India, Pakistan	00	ε	3	NA.	N
heasant, white eared	Crossoptilon crossoptilon	China (Tibet), India	do	E	4	NA.	N
igeon, Azores wood	Cotumba palumbus azorica	East Atlantic Ocean: Azores		E	3	NA.	N
Pigeon, Chatham Island	Hemiphaga novaeseelahdiae chatham- ensis	New Zealand	do	E	3	l MA	N.
tgeon, Mindoro zone-talled	Ducula mindonensis	Philippines	do	Ε	16	NA.	N
Inson Piedo Rican plain	Columba inornata wetmorel	U.S.A. (PR)	do	Ē	2	NA.	l N
Many I rated the library recommendation	A PARTICULAR MEDICAL STREET, AS A SALEMAN AND PROPERTY OF THE			_	_		

Argentine.

15 15 3 NA NA NA NA NA

Pitta koch

Thinomis novasseslandiae... Charadrius melodus.....

	plantaga <u>i, pagaga</u> palamaga marila — o popo menter e e e e e e e e e e e e e e e e e e		•		-		
Piping-guan, black-fronted		Argentina		E [16	NA I	NA
Pilla, Koch's	Pina koci	Phikppines	do,	ε	15	NA	NA
	Thinornia novaeseelandiae	New Zealand		E	3	NA	NA
Piover, piping	Charadrius melodus	U.S.A. (Great Lakes, northern Great Plains, Atlantic and Gull coasts, PR VI), Canada, Mexico, Bahamas, West	watershed in	£	211	NA	NA
		indies.	Mi, MN, NY, OH, PA, and WI and Province of				
			Ontario.	- 1		1	
Do	💋	do	Entire, except	т	211	NA	NA
			those areas where listed as endangured above.	'	511	"^	INA
Po'ouli (honeycreeper)	Melamprosops phaeosoma	USA. (HI)	Entire	Æ	10	l na l	NА
	Tympanuchus cupido attivateri	USA (TX)		Ē	1	l NA	NA NA
	Cyrlonyx montezumae memanu	Maxco (Vera Cruz)	do	Ē	15	NA.	NA
	Pharomachrus mockno	Mexico to Panama		Ě	15	NA I	NA.
	Rulius pectoralis mueller	Now Zealand	do	ĒΙ	3	NA	NA NA
	Rallus longirostas obsoletus		do	Ē	2		
	Railus owstoni	Western Pacific Ocean: U.S.A. (Guam)	40	E	146E, 156	NA NA	NA NA
Rall, fight-looted clapper	Relius longirostris levipes	U.S.A. (CA), Mexico (Baja California)	do	E	150	NA	NA
Rait, Lord Howe wood	Inchokmnas sylvastris	Australia (Lord Howe Island)		Ē	15	NA	NA.
	Rallus longirostris yumanensis	Mexico, U.S.A. (AZ, CA)		£	1	NA	NA.
	Prerocnemia pennata	Argentine, Bolivia, Peru, Uruguay		Ē	á	NA	NA.
	Petroica traversi	New Zealand		έl	3	NA.	AN.
	Patroica multicolor multicolor	Australia (Norfolk Island)		Ě	2	NA I	NA.
	Picatheries ordes	Cameroon, Gabon		Ē	3	NA.	NA NA
	Picathartes gymnocephalus	Africa: Togo to Sierra Leone		Ē	3	NA NA	
	Uratelomis chimaora	Malagasy Republic (-Madagascar)		Ē	3	NA NA	NA
	Atrichomis clamosus	Austraka	do	Ē	3		NA
	Copsychus niger cebuensis		do	Ē	3	NA	NA
	Puttinus suricularis (formerly puttinus)	USA. (HI)	40	ī	10	NA NA	NA NA
	Lanus kdovicienus meamsi	บรล. (CA)		E	28	NA	NA
	Cardvelis (= Spinus) cuculleta	South America	do	Ē	15	NA I	NA.
	Ammodramus (= Ammospica) maniimus mirabilis.		do	Ē	ĭ	17.95(b)	NA.
	Ammodramus (=Ammospiza) muntimus nigrescens.	do	00	ε	١	17.95(b)	NA
	Ammodramus savannanım floridanus	. 60	do	εl	239	NA	NA
	Amphispiza bulk clementage		do	Τ̈́	28	NA I	NA.
	Aplonis petreini		do	E	3	NA	NA

Common name	cles Scientific name	Historic range	Vertebrate population where endangered or threatened	Sta- tue	When	Critical habitat	Special rules
Stift, Hawaiian (=Ae'o)	Himantopus mexicanus («himantopus)	U.S.A. (Hawali)	do	E	2	NA	NA
Stock, oriental white	Ciconia ciconia boyclana	China, Japan, Korea, U.S.S.R		E	3	NA	NA.
Stork, wood	Myctone americane	U.S.A., (CA, AZ, TX, to Carolinas), Mexico, Central and South America.	SC).	E	142	NA	NA.
Swittel, Madana gray (=Vankoto)	bartschi.	Western Pacific Ocean: U.S.A. (Guam, Rota, Tinian, Salpan, Agiguan).	Entre	E	158	NA :	NA.
Teal, Campbell Island liightless	Anas aucklandica nesiotis	New Zealand (Campbell Island)	do	ε	15	NA:	NA.
Tern, California least	Stema antillarum (= albifrons) browni	Mexico, U.S.A. (CA)		E	2, 3	NA	NA.
Tern, least	Stema anillatum	U.S.A. (Attantic and Guil coasts, Miss. R. Basin, CA), Gr. and Lesser Antilles, Bahamas, Mexico; winters G. America, northern S. America.	U.S.A. (AR, CO, IA, IL, IN, KS, KY, IA, (Mas. R. and bibs. N of Baton Rouge), MS (Miss. R), MO, MT, NE, NM, ND, OK, SD, TN, TX (Except within 50 miles of cosst).	£	182 (NA .	, NA
Thresher, white breasted	Ramphocinclus brachyurus	West Indies: St. Lucia, Martinique		E	3	NA	NA.
Thrush, large Kaual	Myadostes (=Phaeomis) myadestinus	U.S.A. (HI)		E	2	NA	NA
Thrush, Molokal (=oloma'o)	Myadestes (=Phaeomis) lanalensis (=obscurus) rutha		do	E	2	NA	NA
Thrush, New Zealand (wattlebtrd)	Turnagra capensis	New Zealand			3	NA :	NA.
Thrush, small Kausi (~pusiohi)	Myadestes (= Phaeomis) palmed	U.S.A. (HI)		[E	1	NA	NA.
Tinamou, solilary	Tinamus solitarkis	Brazil, Paraguay, Argentina			15	NA :	NA.
Trembler, Martinique (thresher)	Cinclocerthia rulicauda gutturalis	West Indies: Martinique		E	3	NA	NA
Viveo, least Bell's	Vireo bellit pusillus	U.S.A. (CA), Mexico		E	228	NA.	NA.
Wanderer, piain (collared-hem/pode)	Pedionomous torquatus	Australia		E	9	NA	NA
Werbler (wood), Bachman's	Vermivora bachmanii	U.S.A.(Southeastern), Cuba		E	1,3	NA	NA
Warbler (wood), Barbadon yellow	Dendroice petechia petechia	West Indies: Berbados		E	3	NA.	NA NA
Warbler (wood), Kirtland's	Dendroice kirtlandii	USA. (principally MI), Canada, West Indies: Bahama Islands.	•		1, 3	NA	NA
Warbler (willow), rightingale read	•	Western Pacific Ocean	Islands).	E	3, 4	NA	NA
Werbler (willow), Rodriques	Bebromis rodericanus	Maurities (Rodrigues Islands)		E	3	NA.	NA.
Warbler (wood), Semper's	Leucopeza semperi	West Indies: St. Lucis			3	NA.	NA.
Warbler (willow), Seychelles	Bebromis sechellensis	Indian Ocean: Seychelles Island		E	3	NA	NA
Whyblid, Western		Australia		Ε	3	NA.	NA
White-eye, bridled	Zusterops conspicitata conspicitata	Western Pacific Ocean: U.S.A. (Guam)	,óò	E	156	ΝA	NA.
White-eye, Norfolk Island	Zoslerops albogularis	Indian Ocean: Norfolk Islands		E	15	NA.	NA.
White-eye, Ponape greater	Rukia longirostra (=sanford)	West Pacific Ocean: U.S.A. (Caroline le- lande).	60	E	3	NA	NA

Indian Ocean; Seychellas

Mexico

U.S.A. (southcentral and southeastern),

Cutia.

USA (southcentral and southeastern).

West Indies: Guadeloupe. West Indies: St. Lucie

Korea

HA HA NA

NA NA NA NA

1, 3

HA HA HA

> AM AM AM AM

17-186 OLS

White eye, Seychelles..... Woodpincker, Imperlat..... Woodpecker, Ivory-billed...

Woodpecker, red-cockaded... Woodpecker, Tristam's Wren, Guadaloupe house...... Wren, St. Lucia house......

AFPTHER

Zosterops modesta Campephilus Imperialis... Campephilus principalis...

Proofdes (= Dendrocopos) borealis ... Dryocopus javensis richardsi......

Troglodytes aedon guadeloupensis ... Troglodytes aedon mescleucus......

	terbus and Company lies	Zosterops modesta	Indian Ocean: Seychelles	do) E	1 3	I AN	NA.
	White-eye, Saychelies		Менсо	do	ΙĒ	3	NA.	NA.
		Campephilus principalis	USA. (southcentral and southeastern).	do	ΙĒ	1.3	NA.	NA.
	Woodpecker, Nory-billed	Cempopianos precipaes	Cuba.		-	"."	,,,,	••••
	Woodpacker, red-cockeded	Picoides (= Dendrocopos) boreste	USA (southcentral and southeastern)	do		2	NA.	NA
	Woodpacker, Tristam's	Doyocopus javensis nchardsi	Korua	do		3	NA.	NA.
	Wren, Guadeloupe house	. Troglodytes sedon guadaloupensis	West Indies: Guadeloupe	do	E	3	NA.	NA NA
	Wien, St. Lucia house	. Troglodytes sedon muscleucus	West Indies: St Lucia	do	ĮΕ	3	NA	N/
	HEPTILES							
	Aligator, American	Alboator mississippiensis	Southeasiern U.S.A	Wherever found in	l e	1, 11, 51,	NA.	NA.
	toland to the termination of the second		\	wild except those	1	60, 113,		
		<u> </u>	ł.	Aldus where	1	134, 186	i i	
		į.	l.	kstud 88	1	•]	
		1 .	1	the as benefacit	1		[
		1 '	1	furth below	Į.			
	Do			U.S.A. (certain	ī	20, 47,	NA.	17 42(4)
			1	break of GA and		51, 60,		, ,
		l	l	SC, as eat forth	l	134, 186	[
			1	in 17.42(a)(1)).	1]	
			l .	17.42(b)(1)).				
	Do.,	do	60	U.S.A. (FL, LA and	T(S/A)	11, 47,	NA	17.42(8
		}	i e	TX); in captivity		51, 60,		
		<u> </u>		wherever found	1	113, 134,		
$\tilde{\mathbf{z}}$					ĺ	186		
_	Alligator, Chinese	Aligator sinensis	China	Entire		15	NA	NA.
	Anole, Culebra Island grant	Anoks roosevelti	USA. (PR: Colebra Island)		.) E	25	17.95(c)	NA
	Boa, Jamakan		. Januica		. E	3	NA	NA.
	Bos, Mons	Epicrates monensis monensis			, T	33	17 85(c)	NA
	Bos, Puerto Rico	Epicrates inornatus			. Ε	2	NA	NA
	Boa, flound laland (no common name)		Indan Ocean: Mauritus		E	88	NA	NA
	Boa, Round Island (no common name)	Bolyena multocarinata	do		E	88	NA	NA NA
	Bos, Virgin Islands tree	Epicrates monensis granti	U.S. and British Virgin Islands		E	2, 86	NA	NA
	Calman, Apapona River	Caiman crocodius apaponunsis			٤١	15	NA	NA
	Carnan, black	Melanosuchus mour	Amazon basin	do.,	E	15	NA	NA
	Caiman, broad-anouted	Celman letrostris	Brazil, Argentina, Paraguay, Uruguay		. 6	15	NA	NA
	Caimen, Yacare	Ceiman crocodius yacare			E	3	NA	NA
	Chuckwalla, San Esteban Island	Sauromakis varus			E	68	NA	NA
	Crocodile, African dwarf	Osleolaemus teluspis teluspis	West Africa		E	15	NA .	NA
	Crocodile, African stander-snouted	Crocodylus Calaphractus	Western and central Africa	do	٤		NA NA	NA.
	Crocodile, American	Crocodykis aculus	U.S.A. (Ft.), Mexico, South America, Con- tral America, Caribbean.		E	10, 87	17-95(c)	NA
	Crocodile, Ceylon mugger	Crocodykis pakains kimbula	Sri Lanka	do.,	ε	15	NA	NA
	Crocodie, Congo dwarf	Osteolaamus tetraspis paborni	Congo River drainage	do.,	E	15	NA.	NA
	Crocodile, Cuben	Crocodylus rhombiler	Cuba	do	É	3	NA	NA
	Crocodie, Morelet's	Crocodylus morelute	Mexico, Baliza, Gustemala	ຜ	Εl	3	NA.	NA
	Crocodie, mugger	Crocodykia pakistria pakistria	India, Pakistan, Iran, Bangladesh		ĒΙ	15	NA	NA

Pacific Ocuen: U.S.7. [Carolina la-

White-eye, Noricik lessnd...... White-eye, Ponspe greater.... tion)

Species

Vertebrate

Common name	Scientific name	Historic range	population where endangered or threatened	Status	When	Critical habitat	Special rules	7.11
Crocodila, Orinoco	Crocodylus intermedius			E.	3	NA	NA	
Crocodite, Philippine				E	15	NA	NA	
Crocodie, saliwater (=estuarine)	Crocodylus porosus	Southeast Asia, Australia, Papua-New Guinea, Pacific Islands.	Enike, except Papus-New Guines	E	67	NA	NA	
Crocodile, Slamese	Crocodylus slamensis	Southeast Asia, Malay Peninsula	Entire	E	15	NA	NA.	
Gavial (~ gharial)		Pakistan, Burma, Bangladesh, India, Nepal.		E	3, 15	NA	NA	
Gecko, day				Ε	3	NA	ŅĀ	
Gecko, Monito	Sphaerodectylus micropithecus	USA (PR)		E	125	17 95(c)	NA	
Gecko, Round Island day	Phalsuma guentheri	Indian Ocean: Maurillus		ĮΕ	3	NA	NA.	
Gecko, Serpent Island	Cyrtodactylus serpensinsula	do	do	1	129	NA	NA.	
Iguana, Acklina ground	Cyclura nieyi nuchalis			T	129	NA	NA	
iguana, Allen's Cay	Cyclura cyclilura inomata			T	129	NA.	NA	
Iguana, Andros Island ground	Cyclura cychlura cychlura		do	T	129	NA.	NA.	
Iguana, Anegada ground		gede Island).		E	3	NA.	NA	
Iguana, Barrington land				Į €	3	NA	NA	
Iguana, Cayman Brac ground		West Indies: Cayman Islands	do	ļΤ	129	NA	NA.	
Iguana, Cuban ground		Cuba	population Introduced in Puerto Rico).	'	129	NA	NA.	
Iguana, Exuma Island	Cyckra cychlura figginsi] West indies: Bahamas		T i	129	NA.	NA	
iguans, Fiji banded	Brachylophus fasciatus	Pacific: Flit, Tongs		€	86	NA	NA	ប្រ
Iguana, Fili crested	Brachylophus vitiensis	Pacific: Fiji		E	88	NA	NA.	0
iguana, Grand Cayman ground				[€.	129	NA	NA	Ω
Iguene, Jamaican	Cychira collei	Wast Indies: Jamaxa		٤١	129	NA:	NA.	Ħ
iguena, Mayeguana	Cyclure carinate bartschi	West Indies: Behemas		T	129	NA.	NA.	
Iguana, Mona ground	Cyclura stejnegeri			T	. 33	17.95(c)	NA.	ξ
Iguaria. Turks and Calcos	Cyclura carinata carinata	West indies: Turks and Calcos Islands	do	T	129	NA	NA.	•
Iguana, Watking Island ground	Cychra nieyi nieyi	West Indies: Bahamas	do	E	120	NA	NA	
guana, White Cay ground	Cyclura nieył cristata	do	do	T	129	NA.	NA	\Box
Lizard, blunt-nosed leopard	Gambelia (= Crotaphytus) situs	U.S.A. (CA)		E'	1	NA NA	NA.	ᅙ
Lizerd, Coschelle Vattey fringe-toed	Uma inomata			T	105	17.95(c)	NA.	
Lizerd, Hierro glent	Gallolla simonyi simonyi	Spain (Canary telands)		E	144	NA	NA	싫
Lizerd, Ibiza wali	Podarda pilyusansia				144	NA	NA	
Lizerd, Island night				Ţ	26	NA.	NA	
Lizard, St. Croix ground				€ '	24	17.95(c)	NA.	Ω.
Monitor, Bengal	Varanus bengalensis	iran, kaq, India, Sri Lanka, Malaysia, Af- ghanistan, Burma, Vietnam, Thailand.	do	E	15	NA.	NA.	Editio

90

Monitor, desert.....

North Airica to Neareast, Caspian Sea through U.S.S.R. to Pakistan, North-

E | 15

NA

ia C

	Monitor, desert	Varanue griesus	North Africa to Nearwast, Caspian Sea Brough USSR. to Pakietan, North-	49	E	15	АИ	NA	J.S.
	Monitor, Komoda Island	Varenus komodovasva	west India: Indonesia (Komodo, Filnija, Padar, and westurn Florus Island).	do	E	15	NA	NA.]; S:
	Monitor, yallow	Varanue flavescens	West Pakistan through India to Bangla- desh.	.do	E	15	NA	NA	۵
	Python, Indian	Python molurus molurus	Sri Lanka and India	00	ΕĹ	15	NA.	NA	ä
	Pattiesnake, Aruba Island.	Grotalus unicolor	Aruba laland (Netherland Antilies)	do	7 !	129	NA .	NA	_
	Rattlesnaks, New Mexican ridge-nosed	Crotakes willards obscurus	U.S.A. (NM), Mexico	do	T 1	43	17 95(c)	NA	Ξ.
	Skink, Round Island	Leiolopisme tellaid	Indian Ocean: Maurillus		7	129	NA	NA	Wildlife
	Snake, Atlantic salt march	Nerodia lesciale teoriale	U.S.A. (FL)		Τĺ	30	NA	NA	<u>#</u>
	Snake, Concho water	Narodia harteri pauckrisculata	USA. (TX)		τ1	241	NA	NA	Ξ;
	Snake, eastern indigo	Drymerchon coreis couperi	U.S.A. (AL, FL, GA, MS, SC)		τ Ι	32	NA.	NA	-
		Thermophia sirtalis tetrataenia	U.S.A. (CA)		Εļ	1	NA.	NA	Ň
	Snaka, San Francisco garter	Podocnemis expense	South America: Orinoco and Amazon	nio.	ĒÌ	a i	NA.	NA	2
	Teneroga	1	River basins.	1	- (_	,,,,		<
	Terrapin, river (=Tuntong)	Batagur baska	Melaysia, Bangladesh, Borma, India, In- donesia	{	E	3	NA.	NA	<u>.</u>
	Tomstoms	Tomstoms schlopeld	Malayais, Indonesia		E	15	NA.	NA	=
	Tortolse, angulated	Geographica yriphora	Malagasy Republic (-Madegascar)	do	ΕÌ	15	NA.	NA	Ž.
	Tortolse, Bolson	Gophania flavomarginatus	Mexico		Εļ	46	NA I	NA	0
	Torloise, desert	Xerobates (Scaptochelys, Gopharus)	U.S.A. (UT, AZ, CA, NV); Mexico		Ti	103	17 B5(c)	NA.	~
	in many and a second	agassitid.	(, , , , , , , , , , , , , , , , , , ,	Usah.	_ [
>	Tortoise, Gatapagos	Geochelone elephantopus	Ecuador (Galapagos Islands)		ΕÌ	3	NA :	NA	
4	Tortolse, rediated	Geochelone (= Testudo) radela	Malagasy Republic (≕Madagascar)	00	€∤	3	NA NA	NA	
	Tracaja	Podocnemis untiks	River basins.		E	3	NA	HA	
	Tualara	Sphenodon punctatus	New Zealand	do	E	3	NA	NA	
	Turtle, equatic box	Terrapane coshulls	Moxico	do	ΕĮ	6	NA.	NA	
	Turtle, black softshed	Trionyx rignicans	Bangiadesh	do	ΕÌ	15	NA	NA	
	Turtle, Burmese peacock	Morania ocellata	Burna		E	15	NA	NA	
	Turtle, Central American river	Dermalarnys mawii		do	E	129	NA.	NA	
	Turtle, Custro Cienegas softshell	Trionyx Alex	Mexico	do	εi	15	NA.	NA	
	Turtie, geometric	Psammobates geometricus (=Geoche-	South Africa		E	15	NA	NA	
	Todle some and	lone geometrica). Chulonia mydes	Circumplobal in tropical and temperate	Wherever found	۲ ا	2, 42	NA.	17.42(b)	
	Turile, green sea	LANCON MYCHALLER MANAGEMENT CONTROL	seas and oceans.	except where	٠,١	-, 1-		and Parts	
		[listed as	- 1			220 and	
			i '	andangered	1			227.	
		}	}	balow.	- 1				
	0.	40	do	Bresding colony	ΕÌ	2, 42	NA.	NA	
	Do.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		quantity and and any particular and a second state of the second	populations in Fi.	-1	4, 76			
		(i	and on Pacific	ſ		1		
		1	Į	coast of Maxico.	į,				
	Tal to lable and some t	Eretmochelys enbricate	Troolcal seas	Entire	ΕÌ	3	17.05(c)	NA	LOD
	Turtle, hawkebill ean (-cerey)			do	Ĕ	15	NA.	NA.	
	Turtle, Indian sawback	Тпопух дводивсив			Ē]	16	NA.	NA NA	17
	THE PROPERTY OF THE PROPERTY O	з <i>секлух ужуроск</i> а-полоненовыныныны	f Terresett Research	F		,,,,		117	

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THE PARTY NAMED IN

Toad, Wyoming ... Bulo hemiophys baxteri USA (WY) 136 FISHER Ain Balik (trout) . .. Salmo platycephalus Ayumodoki (loach).... Hymenophysa (- Botla) curta... Jenan. Bindost, Mexican (cathah)... Proteila phreatophia... Maxico Bonytoungue, Asian Schonnages from: 4

Total, Wyoming	Buto hemiophrys bexter!	U S.A. (WY)					1
FARES		L	-	_			١
lie Balk (trout)				E	3	NA	NA.
\yumodok! (loach)		1		E	3	NA	NA.
ikndcat, Mexican (cathah)	Prietella phreatophila			E	3	NA	NA.
lonytoungue, Aslan	Scieropages formosus			E	15	NA.	NA.
athah (no common name)	Pangasius sanitwongsul	Incland		E	3	NA	NA.
atish, gunt		40		£	3	NA	NA
attab. Yagui		U.S A. (AZ), Mexico		ĭ	157	17.95(e)	17.44(0)
avelish, Alabama		U S A. (AL)	60.,	T	28	17.95(a)	NA
avefish, Ozark,			do	T	164	NÁ	L NA
hub, bonyles			60,	E	92	NA	NA.
Zhub. Borax Lake				E	124	17 95(u)	l na
hub, Chihushus				T	132	NA	17.44(0)
hub, humpback				Ě	1	NA.	NA.
		- 1		T	174	NA.	17.44(0
Chub, Hutton tul	••••••			Ė	2	NA.	NA.
hub, Mohave tul				Ē	195	17.05(e)	l NA
hub, Owens tul				Ē	1	NA NA	l NA
hub, Pahranagat roundtait					2		
hub, alender				Ĭ	28	17 95(6)	17.44(c)
Cheb, Sonors				Ī	227	17 95(e)	17.44(0)
hub, spotin				Ţ	28	17.95(e)	17.44(c)
hub, Yaqui	Gila purpures	U.S.A. (AZ), Mexico		E	157	17.85(e)	NA NA
Cak (mknow)		Turkey		E	3	NA.	NA NA
ان-لا		U.S A. (NV)	do	ε	1	NA.	NA
Dace, Ash Meadows speckled				E	117E,	17.95(e)	l NA
					127E.		l
					130	l	l
Dace, desert	Eremichthys acros	do	do	1	210	17.95(e)	17 44(m)
lace, Foskett speckled				Ì	174	NA	17 440)
				Ė	2	NA	NA.
ace, Kendali Warm Springs	**************************************	1 1 .*		Ē	ī	NA.	NA.
ace, Mosps		U.D.A. (NY)		E	198	17.95(a)	NA.
				7			
)arter, bayou					10	NA.	17.44(b)
arter, fountain] U.S.A. (TX)	do	Ē	2	17.95(a)	NA.
Parter, leopard	Percina pantherina	U.S.A. (AR, OX)	do	Ţ	31	17.95(e)	17.44(d)
Darter, Maryland	Etheostoma sellere			£	1	NA	NA.
Darter, Nuangua	Etheostome nienpuse	U.S.A. (MO)		T	165	17 05(e)	17 44(k)
Darler, Okaloosa	Etheostoma okaloosae	US.A. (FL)		E	8	NA	NA.
Sarter, slackwater		U.S.A. (AL, TN)	do	T	28	17 95(e)	17 44(c)
arter, anali			do	T T	12, 150	NA	NA.
arier, watercress				E	2	NA	NA.
Sembusia, Big Bend				Ë	1	NA	NA.
Sambusia, Clear Creek				Ē	i	NA	NA.
Sambusia, Amatad				Ē	93	NA	NA.
Sambusia, Pecos				Ē	2	NA	NA.
441 100015, FOUI	Gambusia paorpal			Ē	98	17 95(0)	

Woundfin	Plagopierus argendiskmus	U S A. (AZ, NV, UT)	Entire, except Gita R. drainage, AZ, NM.	e \	2, 193	MA	HA	l S.II
Do	50	do	Glia R. draktage AZ, NM.	ΧN	193	NA	17.84(b)	715 h a
Soul that anked these toolhad	Succines chittenengoonels	USA (NY)USA (WV)	NA	Ţ	41	NA NA	NA.	ā.

Do	Succines chittenangoensis Triodopsis platysayoides Discus maccinitocki Papustyla pulcharrina	U.S.A. (NY)	Gila R. drainage AZ, NM,	ЖN	193	NA	17.84(b)	rish
Snait, Chittenango ovale amber	Triodopsis platysayoides] U.S.A. (WV)		_			,	=
Snail, flat-spired three-toothed Snail, lowa Pleistocene Snail, Manus Island tree	Triodopsis platysayoides] U.S.A. (WV)			1		i	۵
Snail, Iowa Pisiatocene Snail, Manus Island tree	Discus macclintocki Papustyla pulcherrima			T	41	NA	NA.	ă
inali, Manua laland tree	Papustyla policherrima	I II A A (A)	NA	Ţ	41	NA	NA.	
			NA	E	41	NA.	NA	MIGHT
inali noonday (Pacific Ocuan: Admirally is. (Manus is)	NA	E	3	NA	NA	=
	Mesodon ciarki nantahala	U.S.A. (NC)	NA	T	41	NA	NA	=
	Achetinelle spp. (all specius)	USA (10)	NA	E	108, 112	NA	NA	=
	Anguispire picte	U.S.A. (TN)	NA	٢	41	NA	NA	-
	Orthalicus reses (not incl. nesociyas)	USA (FL)	NA	T	41	NA.	NA.	Ų
Snall, Virginia tringed mountain	Polygyriscus virginianus	U.S.A. (VA)	NA	E	40	NA	NA	3
CLAME	•						ļ	•
early mussel, Alabama lamp	Lampsils viescens	USA (AL, TN)	NA	E	15	NA.	NA.	5
early mussel, Appalachian monkeylace	Quadrula sparsa	U.S.A. (TN, VA)	NA	E	15	NA.	NA.	=
early museel, birthing	Correctile casieta		NA	Ē	15	NA	NA.	Interior
early mussel, Comberland bean	Vilosa (= Hicromya) trabalis	U.S.A. (KY, TN)	NA	E	15	NA.	NA	ā
early mussel, Cumberland monkeylace	Quadrula intermedia	U.S.A. (AL, TN, VA)		Ë	15	NA.	NA.	-
	Eploblasma (=Dyanomia) florentna curtisi.	U.S.A. (MÓ)	NA	Ē	15	NA	NA	
early mussel, dromedary	Oromus dromas	U.S.A. (TN, VA)	NA	€	15	NA	NA	
early mussel, green-blossom	Epioblasma (= Dysnomia) torulosa pu-	60	NA	Ē	15	NA.	NA.	
	bomaculum.		0.04. (•	"		וות	
early mussel, Higgins' eye	Lampsilis higginal	U.S.A. (IL, IA, MN, MO, NE, WI)	NA I	E	15	NA	NA	
early mussel, Nicklin's	Megalonales nicklineana	Maxico		Ē	15	NA I	NA.	
	Plethobasus cooperlanus	U.S.A. (AL, IN, IA, KY, OH, PA, TN)	NA	Ē	15	NA.	NA NA	
	Toxolasma (Carunculina) cylindrelius	U.S A. (AL, TN)	NA	Ē	15	NA I	NA AM	
	Lampsilia orbiculata	U.S.A. (AL, IL, IN, KY, MO, OH, PA, TN,	MA	Ē	15	NA I		
and thereof has trendent and	parigrams between an institution of the parish of the pari	W.	110	-	'3	n^]	NA	
sarly mussel, Tampico	Cyrlonales templopensis tecomatensis	Mexico	NA	Ε!	15	NA.	NA	
	Epichiasma (=Dyanomia) torulosa toru- insa.	U S.A. (IL, IN, KY, TN, WV)	NA	Ē	15	NA	HA	
early mussel, turgld-blossom	Epioblasma (- Dysnomia) turgulula	U.S.A. (AL, TN)	NA	E	15	NA	NA	
early mussel, white cal's paw	Epioblesma (=Dysnomia) suicata deli- cata	U.S.A. (IN, MI, OH)	NA	Ē	15	NA.	NA NA	
early mussel, white wartyback	Piethobasus cicatricosus	U.S.A. (AL, IN, TN)	MA	اے	15	NA		
	Epioblasma (-Dysnomia) florentina flor-	U.S.A. (AL, TN)		Ē۱	15	NA NA	AA AA	
	entiria.		i	- [13		NA	
igtoe, fine-rayed	Fusconaus cureokis	U.S.A. (AL, TN, VA)	NA	Εl	15]	NA	NA.	
rgioe, rough	Pleuroberna plenum	U.S.A. (IN, KY, TN, VA)	NA	ĒΙ	15	NA.	NA	
	Fusconsia edgariana	U.S.A. (AL, TN, VA)		E	15	NA	NA.	400
		U.S.A. (AR, IN, MO, OH)		Ēί	15	NA	NA	
		U.S.A. (KY, TN, VA)		ĒΙ	27	NA I	NA.	N

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Соттоп лате	Scientific name	Historic range	population where endangered or threatened	tue tue	Hated	habitat	rules
Spkry mussel, Tar River	Eliptio (Canthyria) ateinstansena	U.S.A. (NC)	NA	E	188	NA	NA
CRUSTACEANS							
Amphipod, Hay's Spring	Stypobromus hayl	U.S.A. (DC)	NA	E	115	NA	NA.
Cravish, Nashville	Orconectes shoupl	U.S.A. (TN)	NA	E	242	NA	NA.
leopod, Madleon Caye	Antrolana Ira	U.S A. (VA)	NA	l II	123	NA.	17.46(a)
Isopod, Socorro	Thermosphaeroma (=Exosphaeroma)	U.S A. (NM)	NA	E	36	NA	NA.
	thermophilus.	1164 (8)5	NA.	Ε	135	17.95(h)	NA.
Shrimp, Kentucky cave	Palaemonias garileit	U.S.A. (KY)	[Whaterman community	\ •	133	i trafini	111/1
[HBECTB							
Beetle, delta green ground	Flenhaus viridis	U.S.A. (CA)	NA	T	100	17.05(i)	NA.
Seetle, valley elderberry longhorn	Desmocerus californicus dimorphus	do	NA	1	89	17.95(i)	NA
Butterfly, El Segundo blue	Euchilotes (=Shkimlaeoldes) battoldes	do.,	NA	[E	14	NA.	NA.
Containy, Ci Cogarias ansamment	allyni,						!
Butterfly, Lange's metalmark	Apodemia mormo langei	do	HA	E	14	NA.	NA.
Butterfly, fotta blue	Lycaeides argyrognomon lotes	do	NA	E	14	NA.	NA.
Bulletty, mission blue	Icancia icanoides missionensis	00	M	E	14	NA.	NA.
Butterfly: Oregon aliverspot	Speveria zerene hippolyta	! USA (OA, WA)	NA	T	95	17.95(1)	NA.
Butterfly: Palos Verdes blue	Glaucopsyche fygdamus palosverdesen-	U.S.A. (CA)	NA	E	98	17.95(1)	NA.
•	s/s.			_	ادما	١	١
Butterfly, San Brono elfin		do	NA	\ <u> </u>	14	NA NA	NA NA
Butterlly, Schaus swallowtail	Heraclides (=Papilio) aristodemus pon-	U.S A. (FL)	NA	-	13,159	NA.	l uv
	CORNUS.	HEA ION	NA.	اء	14	NA.	NA.
Bulterfly, Smith's blue	Euphilotes (=Shijimlaeoldes) enoptes smith!	U.S.A. (CA)	***************************************	ן בי	• • •	' * ^	''''
tests Many subsequently	Euproserpinus euterpe	h do	NA) т	91	NA.	NA.
Naucorid. Ash Moadowa	Ambrysus amergosus	USAINV	I NA	i i	181	17 95(0	NA.
DECOMO, VALENDECOME	Mines 1200 Street Areas - Transmission	a.a.r. fizzl monthemaniaminaminaminaminaminaminaminaminamin		<u> </u>			

Vertebrate

Editorial Note: For "When field" citations, see list following; for symbols in "When listed" see below:
#—Indicates FR where species was delisted; relisting of the species is indicated by subsequent number(s).
E—Indicates Emergency rule publication (see FR document for effective dates); subsequent number(s) indicate FR final rule, it applicable under "When listed".

Species

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50 CFR Ch. I (10-1-86 Edition)

2.55 FR 16047; O 2.55 FR 16047; O 2.55 FR 16047; O 2.55 FR 16047; O 2.57 FR 8495; Ju 2.57 FR 8495; Ju 2.57 FR 14678; Ju 2.57 FR 14678; Ju 2.57 FR 14678; Ju 2.57 FR 14678; Ju 2.57 FR 1604; Ju 2.57 FR 1604; Ju 2.57 FR 2968; Ju 2.57 FR

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ion)

_32 FR 4001: March 11, 1967. -35 FR 16047; October 13, 1970. -35 FR 8495; June 2, 1970. 35 FR 18320; December 2, 1970. 37 FR 6476; March 30, 1972. -38 FR 14678; June 4, 1973. 7-39 FR 44991; December 30, 1974. 40 FR 29864; July 16, 1975. 40 FR 31738; July 28, 1975 40 FR 44151; September 25, 1975. 40 FR 44418; September 26, 1975. -40 FR 47506; October 9, 1975. 41 FR 17740; April 28, 1978. 41 FR 22044; June 1, 1976. -41 FR 24064; June 14, 1976. 41 FR 45993; October 19, 1976. 41 FR 51021; November 19, 1976. 41 FR 51612; November 23, 1976. 41 FR 53034; December 3, 1976. 42 FR 2076; January 10, 1977. 42 FR 2968; January 14, 1977. 42 FR 15971, March 24, 1977. 42 FR 28137: June 2, 1977. 42 FR 28545; June 3, 1977. 42 FR 37373; July 21, 1977. 42 FR 40685; August 11, 1977. 42 FR 42353; August 23, 1977. 42 FR 45528; September 9, 1977 42 FR 58755; November 11, 1977 42 FR 80745; November 29, 1977. 43 FR 3715; January 27, 1978. 43 FR 4028; January 31, 1978, 43 FR 4621; February 3, 1978. -43 FR 6233; February 14, 1978. 43 FR 9612; March 9, 1978. -43 FR 12691; March 27, 1978. -43 FR 15429; April 13, 1978, 43 FR 16345; April 18, 1978. 43 FR 20504; May 12, 1978. 43 FR 28932; July 3, 1978. 43 FR 28988; July 28, 1978. -43 FR 34479; August 4, 1978. -44 FR 21289; April 10, 1979. 44 FR. 23064; April 17, 1979. 44 FR 29480; May 21, 1979. 44 FR 37126; June 25, 1979. 44 FR 37132; June 25, 1979. 44 FR 42911; July 20, 1979. -44 FR 49220; August 21, 1979. -44 FR 54007; September 17, 1979. 44 FR 59084; October 12, 1979. 44 FR 69208; November 30, 1979, 44 FR 70677; December 7, 1979. -44 FR 75076; December 18, 1979. 45 FR 18010: March 20, 1980. 45 FR 21833; April 2, 1980. 45 FR 24090; April 8, 1980. -45 FR 27713; April 23, 1980. -45 FR 28722: April 30, 1980. -45 FR 35821: May 28, 1980. 45 FR 44935; July 2, 1980. -45 FR 44939; July 2, 1980. -45 FR 47352; July 14, 1980. -45 FR 47355; July 14, 1980. 45 FR 52803; August 8, 1980. -45 FR 52807; August 8, 1980, 102-45 FR 54678; August 15, 1980.

103-45 FR 55654; August 20, 1980. -45 FR 63812; September 25, 1980. -45 FR 65132; October 1, 1980, 108--46 FR 3178: January 13, 1981. -46 FR 11665; February 10, 1981. -46 FR 40025; August 6, 1981. 112--46 FR 40664; August 10, 1981. -47 FR 4204; January 28, 1982, 114 47 FR 5425; February 5, 1982. 47 FR 19995; May 10, 1982. 47 FR 31670: July 21, 1982. 47 FR 43701; October 4, 1982. 47 FR 43962; October 5, 1982. 47 FR 46093; October 15, 1982, 125 48 FR 612; January 5, 1983. 48 FR 1726; January 14, 1983, 48 FR 28464; June 22, 1983. 130-48 FR 40184; September 2, 1983. 131--48 FR 43043; September 21, 1983, -48 FR 46057; October 11, 1983. 132-134 -48 FR 46336; October 12, 1983. 135 -48 FR 46341; October 12, 1983. 136 -48 FR 49249; October 25, 1983. -49 FR 1058; January 9, 1984. 137-138 -49 FR 1994; January 17, 1984. 130 -49 FR 2783: January 23, 1984. 142 -49 FR 7335; February 28, 1984. 143 49 FR 7394. February 29, 1984. 49 FR 7398; February 29, 1984. 49 FR 10526; March 20, 1984. 49 FR 14356; April 11, 1984. 49 FR 22334; May 29, 1984. 49 FR 27514; July 5, 1984. 49 FR 33885; August 27, 1984. 49 FR 34494; August 31, 1984. 49 FR 34504; August 31, 1984. 49 FR 34510; August 31, 1984. -49 FR 35954: September 13, 1984. -49 FR 43069: October 26, 1984. -49 FR 43969: November 1, 1984. 163 164 -49 FR 45163; November 15, 1984. -49 FR 49639; December 21, 1984. 168 -50 FR 1056; January 9, 1985. 169 -50 FR 4226; January 30, 1985. 170--50 FR 4945: February 4, 1985. 171--50 FR 12302; March 28, 1985. -50 FR 12305; March 28, 1985. -50 FR 20785; May 20, 1985. 181--50 FR 21792; May 28, 1985. -50 FR 23884; June 6, 1985. -50 FR 24530; June 11, 1985. -50 FR 24653; June 12, 1985. -50 FR 25678; June 20, 1985. -50 FR 26575; June 27, 1985. -50 FR 27002; July 1, 1985. 193--50 FR 30194; July 24, 1985. 195 -50 FR 31596; August 5, 1985. 196-–50 FR 31603; August 5, 198**5.** –50 FR 37198; September 12, 198**5.** 203--50 FR 39117; September 27, 1985. -50 FR 39123; September 27, 1985. -50 FR 50308; December 10, 1985. 210--50 FR 50733; December 11, 1985. -50 FR 51252; December 16, 1985. 216-51 FR 6690; February 25, 1986. 222-51 FR 10850; March 31, 1986.

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[48 FR 34182, July 27, 1983; 48 FR 34961. Aug. 2, 1983, as amended at 48 FR 39943, Sept. 2, 1983; 48 FR 46337, Oct. 12, 1983; 48 FR 52743, Nov. 22, 1983; 49 FR 1058, Jan. 9, 1984: 49 FR 33892, Aug. 27, 19841

-51 FR 34425; September 26, 1986.

EDITORIAL NOTE For additional FEDERAL RECISIER citations affecting the table in § 17.11(h), see the listing which follows the table.

EFFECTIVE DATE NOTE At 51 FR 34412, 34425, Sept. 26, 1986, the table in § 17/11(h) was amended by adding "Shrew, Dismal Swamp southeastern" alphabetically under 'Mammals" and adding "Crayfish, Nashville" alphabetically under "Crustaceans", effective October 27, 1986.

§ 17.12 Endangered and threatened plants.

(a) The list in this section contains the names of all species of plants which have been determined by the Services to be Endangered or Threatened. It also contains the names of species of plants treated as Endangered or Threatened because they are sufficiently similar in appearance to Endangered or Threatened species (see § 17.50 et seq.).

(b) The columns entitled "Scientific name" and "Common name" define the species of plant within the meaning of the Act. Although common names are included, they cannot be relied upon for identification of any specimen, since they may vary greatly in local usage. The Services shall use the most recently accepted scientific name. In cases in which confusion might arise, a synonym(s) will be provided in parentheses. The Services shall rely to the extent practicable on the International Code of Botanical Nomenclature.

(c) In the "Status" column the following symbols are used: "E" for Endangered, "T" for Threatened, and "E [or T] (S/A)" for similarity of appearance species.

(d) The other data in the list are nonregulatory in nature and are provided for the information of the

reader. In the annual revision and compilation of this title, the following information may be amended without public notice: the spelling of species' names, historical range, footnotes, references to certain other applicable portions of this title, synonyms, and more current names. In any of these revised entries, neither the species, as defined in paragraph (b) of this section, nor its status may be changed without following the procedures of Part 424 of this title.

(e) The "Historic range" indicates the known general distribution of the species or subspecies as reported in the scientific literature. current present distribution may be greatly reduced from this historic range. This column does not imply any limitation on the application of the prohibitions in the Act or implementing rules. Such prohibitions apply to all individuals of the plant species, wherever found.

(f)(1) A footnote to the FEDERAL REG-ISTER publication(s) listing or reclassifying a species is indicated under the column "When listed." Footnote numbers to §§ 17.11 and 17.12 are in the same numerical sequence, since plants and animals may be listed in the same FEDERAL REGISTER document. That document, at least since 1973, includes a statement indicating the basis for the listing, as well as the effective

date(s) of said listing.

(2) The "Special rules" and "Critical habitat" columns provide a cross reference to other sections in Parts 17, 222. 226, or 227. The "Special rules" column will also be used to cite the special rules which describe experimental populations and determine if they are essential or nonessential. Separate listings will be made for experimental populations, and the status column will include the following symbols: "XE" for an essential experimental population and "XN" for a nonessential experimental population. The term "NA" (not applicable) appearing in either of these two columns indicates that there are no special rules and/or critical habitat for that particular species. However, all other appropriate rules in Parts 17, 217 through 227, and 402 still apply to that species. In addition, there may be other rules in this title that relate to

U.S. Fish

such pla ments. I erences list all th ices whi or to the

Boraginaceae-Borage family: Ameinckie grandiflore Large-flowered fiddleneck, 179 17 96(a) Brassicaceae-Mustard family; Arabis medonaldiana.. McDonald's rock-cress. U.S.A. (CA). E Erysimum capitatum var. angustatum...... Contra Costa waliflower E 39 17 96(a) NA Thelypodium stenopetalum. Signder-petaled muntard... 158 NA NA Buxaceae--- Boxwood family: Buxus vahla ... Vahi'a boxwood. 197 Cartarean Carbia Inmilio

Boraginacese—Borage family: Artisinchia grandifiona	Large-flowered fiddleneck	U.S.A. (CA)	E	179	17.96(a)	NA
kassiceceae Mustard (smily:	to the state and the same	110.00	E	44	NA	NA.
	McDonald's rock-cress		Ē	30		AM
Pri Seritori Debiterra der fra Brancon international	Contra Costa wallflower		=		17.9d(a)	•
Thelypodum steriopatalum	Stender-petaled mustaid		=	158	NA	NA
uxaceaeBoxwood family:	Man har and		E	407	ll	
Buxus yahii	Vahi's boxwood	U.S.A. (PH)	=	197	NA	NA
actacese—Cactus family:			ا۔			•••
VIVERACEOING CONTROL (CONTROL -)	Tobusch fishhook cactus	U.S A. (TX)	E	80	l NA I	NA
	Fragrant prickly-apple	USA (FL)	E	208	j NA j	NA
	Key tree-cacks	USA. (FL), Cuba	E	153	NA į	NA
	Nelke cory caclus	U.S.A. (TX)	E	91	NA	NA
Copphanthe familiose	Bunched cory cactures	U.S.A. (TX), Mexico (Coahulle).	1	77	NA	NA
Coryphantha robbinsorum (~ Cochisela I., Escobaria I.)	Cochise pincushion cactus	U.S.A. (AZ), Maxico (Sonora).	. 1	214	NA I	HA
Coryphantha sneecki var. leei (=Escobena I., Mammitaria I.)	Les pincushion cacing	U.S.A. (NM)	T	61	NA I	NA.
	Sound pincushion cactus	U.S.A. (TX, NM)	E	82	NA	NA.
Delibration and the minutes (- conserved at minutes at		USA. (AZ)	£	71	NA .	NA
Sold Addition (In-this Property of the Control of t		U.S.A. (U'I)	Ē	58	l NA	NA.
Echnocereus fendied var. kuenzied (= E. kuenzied, E. hempeli el authors, not Fobel.		U S.A. (NM)	٤	70	NA.	NA
	Lloyd's hedgehog cactus	U S.A. (TX)	E	67	l NA l	NA
	Black lace caclus	do	E	68	I NA I	NA.
		U.S.A. (AZ)	Ë	62	NA I	NA.
Echinocereus Inglochidatus val. inermis (= E. coccineus val. I, E. phoeni- ceus val. I).		U.S.A. (CO, UT)	E	83	NA	NA
	Davis' green plays	U.S.A. (TX)	٤	81	NA	NA
		U.S.A. (TX), Mexico (Coehule).	T	77	NA	NA
Pedocactus brady (=Tourneys b)	Brady pincushion cactus	U.S A. (AZ)	ε	63	NA	NA.
	Knowlion cactus	U.S.A. (NM. CO)	ĒΙ	72	NA	NA
Pediocactus peeblesianus var. peeblesianus (= Echnocactus p., Navajoa p., Tourneya p., Utahia p.).		U.S.A. (AZ)	E	69	NA	NA
	Siler pincustion cactus	U.S.A. (AZ, UT)	E	64	امعا	NA
Sciencectus glaucus (= Echinocectus g., E. subglaucus, E. wnippiel var. g., Prediocectus g., S. kantiinii, S. whippiel var. g.).			Ť	59	NA	NA
Sciencectus muses-verdee (=Golorauxa m., Echinocectus m., Padaxiactus m.)	Mess Verde cactus	U.S.A. (CO, NM),	Ţ	75	NA	NA
Scierocactus wrighties (=Pediocactus w.)	Wright fishhook caclus	U.S.A. (UT)	Ε	58	NA	NA
Caryophyllaceae—Pink family:	Diamond Head schledea	1	E			

Manger Sale ich bours for

	NUIDPONE (NONEYERISE	PURILIZABLE TREBUNOST.						
•	Cistaceas—Rockrose lamily:	Mountain golden heather	USA (NO	G	107	17.96(a)	NA	
	Crassulacese—Stonecrop family: DexStyre traskles	Sania Barbara Island Iveloraver	usa (ÇA)	£	39	NA	NA	
	Cucurbiaceae—Gourd family: Turnamoca macdougali	Tumamoc globe-berry	U.S.A. (AZ), Mexico (Sonore).	E	228	NA	NA	
	Cupressaceas—Cypress family: Fittnoya cupressories	Chilean (ales terch (-elerce)	Chile, Argentine	т	70	NA	NA	
	Oyperaceae—Sodge family: Carex specurors	None commence commencement	US A. (AZ)	т	176	17.96(a)	AA	
102	Ericaceae—Heath lamity: Anctostophylos pungens var. navenil (= A. hooked sag. ravenil)	Preskito (-Reven's) mantanila		E	95 47	NA NA	AA AA	
	Euphorbia akottsbergii yar, kalasioana	Spurge	USA (HI)	E E E	192 192 120 154	NA NA NA	AA AA AA AA	50
	Astragalus perianus Astragalus prioenia Baptisla aractinilota Galactia amalii Holfmannsegga fanella Lotus dandroldeus sap. Eraklia (– L. acoparius sap. L)	Crenulate lead-plant Mancos milk-vetch Mancos milk-vetch Ash Mesdows milk-vetch tisky rattleweed Small's milkpes Stander rush-pos San Ctemente taland broom Uhluhi Hawailan vetch	USA (CO, NM) USA (UT) USA (NV) USA (GA) USA (FL) USA (TX) USA (CA) USA (CA)		182 167 39 181 39 182 209 26 238	NA NA NA NA 17 98(4) NA NA NA NA	AN AN AN AN AN AN AN AN	CFR Ch. I (10-1-86
	Frankenia johnstonii	Johnston's frankenia	U.S.A. (TX), Mexico (Nuevo Leon).	ε	155	NA	NA	Edition)
						والمراجع المراجع		

Gentlanaceae—Gentlan family: Centeurum nemoylulum	Spring-loving centary	U.S.A. (CA, NY)	7	181	17 90(a)	HA	U.S.
Goodeniaceas—Goodenia tamby: Scaerola coriacea	Dwarf raupaka	U.8 A (H),	Œ	231	NA	АН	Fish
Hydrophyllacese—Waterlest family: Phecaba argiflaces Phacella formosula	Clay phacella North Park phacella	USA. (CO)	E	44 121	AM AM	AA AH	and Y

Frankeniaceas—Frankenia ternity: Frankenia Johnstonii	Johnston BANS	" UBÅ (TX), Mesico (Hueyu Leon).	1	,	
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and the second s					
Фельперсон Остал Інтігу; Септент петрілікт	Sping-loving centaury	U.S.A. (CA, NV)	T 181	17.96(s)	NA
Goodeniscese Goodenia family:		1	_		
Scaryole conscess	Dyell neupaks	U.S.A. (H)	E 231	NA	NA
Hydrophyllaceze Waterleaf family:					
Phaceia stollaces	Clay phacella	U.S.A. (UT)	E 44	NA [NA
Priscola formosula		U.S.A. (CO)	E 121	NA	NA
	<u> </u>		1		
Lamiaceae—Mint family:		1			
Acanthominthe obovate sap. duttonii		U.S.A. (CA)	E 204	NA	NA
Dicerandra cornutasima			E 207	NA	NA
Dicerantira Inviescena		do	E 207	NA	NA
Dicerandra immaculate	Lakula's mkn	60	E 180	NA]	NA
Haplostachys haplostachys var. angustilolia	None	U.S.A. (HI)	E 73	NA	NA
Hedeoma apiculatum			T 118	17.96(a)	NA
Hedeoma todseni	Todaen's pennyroyal	U.S.A. (NM)	E 110, 112	17.08(a)	NA
Россото выгаты	San Diego mesa mint	U.S.A. (CA)	E 44	NA L	NA
Sculellana montana	Large-flowered akulicap	U.S.A. (GA,TN)	E 234	NA I	NA
Stanogyne angustiloke var. angustiloke	None	U.S.A. (HI)	E 73	NA NA	NΑ
Lauraceae—Laurel family:	"	1			
Lindra motssiola	Pontberry	U.S.A. (AL, AR, FL, GA,	E 240	[NA]	NA
		LA, MO, MS, NC, SC).	Ì		
Likeceae—Lify family:			1 '	1	
Erythronium propuliana	Minnesota trout kly	U.S.A. (MN)	E 221	NA	NΑ
Harperocalis fleys	Harper's beauty	U.S.A. (FL)	E 67	NA.	AM
Tolium persistens	Perestent trillium	U.S.A. (GA, SC)	E 30	NA	NA
Lossaceae—Lossa family:		1	_		
Mantralia leucophylia	Ash Meadows blazing star	U.S.A. (NV)	T 181	17.96(a)	NA
MalvaceaeMallow family:		1	1		
Abution menziosii	Koʻologʻula	U.S.A. (HI)	E 243	NA NA	NA
Calienoe scabriuscula	Tuxas poppy-matow		E 109, 112	NA .	NA
Hibiscadulphus distane	Kausi hau kushiwi	U.S.A. (HI)	E 225	NA	NA
liames corel	Peter's Mountain mallow		E 230	NA	NA
Kokie cookel	Cooke's koklo		E 74	NA	NA
Kokia drynanoides	Koliro (=hsu-hele'ula or Hawaii tree	do	E 167	17.95(a)	NA
Malacothamous clementinus		U.S.A. (CA)	E 28	المما	NA
Skielos politis		40	E 158	NA NA	NA
NyctaginaceaeFour-o'clock family:]		ł I	
Mychiganeolae-rou-o cock turny:	MacFarlane's lour-o'clock	1,104,100,000	E 66	المدا	NA

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	Cemissonia pentensis Oenothera avita tap. eurokensis	San Banko evahing-primosa		E	1,171	(41)			
	Oenothers detoides sep. howelds	Antioch Dunes evening primose		E	39	17.98(a)	NA		เลี้ยา เรื่องได้ได้เรียกเรื่องสามารถใน
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	Orchidecess—Orchid lemity:	.	1,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	ΕÌ	122	NA	NA		
	lacina medeoloides	Small whorled pogonia	ME. MI. MO. NG. NH.	1	1==	••••	14.		原建制 计图像设备 中外 注:
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	Spranibes parksii	Navasota ladies'-besses		E	116	NA [ŅĀ		
	Spranihes parksi		{ · · · · · · · · · · · · · · · · · · ·		1	1	•		的现在分词形式的 对话:
	Papavaraceae—Poppy family:	· l	ł	_ 1		1	414		计计划数据 建铁铁
	Accomecon hunita	Dwart bear-poppy	.} U.S.A. (UT)	E	76	NA	NA		
	A MATO TO A STATE OF THE PARTY	•	1 1			}			
	PinaceaePine temily:	Guatematan fir (pinabeta)	Mexico, Guatemata,	7	84	NA	NA		100000000000000000000000000000000000000
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104	Poaceas-Grass family:	1	1 1	_ :		1 1			经验证的证据的
142-	Trefreis microsals (- Occitis m.)	Solano grass	. U.S.A. (CA)	€.	44	17.96(a)	NA NA		
	Paricipo Cassos	{	J U.S.A. (HI)	£.	133	17.96(8) 1 NA	NA NA		
	Swakeria alexandras	/*** FINSKE DING OLESS	.] U.S.A. (CA)	E	39	17.96(4)	NA		
	Zirania tarana	Texas wid-fice	(XI)		38	17.00(6)	1367		
		ļ	1		}	, ,			
	Polygetecese—Mikwort family:	They polygele	U.S.A. (FL)	E	192	. NA	NA	0	
	Polygala smalli	tata baildaga		-	1	1 1		Ü	BOOK BOOK BUILDING
	Polyponacese—Buckwheat family:	1	1 1		.	1		χij	
	Engonen gypsophken	Gypsum wild-buckwhest	U.S.A. (NM)	Ţ	110, 112		NA NA	^	
	Education availablem var. williamsiaa	""] 2/98/JJD08/ DUCKAMIRINI		Ě	237 161	NA 17.96(4)	NA NA	ð	经验证据 建氯磺基磺基
	Eriogonum pelinophikum	Clay-toying wild-buckwheat		E	101	11/30(4)	UIA	-	
	-	}	}		1	1		$\overline{}$	医医生态 (1) (1)
	Primulaceas—Primrosa family:	Maguire primrose		7	199] NA	NA.	ᅙ	医外部性 2011年4月11日
	Princia magnital	wedne has on morning			1	ļ j		Y	
	Ranunculaceas—Buttercup tarrity:		1			ا ا	هده	ī	2012年 1912年 1912年 191
	Aconium noveboracense	Northern wild monkshood	[U.S.A. (IA, NY, OH, WI)	Ţ	39	NA NA	NA NA	8	的是其他的形态器 到一
	Clamatic acciatio	"" VINDEUS ICSNIES ICARA """"	[U.S.A. (AL)	E		I MA	NA NA		"是我们的是我们的
	Delphinum kinklense	San Clemente Island larkspur	. (U.S.A. (GA)	=	20	[~~ [7167	fft.	
		1	1		i	1 1		₹	
	Rhamnaceae—Buckthorn tamily:	None	11 S A (H)	E	165	17.95(a)	HA	dition	
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	Cowarie subinieges	Arizona cittirgae	U.S.A. (AZ)	E	148	NA	NA	ļn	国际经验管理机械
	lyests eremice	Ash Meadows Wesla	U.B.A. (NV)			17 90(E)	NA.		
	Polentille robbinslane		U.S A. (NH, VT)	[€	104	17.95(a)	NA.	31	
	A. 1. A. 1.	k	1	ì	1	1	ł	₽ •	
	Bublareas-Cottee lamity: Gardenia brighamii	Na'u (Hawalian gardania)	11 8 A DID	E	198	NA.	NA.	g	1. 10 · 10 · 10 · 10 · 10 · 10 · 10 · 10
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REFERENCE 15

Hanford Environmental Health Foundation

Drinking Water Report, HEHF-45

HANFORD SANITARY WATER SYSTEMS

During CY-1984, sanitary water was supplied on the Hanford Site by 17 individual drinking water systems, each consisting of a raw water supply, treatment facilities, and distribution piping. All of the systems are operated by Department of Energy contractors with the exception of the City of Richland municipal system which provides water to the 700, 1100, and 3000 Areas. Twelve of the systems utilize Columbia River water as a raw water source, four systems utilize groundwater, and one system (Richland municipal) a combination of the two. The systems range in size from those providing plant-scale treatment and serving extensive areas to nine systems which supply water to single or small complexes of facilities only.

The systems, along with their sources of supply, service areas, and additional pertinent information, are listed below. A Hanford Site map depicting locations of the systems (with the exception of the Richland municipal system) is given in Figure 1.

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System	Source of Supply	Notes
100-8	Columbia River via 181-B Pumphouse	Filtered and chlorinated. Sani- tary water supplied to 182-8 81dg. only. System is Rockwell (RHO) operated.
100 - D	Columbia River via 181-D Pumphouse	Filtered and chlorinated at 183-D. Treated water also supplied to 100-F and 100-H Areas as needed. System is RHO operated.
100 - K	Columbia River via 181-K Pumphouse	Filtered and chlorinated. System is United Nuclear Corporation (UNC) operated.
100-N	Columbia River via · 181-N Pumphouse	Filtered and chlorinated. System is UNC operated.
200 - E	Columbia River via 100-8 raw water export	Filtered and chlorinated at 283-E. System is RHO operated.

REFERENCE 16

<u>Draft Phase I Installation Assessment of Inactive</u>

<u>Waste-Disposal Sites at Hanford</u>, Volume 3

SITE ID NO.: Pickling Acid Crib

ALIAS: None

FACILITY: Crib

STATUS: Inactive DIMENSIONS:

ELEVATION: WATERTABLE:

Length: 50 ft Width: 30 ft Depth: 10 ft Diameter: 0

LOCATION: 600 Area

COORDINATES: To be determined

SITE DESCRIPTION:

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A crib type structure 50 feet by 30 feet and 10 feet deep located 500 yards from the old White Bluffs Ice Plant and 300 yards east of Federal Avenue. Vent pipes extrude every 18 inches at the surface.

SERVICE DATES: 1943 - 1945

SERVICE HISTORY:

The site was used to pickle galvanized piping for use in the reactor buildings during construction. The process used several thousand gallons of nitric and hydrofluoric acid. The site was retired in 1945 and the surface was covered with large cobbles.

REFERENCES:

Documents:

Photographs: 122440-464-CN, 122440-465-CN

Drawings:

SITE ID NO.: Pickling Acid Crib

CHEMICALS DISPOSED

No chemical inventory is available.

RADIONUCLIDE INVENTORY (in curies)

H - 3:	0.00000	CE-144:	0.00000
C-14:	0.00000	PR-144:	0.00000
MN-54:	0.00000	PM-147:	0.00000
CO-60:	0.00000	EU-152:	0.00000
NI-63:	0.00000	EU-154:	0.00000
KR-85:	0.00000	EU-155:	0.00000
SR-90:	0.00000	NP-237:	0.00000
Y-91:	0.00000	PU-238:	0.00000
NB-95:	0.00000	PU-239:	0.00000
ZR-95:	0.00000	PU-240:	0.00000
TC-99:	0.00000	PU-241:	0.00000
RU-103:	0.00000	AM-241:	0.00000
RU-106:	0.00000	U-233:	0.00000
SN-113:	0.00000	U-235:	0.00000
S8-125:	0.00000	U-238:	0.00000
I-129:	0.00000	TH-232:	0.00000
CS-134:	0.00000	BETA:	0.00000
CS-137:	0.00000	GAMMA:	0.00000
CE-141:	0.00000	ALPHA:	0.00000

This site was not used to deliberately dispose of radioactive waste.

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These values are decayed through April 1, 1986.

SITE ID NO.: 107-B

DIMENSIONS:

ELEVATION: 440 feet

WATERTABLE: 40 feet

Length: 450 feet Width: 230 feet Depth: 24 feet

Diameter:

FACILITY: Retention Basin

LOCATION: 100-B/C

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COORDINATES: N71660/W80560, N71660/W80090, N71430/W80560, N71430/W80090

DESCRIPTION OF FACILITY

Concrete lined basin, 230 ft. x 467 ft., with a vertical baffle down the middle (lengthwise). The floor of the basin consists of concrete slabs, their joints originally closed with neoprene water seals. To a height of almost 10 ft. above the floor, the walls slope and are about 4 inches thick. The upper sections of the walls (about 10 ft) area vertical and range in thickness from about 5'8" at the bottom to one foot at the top.

The basin has been backfilled with soil to a depth of almost four feet.

DATE OF OPERATIONS: 1944-1968

DESCRIPTION OF WASTE

Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. Total radionuclide inventories in the vicinity of the basins ranged from 5 to over 400 curies. 80% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The following is a summary of unplanned releases in the basin area:

- In early 1952, gross leakage at the inlet for the 105-B effluent
- line was detected and steadily increased in volume.

 In late 1952, two known leaks from the effluent line occurred: (1) near the #2 diversion box for the 30" line and (2) near the 8" riser for the temporary by-pass line northeast of the 105-8 Building.
- In February, 1954, a break occurred in the 107-B Basin.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

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REFERENCES:
Documents: UNI-946

Photographs:

Drawings:

SITE ID NO.: 107-C

DIMENSIONS:

ELEVATION: 440 feet

WATERTABLE: 40 feet

Length: Width:

Depth: 16 feet Diameter: 330 feet

FACILITY: Retention Basins (2)

LOCATION: 100-B/C

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COORDINATES: N-71045/W-79970, N-71045/W-80320

DESCRIPTION OF FACILITY

Two carbon steel tanks, each 330 feet in diameter and 16 feet deep. Each has a series of steel baffle plates inside to prevent water from channeling across the tank into the discharge line. Both tanks have been backfilled with soil to a depth of about four feet.

DATE OF OPERATIONS: 1942-1969

DESCRIPTION OF WASTE

Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. Total radionuclide inventories in the vicinity of the basins ranged from 5 to over 400 curies. 80% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The basin and its approximate 5 foot diameter effluent line has developed leaks during its operating life. It has been said the leaks could have been as high as 5,000 - 10,000 gallons per minute.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: P-5242

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Ref. 16.6

SITE ID NO.: 107-D

DIMENSIONS:

ELEVATION: 420 feet

WATERTABLE: 35 feet

Length: 450 feet Width: 230 feet Depth: 24 feet Diameter:

FACILITY: Retention Basin

LOCATION: 100 D-DR

COORDINATES: N-94455/W-52730 to N-94685/W-53250

DESCRIPTION OF FACILITY

A concrete-lined basin, 230 ft. x 467 ft., with 20 foot walls. The design is about the same as 107-B. the basin has been backfilled to a depth of about 2 ft. The walls appear to be coated with asphalt.

DATE OF OPERATIONS: 1944-1967

DESCRIPTION OF WASTE

Retained reactor cooling water effluent water from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. Total radionuclide inventories in the vicinity of the basins ranged from 5 to over 400 curies. 70% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

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The following is a summary of unplanned releases in the basin area:

- In early 1950, leakage occurred on the north side between the basin and the river. Effluent water had drained under the road to the section between the road and the perimeter fence.
- In the fall of 1951, excess leakage was detected above ground. As a result, two excavations were made. The holes were then covered, but effluent water continued to seep to the surface.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

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REFERENCES:
Documents: UNI-946

Photographs:

Drawings: M-1901-D Sh.8

SITE ID NO.: 107-DR

DIMENSIONS:

ELEVATION: 420 feet

WATERTABLE: 35 feet

Length: 600 feet Width: 230 feet

Depth: Diameter:

FACILITY: Retention Basin

LOCATION: 100-D/DR

COORDINATES: N-94058/W52516, N94680/W52219

DESCRIPTION OF FACILITY

A concrete-lined basin, 230 ft. x 600 ft., with 20 foot walls. It has been backfilled to a depth of one to 3 feet.

DATE OF OPERATIONS: 1950-1964

DESCRIPTION OF WASTE

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Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. Total radionuclide inventories in the vicinity of the basins ranged from 5 to over 400 curies. 70% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The following is a summary of unplanned releases in the basin area:

- In late 1951, extensive leakage of effluent water at the inlet end of the 107-DR retention basin was caused by the pipes pulling loose from the basin wall.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: M-1901-D Sh. 8

Ref. 16.10

SITE ID NO.: 107-F

DIMENSIONS:

ELEVATION: 400 feet

WATERTABLE: 30 feet

Length: 450 feet Width: 230 feet Depth: 24 feet

Diameter:

FACILITY: Retention Basin

LOCATION: 100-F

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COORDINATES: N79967/W24130, N79967/W28900, N79500/W28900, N79500/W29130

DESCRIPTION OF FACILITY

A concrete-lined basin, 230 ft. x 467 ft., with 20 foot walls, similar in design to 100-B and 100-D. The basin has back-filled to a depth of about 5 feet, with soil piled to cover the walls.

DATE OF OPERATIONS: 1945-1965

DESCRIPTION OF WASTE

Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. 70% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: M-1600-F Sh.5

SITE ID NO.: 107-H

DIMENSIONS:

ELEVATION: 420 feet

WATERTABLE: 10 feet

Length: 600 feet Width: 273 feet Depth: 20 feet

Diameter:

FACILITY: Retention Basin

LOCATION: 100-H

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COORDINATES: N96000/W38740, N96000/W38466, N95368/W38466, N95368/W38740

DESCRIPTION OF FACILITY

Concrete-lined rectangular basin, 273 ft. x 600 ft.; 20 feet deep. The basin has been back-filled to a depth of about 4 feet above the floor and slopes to the top of the walls.

DATE OF OPERATIONS: 1949-1965

DESCRIPTION OF WASTE

letained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. 70% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The basin and its approximate 5 foot diameter effluent line has developed leaks during its operating life. It has been said the leaks could have been as high as 5,000 - 10,000 gallons per minute.

EXTENT OF CONTAMINATION

The extent of contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: M-1600-H Sh.4

SITE ID NO.: 107-KE

DIMENSIONS:

ELEVATION: 480 feet

WATERTABLE: 70 feet

Length: Width:

Depth: 29 feet Diameter: 250 feet

FACILITY: Retention Basins (3)

LOCATION: 100-K

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COORDINATES: N(K)565699/W426802, N(K)532850/W453651, N(K)53000/W480500

DESCRIPTION OF FACILITY

Three carbon steel tanks, 250 ft. in diameter and 29 ft. deep. They have been backfilled with soil to a depth of about 4 feet.

DATE OF OPERATIONS: 1955-1971

DESCRIPTION OF WASTE

Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. 80% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The basin and its approximate 5 foot diameter effluent line has developed leaks during its operating life. The leak rate from the butterfly valves (that went to an adjacent trench) could have been as high as 5,000-10,000 gallons per minute. Most of the basin leakage was diverted to an open canal and disposed to the river.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: H-1-25529

Ref. 16.13

SITE ID NO.: 107-KW

DIMENSIONS:

ELEVATION: 480 feet

WATERTABLE: 70 feet

Length: Width:

Depth: 29 feet Diameter: 250 feet

FACILITY: Retention Basins (3)

LOCATION: 100-K

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COORDINATES: N(K)5245/W(K)6190, N(K)5245/W(K)6460, N(K)5245/W(K)6730

DESCRIPTION OF FACILITY

Three carbon steel tanks, 250 ft. in diameter and 29 ft. deep. They have been backfilled with soil to a depth of about 4 feet.

DATE OF OPERATIONS: 1944-1970

DESCRIPTION OF WASTE

Retained cooling water effluent from the 105 reactor for radioactive decay and thermal cooling prior to release to the Columbia River. 80% of the total radionuclide inventory is contained within the soil adjacent to the basins. Approximately 10 curies have leached into the concrete floor and walls.

UNPLANNED RELEASE

The basin and its approximate 5 foot diameter effluent line has developed leaks during its operating life. The leak rate form the butterfly valves (that went to an adjacent trench) could have been as high as 5,000-10,000 gallons per minute. Most of the basin leakage was diverted to an open canal and disposed to the river.

EXTENT OF CONTAMINATION

The extent of the contamination from these releases is well within the zone encompassed by the retention basin and is within the AC-5-40 permanent posting.

*** MEMO FIELD EMPTY ***

REFERENCES:

Documents: UNI-946

Photographs:

Drawings: H-1-24429

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Ref. 16.15

Drawings: P-5242

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REFERENCE 17 File Memo regarding recreational use of the Columbia River

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Project Number	
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Internal Distribution

KH Cramer RD Stenner File/LB

Date

August 26, 1987

To

IWSS Files

From

DR Sherwood

Subject

Recreational Use of Hanford Reach

The Hanford Reach of the Columbia River has many recreational uses. Sportsman's access for fishing and waterfowl, big game, and upland bird hunting is provided at several locations. Access points to the Columbia River along the Hanford Reach are located at the old White Bluffs Ferry landing, the Ringold Hatchery, and Leslie Grove Park in Richland. I have fished for steelhead and salmon along the Hanford Reach since 1980.

DRS/mgs

REFERENCE 18

Letter from RD Stenner to DM Bennett regarding 100 Area Ground Water Contaminant Plume, October 14, 1987

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Pacific Northwest Laboratories P.O. Box 999 Richland, Washington U.S.A. 99352 Telephone (509)

Telex 15-2874

October 14, 1987

Mr. D. M. Bennett U.S. Environmental Protection Agency Region X Superfund Program 1200 6th Avenue Seattle, WA 98101

Dear Dave:

Enclosed are the three descriptions of the 100 Area, 200 Area and 300 Area ground water contaminant plumes we discussed on the telephone yesterday. I have included some attached figures and maps to help show the independency of the detected ground water contamination in each of the three aggregate areas.

If there are any questions regarding the descriptions, please contact me at 509-375-2916.

Sincerely,

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R.D. Stenner, Sr. Research Engr. Earth and Environmental Sciences Center GEOSCIENCES DEPARTMENT

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100 Area Ground Water Contamination

The ground water chromium and strontium-90 concentrations in the 100 Area are attributable only to the activities in the 100 Area because of the isolation of these areas and the fact that several years of ground water monitoring in the area show that any concentrations in ground water are found contained in close proximity to each of the individual reactor sites. Upstream samples of the surface water show that the contamination detected downstream are attributable to activity in the 100 Area. The furthest downstream location of contaminants from the 100 Area entering the river is at river mile 22. The furthest upstream entry point of the 100 Area contaminants is just above river mile 2, which is near the 100 B/C Area.

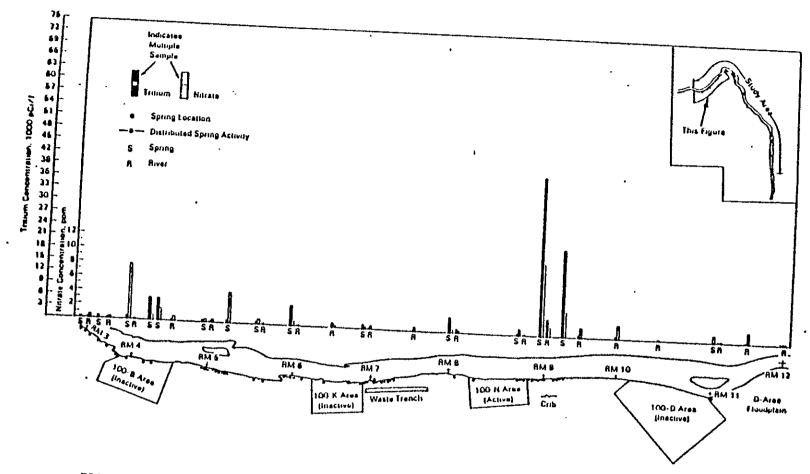
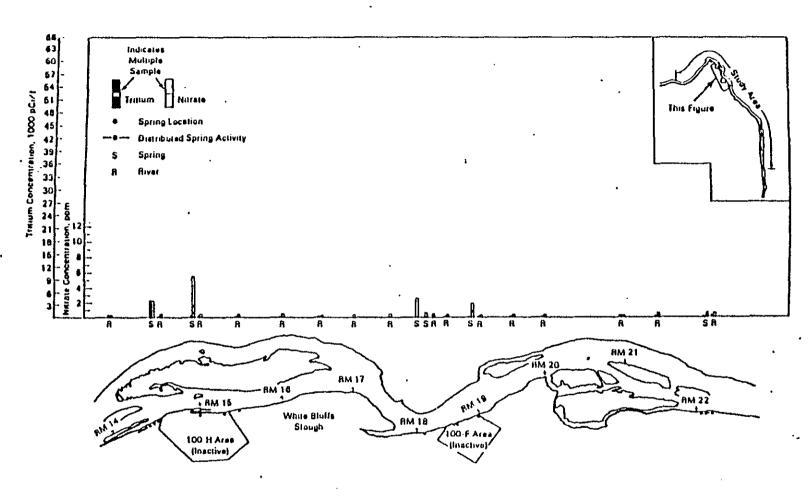


FIGURE 3. Locations and Analytical Results for Spring and River Samples from River Mile 3 through River Mile 12



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FIGURE 4. Locations and Analytical Results for Spring and River Samples from River Mile 14 through River Mile 22

REFERENCE 19

Letter from RD Stenner to DM Bennett regarding Liquid
Waste Sites and Burning Pits, October 26, 1987

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Pacific Northwest Laboratories P.O. Box 999 Richland, Washington U.S.A. 99352 Telephone (509)

Telex 15-2874

October 26, 1987

D.M. Bennett
EPA/NPL Coordinator
EPA Superfund Program
U.S. Environmental Protection Agency Region X
1200 Sixth Ave.
Seattle, Washington 98101

Dear Dave:

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Per your telephone request, enclosed are the statements on liquid waste sites and burning pits. It is our understanding that you will incorporate them as references in the appropriate sections of the NPL packages and provide us with a finalized copy of the packages following completion of the MITRE Corp. review.

If there are any questions regarding these statements, please don't hesitate to contact me at 509-375-2916.

Sincerely,

Bob

R.D. Stenner, Senior Research Engineer Environmental Pathways and Assessment Section Geosciences Department Earth and Environmental Sciences Center

RDS:dar

ENCLOSURE

bcc: KH Cramer

MS Hanson

DA Lamar

RM Mangin - DOE/RL

TJ McLaughlin

RG Schreckhise

WB Schulze - DOE/RL

DR Sherwood

100, 200, 300 Area Statement Regarding Liquid Wastes

The general operating procedures for liquid waste sites in the 100, 200 and 300 Areas were such that the waste constituents Tisted for each site generally entered the process lines and were mixed with each other prior to being disposed of at the site. This process mixing of these waste constituents occurred over the period of site operation.

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Burning Pit Statement

Due to the time period for which the burning pits operated, the nonhazardous combustible waste materials (i.e., paper products, cans, etc.) would have been mixed (i.e., mixed together in the garbage truck or waste container) with the hazardous waste materials (i.e., paints, solvents, etc.) prior to the waste mixture being disposed of at the site.

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Reference 20

Memo from Kathleen Galloway, MITRE, to Sandy Crystall, EPA December 29, 1987

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MITRE

To:

Sandy Crystall,

Date: 29 December 1987

Acting Chief for NPL Operations,

Environmental Protection Agency (EPA)

From:

Kathleen Galloway, Member of the Technical Staff,

The MITRE Corporation

Subject: The Toxicity of Uranium and Plutonium

Copies:

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B. Myers, S. Parrish

According to Sax, uranium is a highly toxic element on an acute basis as well as on a ratio-toxic basis. For this reason the Agency feels it is appropriate to assign a value of three for the toxicity of uranium.

Sax states that the toxicity of plutonium compounds is based first upon the very high radio-toxicity of the plutonium atom. In addition, the permissible levels for plutonium are the lowest for any radioactive element. Therefore, although a method for assigning toxicity values for radio-nuclides is not established, the language in Sax would appear to justify a toxicity value of 3 for plutonium.

KG/js'

The MITRE Corporation
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